



**HAL**  
open science

## Automatic ice-cream characterization by impedance measurements for optimal machine setting

Marco Grossi, Massimo Lanzoni, Roberto Lazzarini, Bruno Riccò

► **To cite this version:**

Marco Grossi, Massimo Lanzoni, Roberto Lazzarini, Bruno Riccò. Automatic ice-cream characterization by impedance measurements for optimal machine setting. *Measurement - Journal of the International Measurement Confederation (IMEKO)*, 2012, 45 (7), pp.1747-1754. 10.1016/j.measurement.2012.04.009 . hal-01276540

**HAL Id: hal-01276540**

**<https://hal.science/hal-01276540>**

Submitted on 25 Feb 2016

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

1 **Automatic Ice-Cream Characterization by Impedance Measurements for Optimal Machine**  
2 **Setting**

3

4 M. Grossi<sup>a,\*</sup>, M. Lanzoni<sup>a</sup>, R. Lazzarini<sup>b</sup>, B. Riccò<sup>a</sup>

5 \* Corresponding author: [marco.grossi8@unibo.it](mailto:marco.grossi8@unibo.it), Tel. 0039-0512093082, Fax 0039-0512093785

6 <sup>a</sup> Department of Electronic Engineering (D.E.I.S.), University of Bologna, Bologna, Italy

7 <sup>b</sup> Carpigiani Group, Anzola Emilia, Bologna, Italy

8

9 **Abstract**

10 Electrical characterization of products is gaining increasing interest in the food industry for quality  
11 monitoring and control. In particular, this is the case in the ice-cream industry, where machines  
12 dedicated to store ice-cream mixes are programmed “ad hoc” for different groups of products. To  
13 this purpose, the present work shows that essential product classification (discrimination between  
14 milk based and fruit based ice-cream mixes) can be done by means of a technique based on the  
15 measurements of non-linear response in the electrical behavior of the electrode-electrolyte interface.  
16 The addition of pH measurements allows to further reach the three parts classification occasionally  
17 required for advanced applications. The proposed idea is validated by means of measurements on  
18 21 ice-cream mixes, different for producers and composition.

19

20 *Keywords:* food quality control; ice-cream mix; industrial sensors; electrical impedance  
21 spectroscopy; electrode-electrolyte interface.

22

23 **1. Introduction**

24 While in the early stage of the food industry the competition was mainly focused on costs, today  
25 product quality and safety are a primary concern. Consequently, products are routinely screened for

26 important organoleptic characteristics (such as smell, aroma, color,...) as well as to guarantee that  
27 microbial content is below the maximum allowed threshold concentration.

28 In the past, all tests were performed off-line, i.e. a limited number of product samples were sent to a  
29 laboratory to be tested and the results became available with long delays. Today, automated  
30 production methods with integrated monitoring systems allow much faster response and the  
31 possibility to screen all the products with non destructive measurements.

32 In this context, food characterization by means of electrical measurements easily implementable in  
33 automatic form plays a crucial role and a number of significant examples can be mentioned:  
34 detection of water and lipid content in meat [1], dilution factor in apple puree [2], determination of  
35 pH, acidity and hardness in yogurt [3], quality control of vegetable oils [4]. In particular, as far as  
36 dairy products are considered, techniques have been proposed for the characterization of milk  
37 content [5][6][7], detection of mastitis in raw milk samples [8][9], measurement of microbial  
38 concentration in milk [10] and ice-cream [11][12][13].

39 A different, though related, application is automatic product recognition to optimize machine setting  
40 when dealing with different versions of the same basic product needing specific processing  
41 parameters. This is in particular the case for the ice-cream mixes, that are normally stored in  
42 dedicated machines maintaining the product at the target conservation temperature (normally in the  
43 range 2 – 6 °C), while pasteurization cycles are carried out at regular intervals (one or few days) to  
44 lower the microbial concentration below the legally allowed threshold. Different types of mixes,  
45 however, require different machine parameters setting and, to this purpose, in practice a distinction  
46 is made between milk and fruit based ice-creams, requiring different conservation temperatures and  
47 frequency of pasteurization cycles. Within the milk based products a further, less decisive  
48 discrimination is that between creamy and frozen yogurt products.

49 At present, specific machine parameters for these three ice-cream groups (but often differences are  
50 considered only among milk and fruit based mixes) are set manually: hence an operator is needed  
51 every time the stored ice-cream mix type is changed. Instead, if the product type could be

52 discriminated by means of an electronic system embedded in the machine, such an intervention  
53 would be no longer necessary, with significant advantages in terms of costs, time and error  
54 reduction.

55 Electrical Impedance Spectroscopy (EIS) is often used for electrical characterization of food  
56 products. In EIS the sample under test is placed in direct contact with electrodes and stimulated  
57 with a sinusoidal test voltage  $V_{in}(t) = V_M \sin(\omega t) = V_M \sin(2\pi f t)$  with fixed amplitude  $V_M$  in a  
58 definite range of frequencies. The current  $I_{in}(t)$  through the electrodes due to the test signal is  
59 measured and, if the system electrode-electrolyte can be considered linear (i.e. if  $V_{in}(t)$  is the  
60 weighted sum of several signals, then  $I_{in}(t)$  is the weighted sum of the system response to each of  
61 the signals), the complex impedance  $Z$  can be calculated as  $Z = |Z|e^{j\text{Arg}(Z)} = V_{in}(j\omega)/I_{in}(j\omega)$ , where  
62  $V_{in}(j\omega)$ ,  $I_{in}(j\omega)$  are the Steinmetz phasors of the sinusoidal signals  $V_{in}(t)$  and  $I_{in}(t)$  respectively,  $|Z|$  is  
63 the impedance modulus and  $\text{Arg}(Z)$  the impedance phase. The acquired spectra can be represented  
64 with different graph types, such as Bode plots, where  $|Z|$  and  $\text{Arg}(Z)$  are plotted as function of the  
65 test signal frequency, or Nyquist plots where the impedance imaginary component  
66  $\text{Im}(Z) = |Z|\sin(\text{Arg}(Z))$  is plotted vs. the real component  $\text{Re}(Z) = |Z|\cos(\text{Arg}(Z))$  for different  
67 frequencies.

68 EIS data for a set of samples, featuring different values of the parameter under study, are analyzed  
69 in order to extract a relation between the measured electrical parameters and the product parameter.

70 The electrochemical system composed of sample under test and electrodes is, however, a non-linear  
71 system [14]. To effectively apply EIS, the test signal must feature small amplitude  $V_M$  so to confine  
72 the system in a pseudo-linear region.

73 Nevertheless, study of the electrical response in the non-linear region (using larger amplitude  
74 excitation potentials) can extend the knowledge and provide additional data on the product.

75 In this paper ice-cream mixes, different for composition and producers, are tested both with EIS and  
76 in the non-linear region to achieve the products discrimination needed for practical purpose.

77

78 **2. Experimental approach**

79 The objective of the study is to discriminate a set of ice-cream mixes, different for composition and  
80 producers, in two different groups (milk based and fruit based mixes) with eventually a second level  
81 discrimination of the milk based products in creamy mixes and frozen yogurts. To this purpose, the  
82 whole set of ice-cream mixes has been subjected to the following tests:

83 1) EIS measurements with a sinusoidal test signal of amplitude 100 mV in the frequency range 20  
84 Hz to 10 KHz. The acquired spectra have been analyzed to validate the electrical model and the  
85 model parameters estimated with “ad hoc” developed LabVIEW (National Instruments, Austin,  
86 USA) programs using least squares error method.

87 2) The electrical response in the non-linear region has been studied by stimulating the sample with a  
88 sinusoidal test signal of frequency 20 Hz and amplitude in the range 10 mV to 2 V. The measured  
89 data has been fitted to a non-linear empirical model (described in section 5) and the model  
90 parameters calculated by Levenberg-Marquardt algorithm (LMA). LMA is an iterative technique  
91 that locates the minimum of a multivariate function that is expressed as the sum of squares of non-  
92 linear real valued functions. The algorithm needs initial guess for the function parameters to be used  
93 as starting values for the iterative procedure. It has been implemented using built-in project libraries  
94 from LabVIEW.

95 3) pH measurements have been performed by means of a Crison micropH 2000 (Crison  
96 Instruments, South Africa). The instrument has been calibrated before each measure using standard  
97 buffer solutions featuring pH 4 and 7 respectively.

98 Statistical analysis has been carried out with PHStat (Prentice Hall statistical add-on for Microsoft  
99 EXCEL). Student t-test assuming unequal variances and non-parametric Mann Whitney test have  
100 been performed to find significant differences in the measured mean values of measured parameters  
101 (confidence level of 95 %). Multiple regression analysis has been carried out using the Best-Subset  
102 procedure to investigate the correlation between pH values and measured electrical parameters.

