

Short Review Article

Application of Arylglyoxals in Synthesis of Pyrrolo[2,3-*d*]pyrimidines via Multicomponent Reactions



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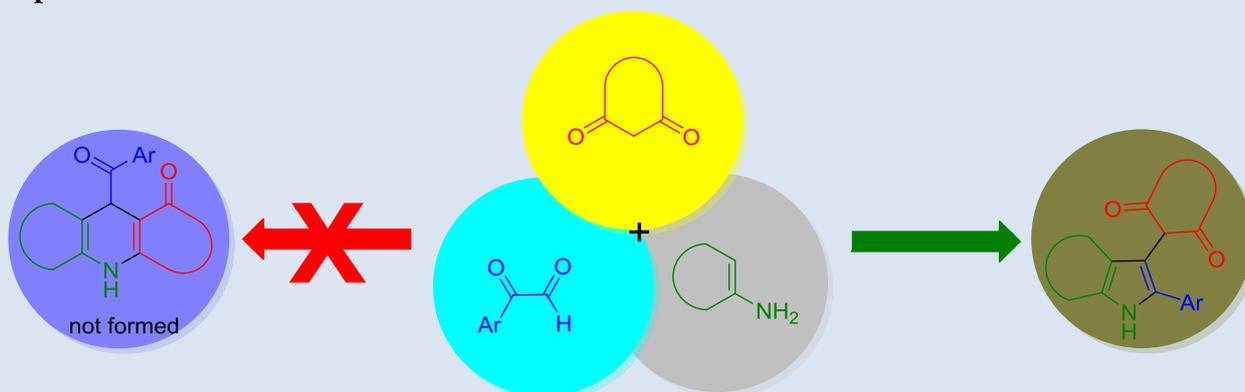
Abstract:

This review provides an overview of the recent literature on application of arylglyoxals the synthesis of pyrrolo[2,3-*d*]pyrimidines via multicomponent reactions in the period of 2008–2018. 1,2-Dicarbonyl compounds are attractive precursors for synthesis of various heterocyclic compounds, and arylglyoxals are frequently applied in synthesis of various organic compounds, and in particular of pyrrolo[2,3-*d*]pyrimidines derivatives, which are important due to their biological and pharmaceutical activities.

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Keywords: Arylglyoxals, Pyrrolo[2,3-*d*]pyrimidines, Multicomponent reactions, Enamines

Graphical Abstract:



Biography:



Ramin Javahershenas was born in Iran, in 1971. He received his B.Sc. degree in Applied Chemistry from Tabriz University, Tabriz, Iran, in 1993, his M.Sc. degree in Organic Chemistry from the Urmia University, Urmia, Iran, under the supervision of Professor Naser Ardabilchi in 1999 and his Ph.D. degree in Organic chemistry from Urmia University, Urmia, Iran under the supervision of Professor Jabbar Khalafy, in 2017. His research interests include organic synthesis, heterocyclic synthesis, asymmetric synthesis, natural products synthesis, synthetic methodology and applications of various catalysts in multicomponent reactions.

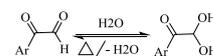


As the natural bases do not have any fluorescence, several groups have been attached or the base itself has been modified [49,50]. Although these heterocyclic compounds have been known since the middle of the twentieth century [51,52], they were not extensively studied until the last few decades. More recently, the interest of the chemical and pharmaceutical industry in heterocyclic fused pyrimidines, also named deazapurines, has increased notably, resulting in a large increase in the number of patents, research papers, and reviews, all of which led to the introduction of several drugs in the market or late clinical stages (Scheme 1) [53-54].

1.1. Aryl glyoxals

Phenylglyoxal (PG), the simplest member of this family, is a yellow liquid that polymerizes upon standing. Upon heating, it loses a molecule of water and the polymeric material changes to the aldehyde form or anhydrous AG. To form the colorless crystalline hydrate, PG should be recrystallized in hot water. The

AG-hydrate appears to contain one molecule of water (Scheme 2).



Scheme 2. Structures of Aryl glyoxals

AGs contain aldehyde and ketone functional groups with different reactivity, the reactivity of the aldehyde group is greater than that of benzaldehyde because of the electron-withdrawing keto group and reacts quickly with different nucleophiles, the resulting product then undergoing cyclization in a number of ways. The resulting products have received considerable attention due to their biological and pharmacological activities, such as selective bronchodilators such as salbutamol and terbutaline, used for their selective and antiviral activity in the embryonated egg against several viruses, including influenza (PR-8) and newcastle disease (NJKD strain) viruses [55-57].

Table 1. Various methods for the synthesis of AGs

Entry	Method	Condition	Ref.
1	Oxidation of aryl methyl ketones	SeO ₂ , dioxane-water, reflux	[56], [58-61]
		H ₂ SeO ₃ , dioxane-water, reflux, 4 h	[62]
		SeO ₂ , EtOH, 10% HNO ₃ (aq), 90 °C, 1h	[63]
		(PhSe) ₂ , (NH ₄) ₂ S ₂ O ₈ , MeOH, reflux, 1-4h	[64]
		48% HBr (aq), DMSO, 55 °C, 0.5-24 h	[65]
2	Oxidation of phenacylbromides	DMSO, rt, 9 h	[66]
		α-picoline n-oxide, 0 °C, then Na ₂ CO ₃ , water	[67]
3	Oxidation of phenacyl nitrate esters	Et ₂ NOH, MeOH, reflux, 2 h	[68]
		NaOAc·3H ₂ O, DMSO, 20-25 °C, 25-55 min	[69]
4	Oxidation of α-diazo ketones	DMDO, acetone, rt	[70]
		(HMPA)MoO(O ₂), Hg(OAc) ₂ , DCE-MeOH, 0 °C, 15 min	[71]
5	Oxidation of aryl acetylenes	NBS, dry DMSO, rt, 20 h	[72]
		(PhSe) ₂ , (NH ₄) ₂ S ₂ O ₈ , water-CH ₃ CN, 60 °C, then chromatographed on SiO ₂ , DCM-ROH (99/1)	[73]
6	Reaction of methyl benzoates with KDMSO then oxidation	(1) DMSO, KOt-Bu, t-BuOH, rt, 4 h, then HCl, water, rt, 30 h	[74]
		(2) Cu(OAc) ₂ ·H ₂ O, CHCl ₃ , rt, 1 h	
7	Reaction of organolithium compounds with diethoxyacetyl piperidine	piperidine-1-yl-COCH(OEt) ₂ , p-Me ₂ NC ₆ H ₄ Li, Ether, reflux, 2h, then HCl, water, N ₂ (atm.), rt, 41 h	[56]
8	Chlorination of aryl methyl ketones	1,3-Cl ₂ -5,5-Me ₂ hydantoin, Cu(OTf) ₂ , CHCl ₃ , reflux, 5-8 h	[75]

1.2. Synthesis of Aryl glyoxals