103

### 104 **3. Measurement setup**

105 All the measurements of this work are made with Lab instruments.

106 The setup used for the experiments is illustrated in Fig. 1 (a): the product samples are incubated in a  
107 thermal chamber WTC Binder providing the target temperature with an uncertainty of 0.1 °C.  
108 Measures have been carried out at two different temperatures (4 °C and 35 °C). The two  
109 temperatures have been chosen according to the following rules: 4 °C is the standard temperature  
110 the ice-cream mixes are stored while 35 °C is the temperature used in a microbial biosensor system  
111 recently developed by the authors [13]. Thus, the choice has been made with the idea of a future  
112 implementation in industrial environment as a sensor integrated in the tank of the storing machine  
113 (4 °C) or in a separate chamber controlled with the embedded biosensor system (35 °C). However,  
114 the measures at 4 °C resulted in poor repeatability (see Supplementary Material for more  
115 information). This can be related to the fact that at 4 °C the ice-cream mixes are in a semi-viscous  
116 frozen state and also small temperature variations can produce relatively large changes in the  
117 product structure. Thus, in the following, only measures at 35 °C are discussed. Electrical  
118 characteristics of the sample ( $|Z|$  and  $\text{Arg}(Z)$ ) are measured with an LCR meter Agilent E4980A,  
119 controlled via USB interface by a PC system that is also used to acquire measured data and further  
120 data processing. The sample under test is placed in a 10 ml container with cap shaped stainless steel  
121 electrodes. Two types of sensors are used: sensor A (Fig. 1 (b)) consists of two electrodes while  
122 sensor B (Fig. 1 (c)) has four electrodes that are shorted together in couples as shown in the figure.  
123 The difference between the two sensors is related to the generated electric field. Both sensor  
124 geometries have been simulated using the software Comsol Multiphysics v4 (Comsol Inc, Palo  
125 Alto, USA). The electric field distribution is also shown in Fig. 1 (b) and (c). Sensor B is  
126 characterized by more homogeneous electric field with higher values than sensor A for both the  
127 field and its gradient. All the ice-cream mixes have been tested with both types of sensors.

128

### 129 **4. Ice-cream mixes**

130 Measurements have been performed on a set of 21 ice-cream mixes, different for ingredients and  
131 producers, which can be classified in two main categories: fruit and milk based. The latter category  
132 can be further divided in creamy and frozen yogurt products. The ice-cream mixes as well as the  
133 measured pH values are listed in Table 1: those from 1 to 14 are milk based (“creamy” ones from 1  
134 to 10 and frozen yogurt mixes from 11 to 14), while those from 15 to 21 are fruit based. The  
135 composition of the 21 ice-cream mixes is reported in the Supplementary Material.

136 The ice-cream mixes production has been carried out using a Carpigiani Pastomaster RTL machine  
137 to prepare, pasteurize and age ice-cream mixes. The basic steps in the manufacturing are as follows:

- 138 - Mixing of ingredients (i.e. mixing powder with water in the tank of the pasteurizer).
- 139 - Pasteurization.
- 140 - Cooling to 4°C.
- 141 - Aging at 4°C for 10 hours.

142

## 143 **5. Electrical circuit model**

144 EIS has been carried out with a sinusoidal test signal of amplitude ( $V_M$ ) 100 mV on the frequency  
145 range 20 Hz to 10 KHz (logarithmically spaced). Preliminary measurements were performed with  
146 the LCR meter full frequency range (20 Hz to 2 MHz). However, since the high frequency response  
147 resulted in higher noise-to-signal ratio and lower repeatability (see the Supplementary Material for  
148 more details) only the frequency range 20 Hz to 10 KHz is discussed. Fig. 2 (a) shows the Nyquist  
149 plot for three different samples measured with sensor A: the samples are two milk based mixes (#3  
150 and #9 in Table 1) characterized by different fats content and a fruit based mix (#19 in Table 1). As  
151 can be seen a linear relation exists between  $\text{Im}(Z)$  and  $\text{Re}(Z)$ .

152 The electrical model used to fit the data in the investigated frequency range is shown in Fig. 2 (b): it  
153 is composed of a resistance  $R_m$  (accounting for the resistance of both the sample and the interface)  
154 and a constant phase element CPE (resulting from the essentially capacitive component due to the  
155 interface electrode–medium) in series. Since the ice-cream is an unstructured material no distinct

156 dispersion exists within the radiofrequency range up to several MHz. The sample impedance is thus  
 157 purely resistive while the reactive component is essentially due to the electrode interface. The  
 158 impedance of CPE is described by two parameters ( $Q$  and  $\alpha$ ) where  $Q$  represents the double layer  
 159 capacitance, while  $\alpha$  accounts for the non ideal electrode-medium interface (the case  $\alpha=1$  refers to  
 160 an ideal capacitance). The reason for using a CPE instead of a linear capacitor is the non ideal  
 161 behavior of electrode interface [15]. Moreover, the calculated impedance using a linear capacitor  
 162 results in  $\text{Re}(Z)$  to be independent of frequency, contrary to the Nyquist plot of Fig. 2 (a).

163 With the model of Fig. 2 (b) it is:

$$164 \quad Z = R_m + Z_{CPE} = R_m + \frac{1}{Q(j\omega)^\alpha} = R_m + \frac{e^{-j\frac{\pi}{2}\alpha}}{Q\omega^\alpha} = R_m + \frac{\cos(\frac{\alpha\pi}{2})}{Q\omega^\alpha} - j\frac{\sin(\frac{\alpha\pi}{2})}{Q\omega^\alpha} \quad (1)$$

165 Thus:

$$166 \quad \text{Re}(Z) = R_m - \text{ctg}\left(\frac{\alpha\pi}{2}\right) \times \text{Im}(Z) \quad (2)$$

167 The parameters  $R_m$ ,  $Q$  and  $\alpha$  are determined by best fitting the experimental data with the proposed  
 168 electrical model for all ice-cream mixes and both sensors. The determination coefficient  $R^2$  between  
 169 experimental and fitted data is found to be never lower than 0.998, thus validating the electrical  
 170 model. The parameter  $\alpha$  is found to be almost independent on the measured samples with values in  
 171 the range 0.73-0.79 for both sensors. Fig. 3 shows the experimental data ( $|Z|$  and  $\text{Arg}(Z)$ ) as well as  
 172 the curves fitting the model in the case of the vanilla flavored Angelito mix (# 9 in Table 1) and  
 173 sensor A.

174 The electrical characterization in the non-linear region for the electrode-medium system is  
 175 investigated measuring  $|Z|$  with a sinusoidal voltage signal of fixed frequency and  $V_M$  in the range  
 176 10 mV to 2 V (with logarithmic spacing). Fig. 4 shows  $|Z|_{10\text{mV}} - |Z|$  vs.  $V_M$  (logarithmic scale) for  
 177 different frequencies in the case of the Angelito mix (# 9 in Table 1) and sensor A. The value of  $|Z|$   
 178 is almost constant for small signal amplitude (i.e.  $V_M \ll V_{MT}$ , with  $V_{MT}$  never lower than 200 mV),  
 179 while for higher values of  $V_M$ , it decreases linearly with  $\text{Log}_{10}(V_M)$ , thus producing an increase of

180  $|Z|_{10mV} - |Z|$ . The results clearly indicate that increasing the test signal frequency results in an increase  
 181 of the cut-off amplitude  $V_{MT}$  and a decrease of the slope in the non-linear region.

182 In order to characterize the electrical response in the non-linear region for the tested products, the  
 183 curves have been fitted with the empirical model  $|Z| = |Z|_{10mV} + \beta_1 \cdot \text{Log}_{10}(1 + (V_M / V_{MT})^{\beta_2})$ , where  
 184  $|Z|_{10mV}$  is the value of  $|Z|$  for small signal amplitude (i.e. linear response region),  $V_{MT}$  the cut-off  
 185 amplitude (separating the linear from non-linear region) while  $\beta_1$  and  $\beta_2$  are empiric parameters  
 186 used to fit the curve. Fitting procedure has been carried out by LMA. The iterative algorithm has  
 187 been run with the following initial guess for parameters  $|Z|_{10mV}=300$   $\beta_1=0.6$   $V_{MT}=600$   $\beta_2=11$  for all  
 188 ice-cream mixes and both sensors. Fitting procedure resulted in high determination coefficient ( $R^2 >$   
 189  $0.99$ ), thus validating the empirical model. The slope  $\lambda$  in the non-linear region has been estimated  
 190 with the following procedure: when  $V_M \gg V_{MT}$  the empirical model function can be simplified as

191  $|Z| \approx |Z|_{10mV} + \beta_1 \cdot \text{Log}_{10}(V_M / V_{MT})^{\beta_2} = |Z|_{10mV} + \beta_1 \cdot \beta_2 \cdot \text{Log}_{10}(V_M / V_{MT})$ , thus:

$$192 \quad \lambda = \frac{\partial |Z|}{\partial \text{Log}_{10}\left(\frac{V_M}{V_{MT}}\right)} \approx \beta_1 \cdot \beta_2 \quad (3)$$

193 The model parameters are estimated by LMA for  $f = 20$  Hz, since, as shown in Fig. 4, non-linear  
 194 response is stronger (higher values of  $\lambda$ ) at lower frequencies. However, only values of  $\lambda$  are  
 195 reported in section 6, since the other parameters exhibit lower correlation with the ice-cream mixes  
 196 groups.

197

## 198 **6. Results and discussion**

### 199 *6.1. Electrical impedance spectroscopy*

200 The ice-cream mixes have been tested following the procedures described in section 2 and section  
 201 5. The experimental results are illustrated in Fig. 5. Statistical analysis of the presented data  
 202 indicates that significant differences exist between fruit based and both creamy and frozen yogurt  
 203 mixes in the case of  $R_m$ . Fruit based mixes are generally characterized by higher values of  $R_m$  than

204 milk based mixes. In particular, mean values of  $R_m$  for fruit based mixes are 838.9  $\Omega$  and 276.2  $\Omega$   
205 for sensor A and sensor B, respectively. Instead, the corresponding values for creamy mixes are  
206 328.7  $\Omega$  and 95.3  $\Omega$ , while for frozen yogurt products these values are 406.9  $\Omega$  and 74.1  $\Omega$ .  
207 However, a small number of fruit based mixes (banana and kibana based mixes, # 17 and 20,  
208 respectively) exhibit values of  $R_m$  comparable with those of the milk based group, due to the  
209 presence of potassium salts that greatly enhances conductivity.

210 Conductivity for the dairy products is mainly related to fats and salt content: higher concentration of  
211 milk fats results in higher values of  $R_m$ , while the increase in salt concentration leads to resistance  
212 decrease, as can be clearly seen comparing the values of  $R_m$  for mixes 1 and 2. However,  
213 comparison between mixes 2 and 3 clearly indicates that pasteurization temperature also plays a  
214 role in the measured electrical characteristics, since the same mix subjected to high temperature  
215 pasteurization cycle (85 °C) results in a higher value for  $R_m$  than that of the mix pasteurized at  
216 lower temperature (65 °C). On the other hand, it is known that differences in pasteurization  
217 temperature can significantly alter some organoleptic characteristics of the product: for instance,  
218 [16] showed that ice-cream mixes pasteurized with high thermal cycle (between 75 °C and 82 °C)  
219 exhibit lower fat clumping, viscosity and freezing time, and higher protein stability. To investigate  
220 if repeated pasteurization cycles can effectively alter the product electrical characteristics, mix 2 has  
221 been subjected to low temperature pasteurization cycles (65 °C) at time intervals of 1 day and the  
222 electrical parameters have been measured after each cycle : no correlation was observed between  
223 the electrical parameters and the number of pasteurization cycles, thus showing that only thermal  
224 cycling with high temperature significantly affects the product characteristics, while repeated  
225 cycling at lower temperature produces no detectable change.

226 As far as values of  $Q$  are considered (expressed as  $10^6 s^a/\Omega$ ), fruit based and frozen yogurt mixes  
227 exhibit values significantly higher than those of creamy mixes: mean values of 66.6 and 64.6 for  
228 sensor A and 121.9 and 114.7 for sensor B as compared to 51.5 and 93.5 for creamy mixes. Once

229 again, overlapping values of  $Q$  exist among different groups and no significant difference between  
230 frozen yogurt and fruit based mixes is detected.

231 On the whole, the results shown in Fig. 5 indicate that EIS does not allow to reliably discriminate  
232 milk and fruit based ice-creams.

233

### 234 *6.2. Electrical response in the non-linear region*

235 As already anticipated, searching for a method to reliably discriminate the different products  
236 groups, the electrical response in the non-linear region has been investigated.

237 Fig. 6 shows the values of  $\lambda$  in the non-linear region for all mixes and both sensors. As can be seen  
238 fruit based mixes are all characterized by values of  $\lambda$  lower than the other groups: this is particularly  
239 evident for sensor B, that is able to reliably discriminate fruit based mixes, while with sensor A the  
240 gap between values of  $\lambda$  for the different groups is smaller, thus leading to less accurate detection  
241 and possible misclassification of ice-cream mixes with high milk fat content. In the case of frozen  
242 yogurt, the values of  $\lambda$  are not significantly different from those of creamy mixes: thus the two  
243 groups cannot be discriminated with this parameter.

244 These results clearly indicate that measurements of  $\lambda$  provides a reliable method to automatically  
245 discriminate milk based from fruit based ice-creams, as required for automatic machine setting  
246 (with sensor B resulting in larger differences for the values of  $\lambda$  of the two groups and thus more  
247 reliable discrimination). The distinction between creamy and frozen yogurt mixes cannot be  
248 performed by analyzing values of  $\lambda$  (as can be seen in Fig. 6).

249 If required, reliable discrimination between creamy and frozen yogurt products can be achieved by  
250 measuring the mix pH. As can be seen in Table 1, creamy mixes are almost neutral ( $\text{pH} > 6$ ) while  
251 frozen yogurt mixes exhibit pH values in the range 4.5 to 5. Fruit based products, instead, have pH  
252 values in the range 2 to 5, with few exceptions, such as # 16 in Table 1, featuring  $\text{pH} = 6$ , due to a  
253 content of organic acid much lower than other fruit based mixes [17].

254 Since all the three product groups of interest can be discriminated by means of combined  
255 measurements of pH and electrical parameters, correlation between electrical parameters and pH  
256 has been studied.

257

### 258 *6.3. Correlation between electrical and pH measures*

259 The correlation between measured electrical parameters  $R_m$ ,  $Q$  and  $\lambda$  for both types of sensors and  
260 pH has been investigated, and the results are presented in Supplementary Material, where the values  
261 of pH are plotted versus the corresponding electrical parameter for all ice-cream mixes and both  
262 sensors. The results indicate poor correlation between pH and the corresponding electrical  
263 parameter, especially  $R_m$ , where a determination coefficient as low as 0.216 and 0.17 (for sensor A  
264 and B respectively) is found.

265 Better results are obtained for the correlation of pH with  $\lambda$  (0.444 and 0.489) and  $Q$  (0.533 and  
266 0.492). However, the determination coefficient never higher than 0.533 is not satisfactory,  
267 preventing a reliable estimate of pH with electrical measures. Multiple regression analysis has been  
268 carried out to test if expressing pH as linear function of more than a single electrical parameter  
269 significantly increases the correlation. The best results are obtained (for both sensors) by using  $Q$   
270 and  $\lambda$  as independent variables, namely:  $\text{pH} = b_0 + b_1 * Q + b_2 * \lambda$ , where  $b_0$ ,  $b_1$  and  $b_2$  are numerical  
271 parameters. In this way, however, the determination coefficient  $R^2$  (corrected for the use of multiple  
272 variables) increases only slightly in the case of the data obtained with sensor A (0.621) while no  
273 improvement is found in the case of sensor B.

274 Thus, pH can not be reliably inferred by electrical parameters  $R_m$ ,  $Q$  and  $\lambda$ .

275 The overall results from the study are presented in Table 2. Regarding the primary discrimination  
276 between milk based and fruit based mixes, although the resistive component of the impedance  $R_m$  is  
277 characterized by statistical significantly different values for the two groups, the discrimination is not  
278 reliable due to few fruit based mixes characterized by higher conductivity. On the contrary, the  
279 parameter  $\lambda$  measured in the non-linear region can provide reliable discrimination between the two

280 groups (in particular using sensor B). The second level discrimination of milk based mixes in  
281 creamy mixes and frozen yogurts can be reliably achieved by pH measure, while electrical  
282 parameter Q (although characterized by significantly different values for the two subgroups) doesn't  
283 provide a reliable solution due to overlapping values between the two subgroups.

284

## 285 **7. Conclusions**

286 In this paper the possibility to discriminate different groups of ice-cream mixes by means of  
287 electrical measurements, so as to allow automatic setting of product storing machines has been  
288 studied. To this purpose, a distinction between milk and fruit based products is essential, and  
289 sufficient for most practical purposes. Furthermore, within the first category it is sometimes  
290 required to discriminate frozen yogurt products from the remaining (creamy) ice-creams.

291 To reach the goal, this work has investigated the possibility to use Electrical Impedance  
292 Spectroscopy (performed by stimulating the sample with a sinusoidal test signal of amplitude 100  
293 mV and frequency in the range 20 Hz to 10 KHz) showing that it does not provide a reliable  
294 solution.

295 Instead, electrical characterization in the non-linear region (obtained with a sinusoidal test signal of  
296 frequency 20 Hz and amplitude in the range 10 mV to 2 V) is shown to do the work as far as the  
297 basic distinction between milk and fruit based products is concerned.

298 As the second level distinction within the first category, it can be done measuring the pH values of  
299 the products, which is lower for frozen yogurt than for the creamy mixes.

300 Although the experiments of this work have been carried out with Lab instruments, measurements  
301 of electrical response in the non-linear region can be implemented in the form of a low-cost  
302 electronic board, and this holds also for pH determination. Thus, the present work provides the  
303 fundamentals for a possible future development of an embedded system that can open the road for  
304 fully automatic industrial ice-cream machines.

305

306 **Acknowledgements**

307 The authors thank Dr. Anna Pompei (Department of Pharmaceutical Sciences, University of  
308 Bologna) for pH measurements and many helpful discussions.

309

310 **References**

311

312 [1] M. Chanet, C. Riviere, P. Eynard, Electric impedance spectrometry for the control of  
313 manufacturing process of comminuted meat products. *Journal of Food Engineering* 42 (1999) 153-  
314 159

315

316 [2] R. Zywica, G. Pierzynowska-Korniak, J. Wojcik, Application of food products electrical model  
317 parameters for evaluation of apple puree dilution. *Journal of Food Engineering* 67 (2005) 413-418

318

319 [3] Y. Kitamura, K. Toyoda, B. Park, Electric impedance spectroscopy for yogurt processing. *Food  
320 Science and Technology Research* 6 (4) (2000) 310-313

321

322 [4] A. Cataldo, E. Piuzzi, G. Cannazza, E. De Benedetto, L. Tarricone, Quality and anti-adulteration  
323 control of vegetable oils through microwave dielectric spectroscopy. *Measurement* 43 (2010) 1031-  
324 1039

325

326 [5] M.F. Mabrook, M.C. Petty, Application of electrical admittance measurements to the quality  
327 control of milk. *Sensors and Actuators B* 84 (2002) 136-141

328

329 [6] J. Yang, M. Huang, J. Peng, J. Shi, Rapid determination of moisture content in milk powder by  
330 microwave sensor. *Measurement* (2010) doi: 10.1016/j.measurement.2010.08.007

331

- 332 [7] B.A. Lawton, R. Pethig, Determining the fat content of milk and cream using ac conductivity  
333 measurements. *Measurement Science and Technology* 4 (1993) 38-41  
334
- 335 [8] E. Norberg, H. Hogeveen, I.R. Korsgaard, N.C. Friggens, K.H.M.N. Sloth, P. Lovendahl,  
336 Electrical conductivity of milk : ability to predict mastitis status. *Journal of Dairy Science* 87 (2004)  
337 1099-1107  
338
- 339 [9] F.J. Ferrero, G. Grillo, M.A. Perez, J.C. Anton, J.C. Campo, Design of low cost mastitis detector  
340 in cows by measuring electrical conductivity of milk. *IEEE Instrumentation and Measurement*  
341 *Technology Conference* (2002) 375-378  
342
- 343 [10] C.J. Felice, R.E. Madrid, J.M. Olivera, V.I. Rotger, M.E. Valentinuzzi, Impedance  
344 microbiology: quantification of bacterial content in milk by means of capacitance growth curves.  
345 *Journal of Microbiological Methods* 35 (1999) 37-42  
346
- 347 [11] M. Grossi, M. Lanzoni, A. Pompei, R. Lazzarini, D. Matteuzzi, B. Riccò, Detection of  
348 microbial concentration in ice-cream using the impedance technique. *Biosensors & Bioelectronics*  
349 23 (2008) 1616-1623  
350
- 351 [12] M. Grossi, A. Pompei, M. Lanzoni, R. Lazzarini, D. Matteuzzi, B. Riccò, Total bacterial count  
352 in soft-frozen dairy products by impedance biosensor system. *IEEE Sensors Journal* 9 (10) (2009)  
353 1270-1276  
354
- 355 [13] M. Grossi, M. Lanzoni, A. Pompei, R. Lazzarini, D. Matteuzzi, B. Riccò, An embedded  
356 portable biosensor system for bacterial concentration detection. *Biosensors & Bioelectronics* 26  
357 (2010) 983-990

358  
359  
360  
361  
362  
363  
364  
365  
366  
367  
368  
369  
370  
371  
372  
373  
374  
375  
376  
377  
378  
379  
380  
381  
382  
383

[14] E.T. McAdams, A. Lacknermeier, J.A. McLaughlin, D. Macken, The linear and non-linear electrical properties of the electrode-electrolyte interface. *Biosensors & Bioelectronics* 10 (1995) 67-74

#	Ice-cream mix	pH
---	---------------	----

[15] W.H. Mulder, J.H. Sluyters, T. Pajkossy, I. Nyikos, Tafel current at fractal electrodes. Connection with the admittance spectra. *Journal of Electroanalytical Chemistry* 285 (1990) 103-115

[16] L.R. Dowd, E.O. Anderson, Study of short-time-high-temperature pasteurization of ice cream mix. *Journal of Dairy Science* 26 (1) (1943) 37-46

[17] S. Gurrieri, L. Miceli, C.M. Lanza, F. Tommaselli, R.P. Bonomo, E. Rizzarelli, Chemical characterization of sicilian prickly pear (*opuntia ficus indica*) and perspectives for the storage of its juice. *Journal of Agricultural and Food Chemistry* 48 (2000) 5424-5431

384	1	<i>Soft serve mix (low fat content – pasteurization at 65 °C)</i>	6,2
385	2	<i>Soft serve mix (high fat content – pasteurization at 65 °C)</i>	6,3
386	3	<i>Soft serve mix (high fat content – pasteurization at 85 °C)</i>	6,2
387	4	<i>Egg based ice-cream mix</i>	6,6
388	5	<i>Fiordilatte ice-cream mix</i>	6,6
389	6	<i>Chocolate ice-cream mix</i>	7
390	7	<i>Fabbri soft serve Chocolate mix</i>	6,8
391	8	<i>Pregel soft serve Chocolate mix</i>	6,7
392	9	<i>Angelito Vanilla Flavour Dairy Ice Cream Mix</i>	6,4
393	10	<i>Mondi ice-cream mix</i>	6,9
394	11	<i>Pregel Yogursprint mix</i>	4,6
395	12	<i>Pregel Yogursprint mix + fresh yogurt</i>	4,4
396	13	<i>Yogurt mix</i>	5,1
397	14	<i>Yogurt soft serve mix</i>	5,2
398	15	<i>Orange based ice-cream mix</i>	3,5
399	16	<i>Prickly pear based ice-cream mix</i>	6,2
400	17	<i>Banana based ice-cream mix</i>	4,8
401	18	<i>Strawberry based ice-cream mix</i>	3,6
402	19	<i>Pear based ice-cream mix</i>	4,5
403	20	<i>Kibana based ice-cream mix</i>	3,8
404	21	<i>Lemon based ice-cream mix</i>	2,7

399

400

401

402

403

404

405 **Table 1** Ice-cream mixes tested in this work as well as measured values of pH.

406

407

408

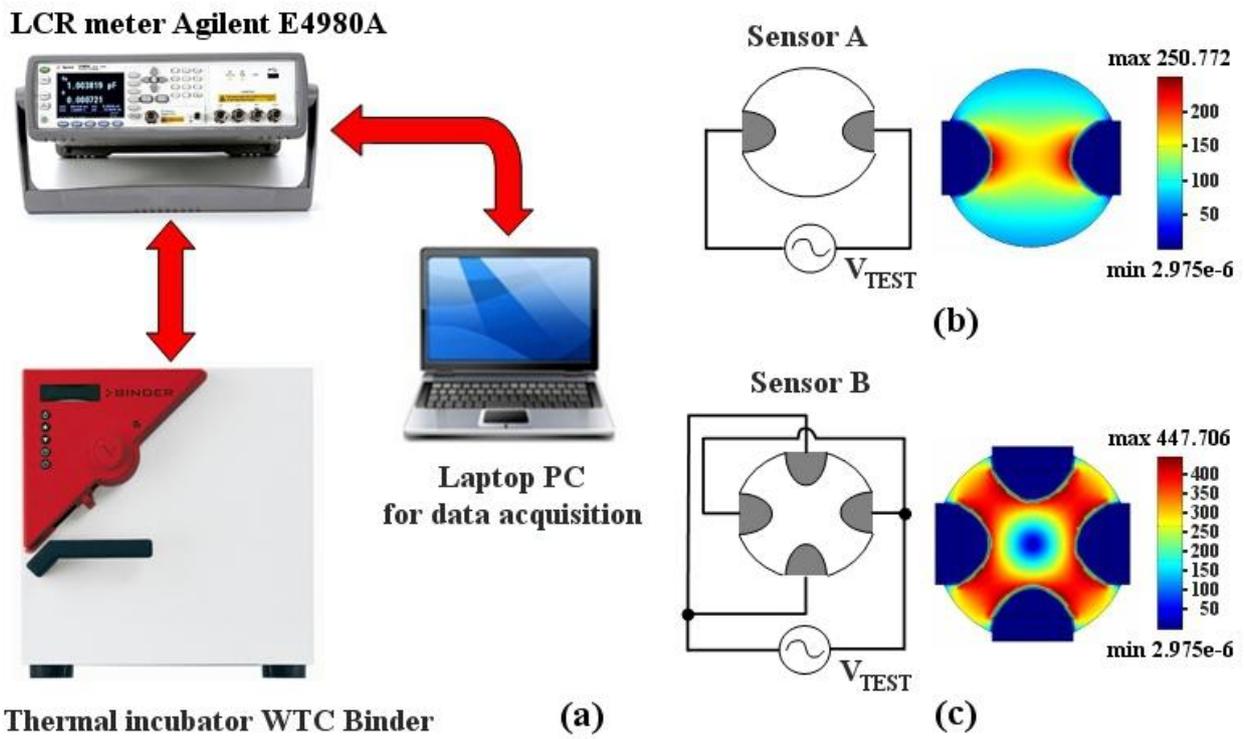
409

410  
411  
412  
413  
414  
415  
416  
417  
418  
419  
420  
421  
422  
423  
424  
425  
426  
427  
428  
429  
430  
431  
432  
433

	SENSOR A			SENSOR B			pH
	R <sub>m</sub>	Q	λ	R <sub>m</sub>	Q	λ	
Milk based mixes/Fruit based mixes	*		X	*		X	*
Creamy mixes/Frozen Yogurt mixes		*			*		X

**Table 2** Feasibility of ice-cream mixes discrimination based on the measures of electrical parameters and pH. X indicates that the corresponding parameter is suitable for the discrimination of the corresponding groups. \* indicates significantly different values between the two groups but non reliable discrimination due to overlapping values.

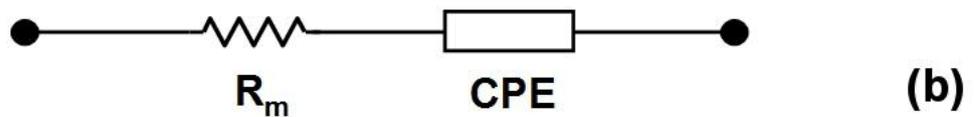
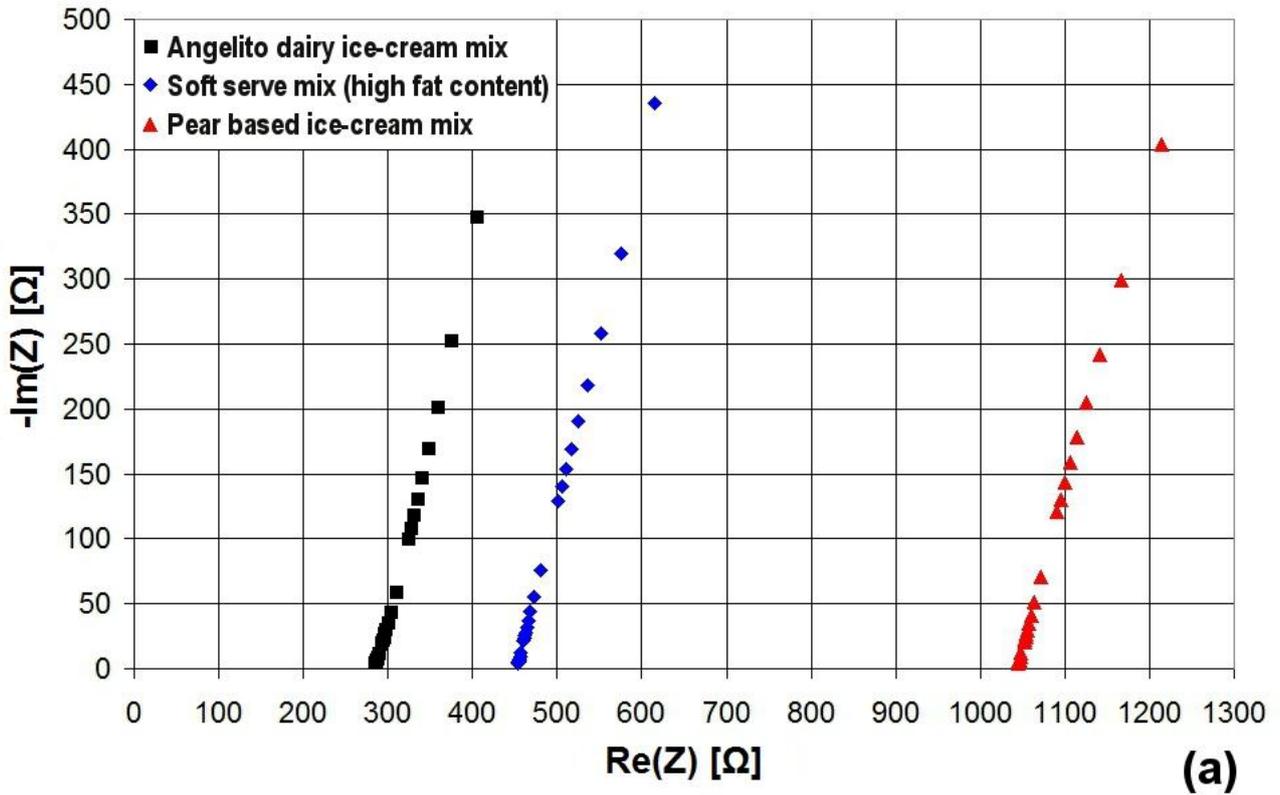
434  
435  
436  
437  
438  
439



440  
441  
442  
443  
444  
445  
446  
447  
448  
449

**Fig. 1** Measurement setup used in the electrical characterization of ice-cream mixes (a). Geometries and simulations of the generated electric field for sensor A (b) and sensor B (c).

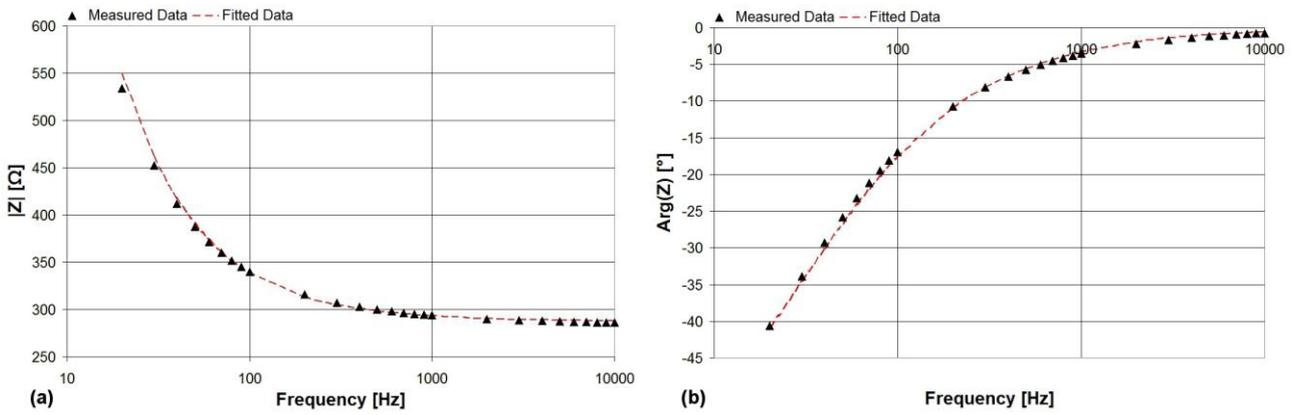
450  
451  
452  
453  
454



455  
456 **Fig. 2** Nyquist plot for three different ice-cream mixes (a) and electrical circuit used to model the  
457 sensors electrical response (b).

458  
459  
460  
461

462  
463  
464  
465  
466  
467



468

469 **Fig. 3**  $|Z|$  (a) and  $\text{Arg}(Z)$  (b) vs. frequency of the applied test signal for the Angelito mix (# 9 in  
470 Table 1) and sensor A. High correlation ( $R^2 > 0.998$ ) is achieved for all ice-cream mixes and both  
471 sensors between measured data and the electrical model of Fig. 2 (b).

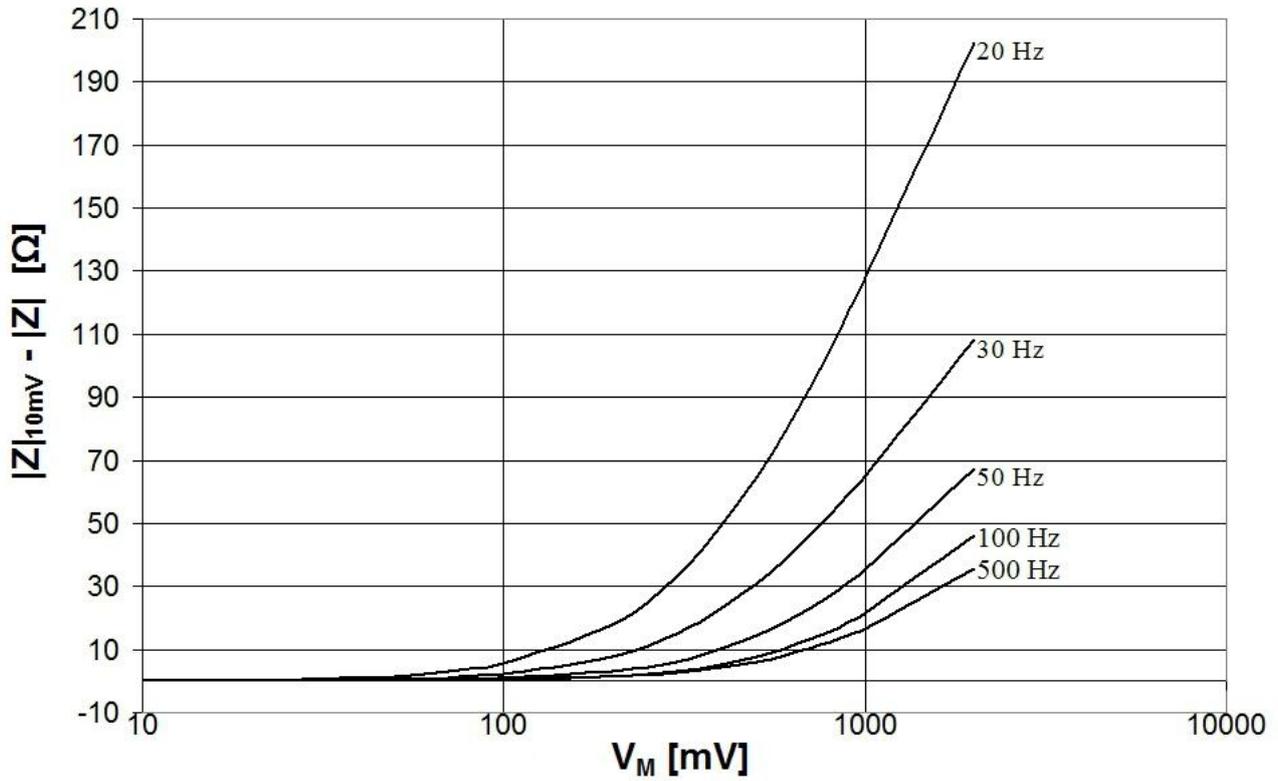
472  
473  
474  
475  
476  
477  
478  
479  
480  
481  
482

483

484

485

486



487

488 **Fig. 4**  $|Z|_{10mV} - |Z|$  vs. the amplitude  $V_M$  of the applied test signal for different frequencies in the  
489 case of the Angelito mix (# 9 in Table 1) and sensor A. Non-linear response is stronger at low  
490 frequencies.

491

492

493

494

495

496

497

498

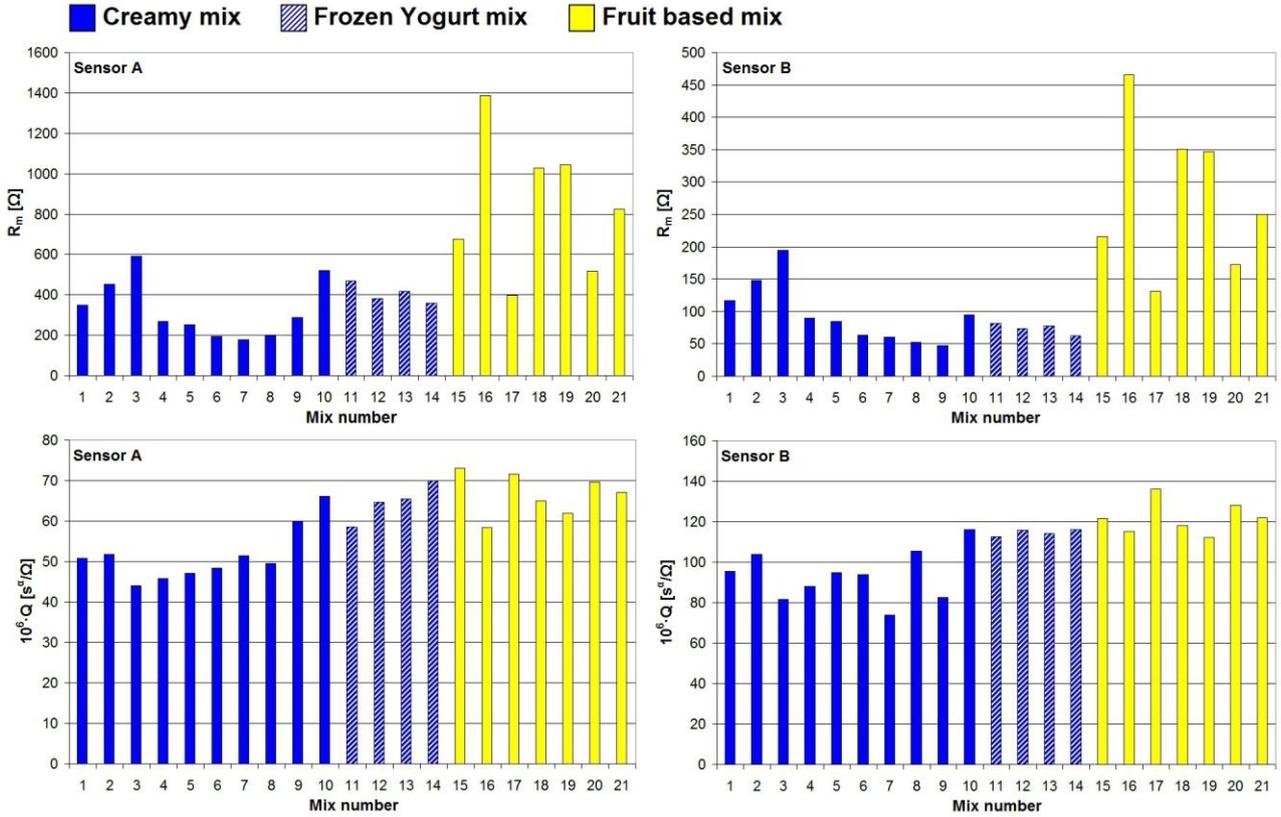
499

500

501

502

503



504

505 **Fig. 5** Histograms of  $R_m$  and  $Q$  for all ice-cream mixes and both sensors used in this work. Creamy

506 mixes bars are blue colored, frozen yogurt mixes are blue/white colored and fruit based ice-cream

507 mixes are yellow colored.

508

509

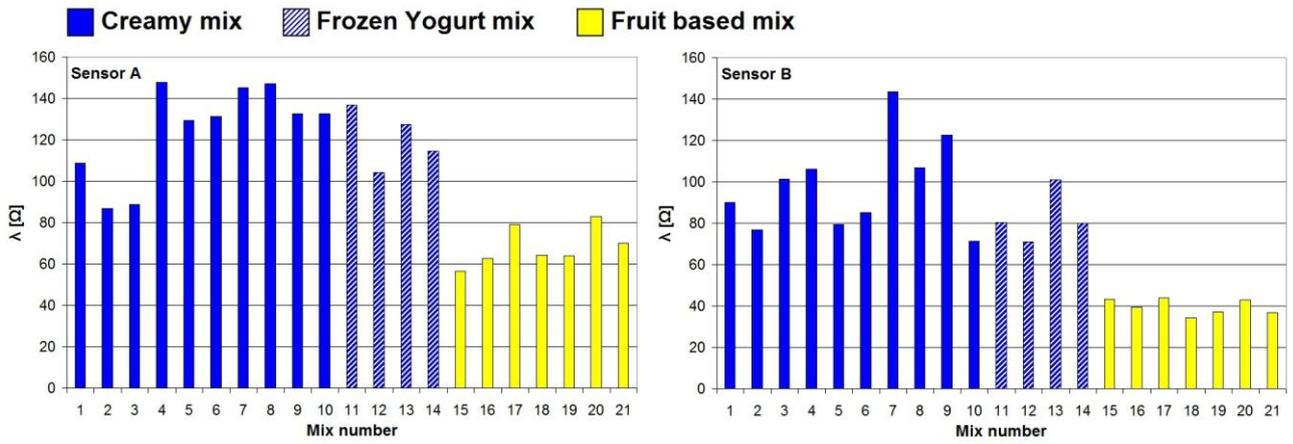
510

511

512

513

514  
515  
516  
517  
518  
519



520

521 **Fig. 6** Histograms of  $\lambda$  for all ice-cream mixes and both sensors. Creamy mixes bars are blue  
522 colored, frozen yogurt mixes are blue/white colored and fruit based ice-cream mixes are yellow  
523 colored.

524

525

526

527

528

529

530

531

532

533

## Supplementary material

### Automatic Ice-Cream Characterization by Impedance Measurements for Optimal Machine Setting

Marco Grossi, Massimo Lanzoni, Roberto Lazzarini, Bruno Riccò

#### 1. Ice-cream mixes

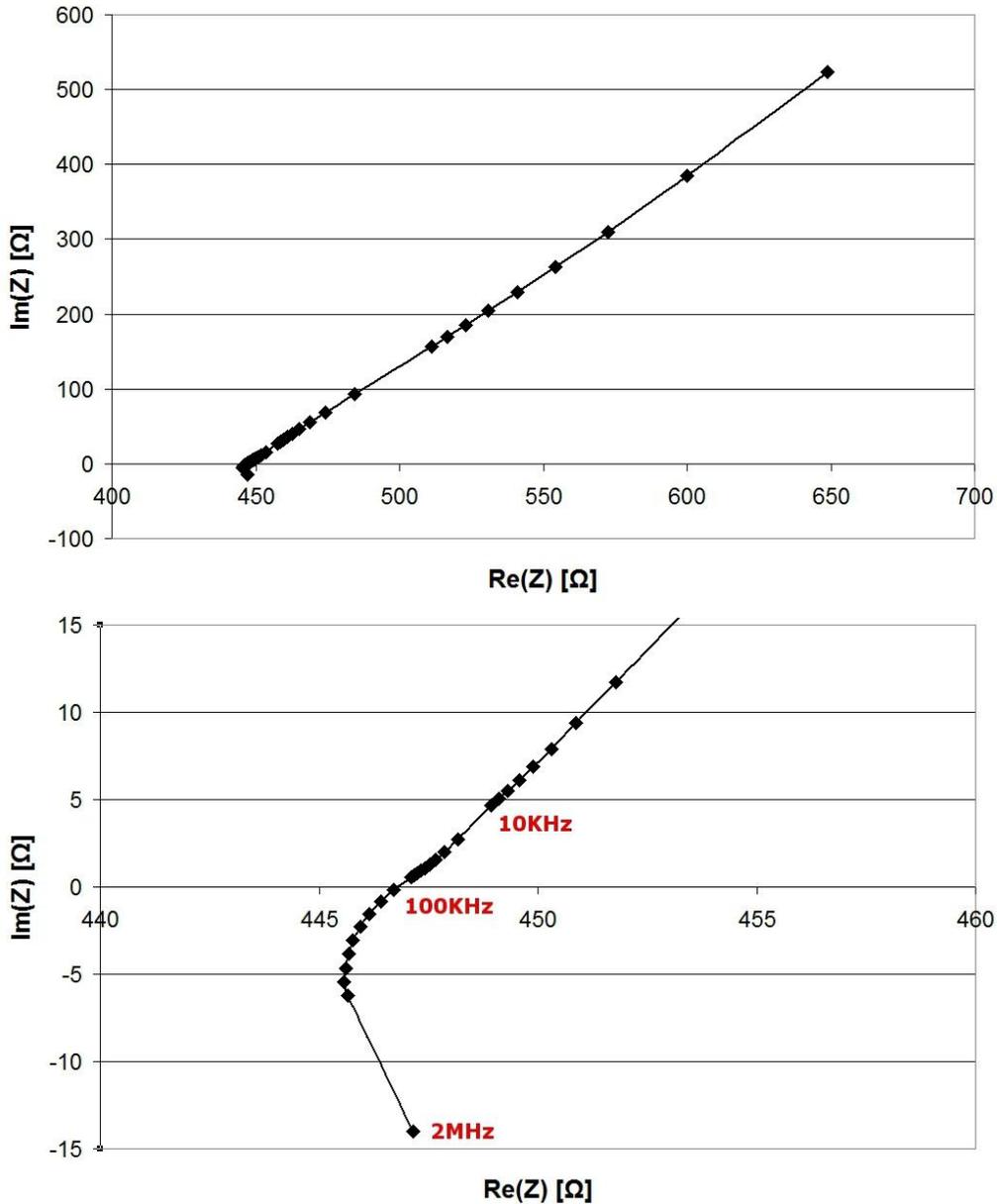
Measurements have been performed on a set of 21 ice-cream mixes, different for ingredients and producers. In the following, the recipes for every mix used in the study are presented.

#	Ice-cream mix	Composition
1	<i>Soft serve mix (low fat content – pasteurization at 65 °C)</i>	Water, Skimmed Milk Powder, Whipping Cream, Pregel Base Diamant 100 and Sugar in the following composition: Sugar (11.9%), Fat (4%), Milk Solids-nonfat (8.9%), Stabilizers (0.3%)
2	<i>Soft serve mix (high fat content – pasteurization at 65 °C)</i>	Water, Skimmed Milk Powder, Whipping Cream, Pregel Base Diamant 100 and Sugar in the following composition: Sugar (17.2%), Fat (11.2%), Milk Solids-nonfat (12%), Stabilizers (0.3%)
3	<i>Soft serve mix (high fat content – pasteurization at 85 °C)</i>	Water, Skimmed Milk Powder, Whipping Cream, Pregel Base Diamant 100 and Sugar in the following composition: Sugar (17.2%), Fat (11.2%), Milk Solids-nonfat (12%), Stabilizers (0.3%)
4	<i>Egg based ice-cream mix</i>	Whole Milk (66.5%), Skimmed Milk Powder (8%), Fresh Whipping Cream @ 35% (8.2%), Sucrose (15.6%), Dextrose (2%), Egg Yolk (7.5%), Pregel Base Diamant 50 (3.4%)
5	<i>Fiordilatte ice-cream mix</i>	Whole Milk (70%), Skimmed Milk Powder (1.3%), Fresh Whipping Cream @ 35% (8.2%), Sucrose (15.4%), Dextrose (1.7%), Pregel Base Diamant 50 (3.4%)
6	<i>Chocolate ice-cream mix</i>	Whole Milk (63%), Fresh Whipping Cream @ 35% (10%), Sucrose (15.6%), Dextrose (2%), Cocoa Powder (24%), Pregel Base Diamant 50 (3.4%)
7	<i>Fabbri soft serve Chocolate mix</i>	Sugar, Cocoa Powder, Whole Milk Powder, Skimmed Milk Powder, Maltodextrins, Stabilisers (E412, E466), Emulsifiers (E471), Flavouring in the following composition: 2.25 liters by water and 1 Kg by powder
8	<i>Pregel soft serve Chocolate mix</i>	Sugar, Cocoa Powder, Skimmed Milk Powder, Hydrogenated Vegetable Fat, Dextrose, Dehydrated Glucose Syrup, Stabilisers (E412, E410, E466), Emulsifiers (E471, E472a, E472b), Acidifier (E330)
9	<i>Angelito Vanilla Flavour Dairy Ice Cream Mix</i>	Skimmed Milk, Sugar, Butter, Skimmed Milk Powder, Dried Glucose, Syrup, Emulsifiers (E477, E471), Stabilisers (E466, E412, E407, E451), Flavouring
10	<i>Mondi ice-cream mix</i>	Sugar, Coconut Oil, Dextrose, Glucose Powder, Milk Proteins, Stabilisers (E412, E410, E466), Emulsifiers (E471, E473), Flavouring in the following composition: 17.5 Kg by powder and 30 liters by water
11	<i>Pregel Yogursprint mix</i>	Sugar, Dextrose, Skimmed Milk Powder, Skimmed Yogurt Powder, Maltodextrins, Acidifier (E330), Flavouring, Hydrogenated Vegetable Fat, Stabilisers (E412, E410, E466), Emulsifiers (E471, E472a, E472b, E477) in the following composition: 2.5 liters by Whole Milk and 1 Kg by powder
12	<i>Pregel Yogursprint mix + fresh yogurt</i>	Sugar, Dextrose, Skimmed Milk Powder, Skimmed Yogurt Powder, Maltodextrins, Acidifier (E330), Flavouring, Hydrogenated Vegetable Fat, Stabilisers (E412, E410, E466), Emulsifiers (E471, E472a, E472b, E477) in the following composition: 2 liters by Whole Milk, 500 g by Fresh Skimmed Yogurt and 1 Kg by powder
13	<i>Yogurt mix</i>	Skimmed Yogurt (2 liters), White Base (1 liters), Sucrose (450 gr), Skim Solids (8 gr) with White Base in the following composition: Milk (1 liters), Skimmed Milk Powder (50 gr), Whipping Cream (258 gr), Sucrose (250 gr), Dextrose (25 gr), Glucose (33 gr), Skim Solids (8 gr), Proteins (25gr)
14	<i>Yogurt soft serve mix</i>	Whole Milk (482 gr), Skimmed Yogurt (300 gr) and Yosoft Powder (220 gr) with the following ingredients: Sucrose, Vegetable Fiber, Skimmed Milk Powder, Skimmed Yogurt Powder, Dextrose, Emulsifiers (E471, E472b, E472, E477), Acidifier (E330), Stabilisers (E410), Flavouring
15	<i>Orange based ice-cream mix</i>	Orange (69.3%), Water (4.9%), Sugar (20.8%), Fruit Base 50 (3.5%), Lemon Juice (1.5%)
16	<i>Prickly pear based ice-cream mix</i>	Prickly Pear (50%), Water (25.5%), Sugar (22%), Fruit Base 50 (2.5%), Lemon Juice (1.5%)
17	<i>Banana based ice-cream mix</i>	Banana (50%), Water (27.6%), Sugar (20.4%), Fruit Base 50 (2%), Lemon Juice (1.5%)
18	<i>Strawberry based ice-cream mix</i>	Strawberry (50%), Water (23.5%), Sugar (24%), Fruit Base 50 (2.5%), Lemon Juice (1.5%)
19	<i>Pear based ice-cream mix</i>	Pear (50%), Water (25%), Sugar (22.5%), Fruit Base 50 (2.5%), Lemon Juice (1.5%)
20	<i>Kibana based ice-cream mix</i>	Kiwi (35%), Banana (15%), Water (24.9%), Sugar (23%), Fruit Base 50 (2.1%), Lemon Juice (1.5%)
21	<i>Lemon based ice-cream mix</i>	Lemon Mashed Fruit Pulp (69.3%), Water (4.9%), Sugar (20.8%), Fruit Base 50 (3.5%), Lemon Juice (1.5%)

544 **2. Electrical characterization in the frequency range 20 Hz to 2 MHz**

545 Preliminary measurements have been carried out on a limited number of ice-cream mixes using a  
546 sinusoidal test signal of amplitude 100 mV and frequency in the range 20 Hz to 2 MHz for the  
547 incubation temperature of 35 °C. The Nyquist plot for the entire frequency range as well as a  
548 particular of the higher frequencies are shown in Fig. S1 for mix # 2 and sensor A (different mixes  
549 and sensor B result in similar behavior).

550



551

552 **Fig. S1** Nyquist plot in the frequency range 20 Hz to 2 MHz for mix # 2 and sensor A at 35 °C.

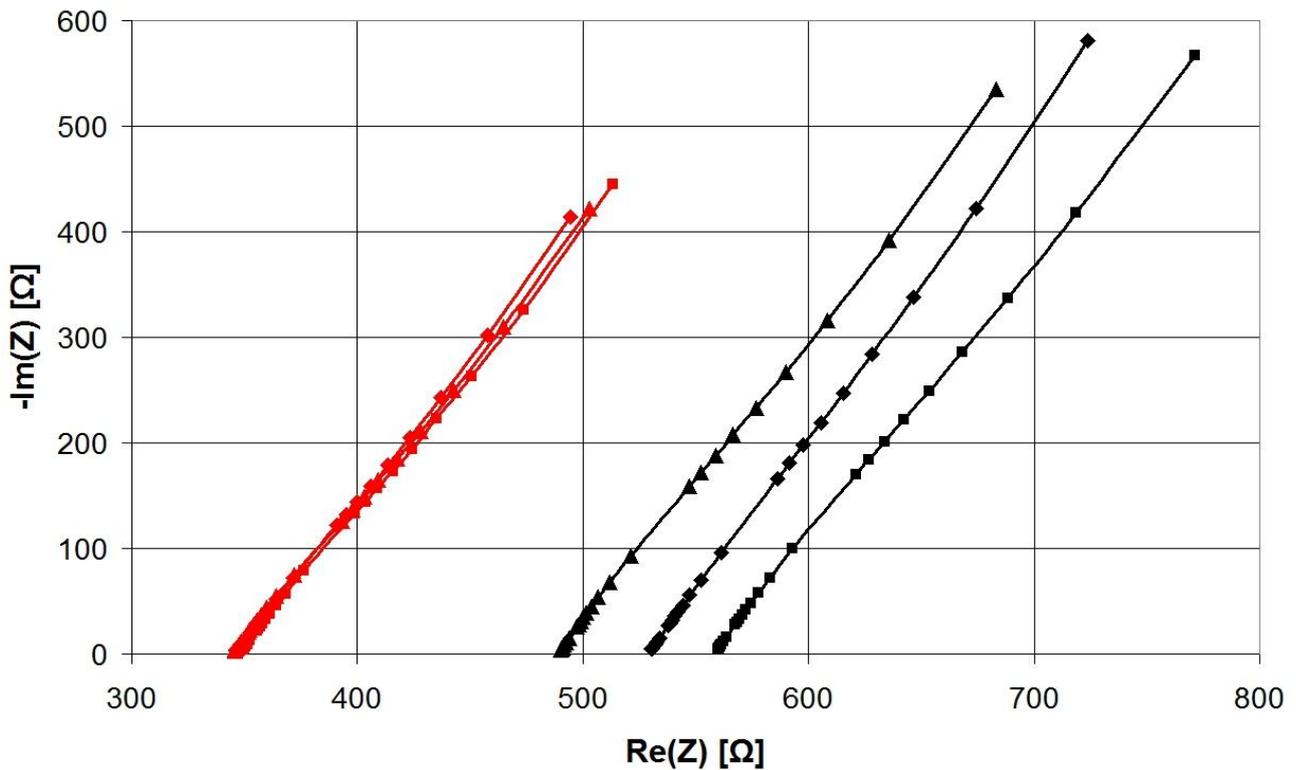
553

554 The results from Fig. S1 clearly indicates that the electrical model featuring a resistance ( $R_m$ ) and a  
555 CPE ( $Q$ ) in series is adequate for frequencies up to 400 kHz. For higher frequencies a deviation  
556 from the model occurs. The real component of the impedance  $\text{Re}(Z)$  was almost the same at 10  
557 kHz and 2 MHz (differences lower than 0.2%). Repeated measures on the same ice-cream mix (# 2)  
558 were carried out to test the repeatability of the measures. The results show that  $\text{Re}(Z)$  results in  
559 comparable repeatability at 10 kHz and 2 MHz (with a ratio of standard deviation to mean value  $\sigma/\mu$   
560 of 0.03), while the imaginary component  $\text{Im}(Z)$  resulted in higher dispersion at higher frequency  
561 (with a  $\sigma/\mu$  value of 0.19 at 2 MHz almost twice than at 10 kHz). The higher dispersion at higher  
562 frequencies can be related to some parasitic effects in the sensor that have to be further

563 investigated. Since the less repeatability of the measures and the fact that, from preliminary  
 564 measurements, data on the extended frequency range don't provide further informations, the  
 565 investigated frequencies in the paper has been limited to the range 20 Hz to 10 kHz.  
 566  
 567

568 **3. Measurements at 4 °C and 35 °C: results comparison**

569 Mix # 2 from Table 1 in the paper has been analyzed in triplicate using EIS with a sinusoidal test  
 570 voltage of amplitude 100 mV and frequency range from 20 Hz to 10 kHz with both sensors. In Fig.  
 571 S2 the Nyquist plot is shown for the three measures at both temperatures of 4 °C and 35 °C for  
 572 sensor A. Measures at 4 °C resulted in less repeatability than 35 °C. This can be related to the fact  
 573 that at 4 °C the ice-cream mixes are in a semi viscous frozen state and also small temperature  
 574 variations can produce relatively large changes in the product structure.  
 575



576  
 577  
 578 **Fig. S2** Nyquist plot in the frequency range 20 Hz to 10 kHz for mix # 2 (measures in triplicate) and  
 579 sensor A at two different incubation temperatures: 4 °C and 35 °C.  
 580

581 Statistical analysis has been carried out and mean value  $\mu$  as well as the ratio of standard deviation  
 582 to mean value  $\sigma/\mu$  have been calculated for both sensors and temperatures. The results are  
 583 presented in the following table.  
 584  
 585

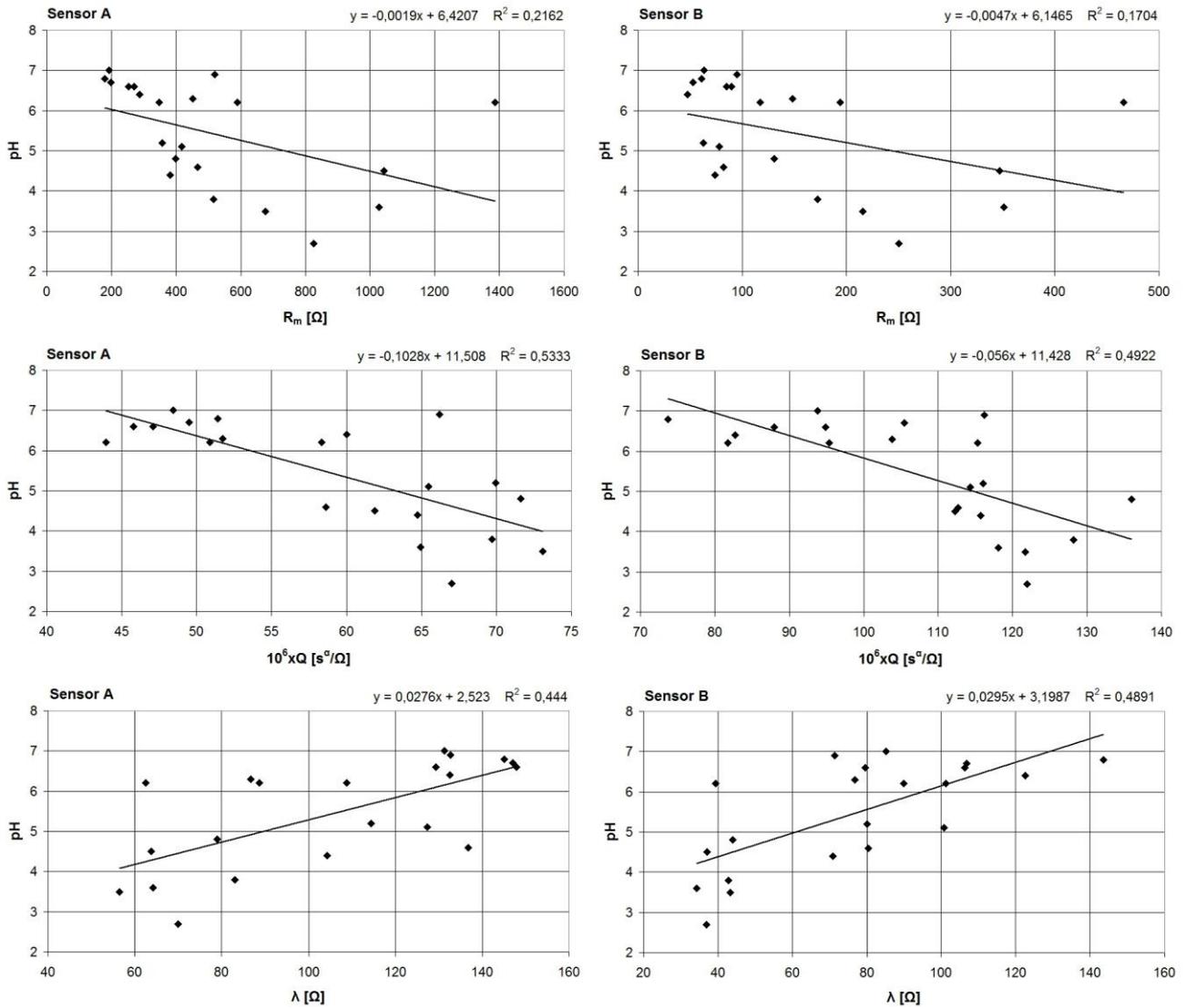
		SENSOR A		SENSOR B	
		T = 4 °C	T = 35 °C	T = 4 °C	T = 35 °C
$R_m$	$\mu$ [Ω]	524.7	345.5	186	116.7
	$\sigma/\mu$	0.066	0.005	0.055	0.005
Q	$\mu$ [ $10^6 s^{\alpha}/\Omega$ ]	38.7	52.7	71.3	86.7
	$\sigma/\mu$	0.08	0.03	0.18	0.09

586  
 587

588  
589  
590  
591  
592

#### 4. Correlation between pH and electrical measures

The correlation between measured electrical parameters  $R_m$ ,  $Q$  and  $\lambda$  for both type of sensors and pH has been investigated, and the results are presented in Fig. S3, where the values of pH are plotted versus the corresponding electrical parameter for all products tested and both sensors.



593  
594  
595  
596  
597  
598  
599

**Fig. S3** Scatter plots of pH values vs.  $R_m$ ,  $Q$  and  $\lambda$  for all tested mixes and both sensors. Linear regression lines as well as determination coefficient  $R^2$  have been calculated.

600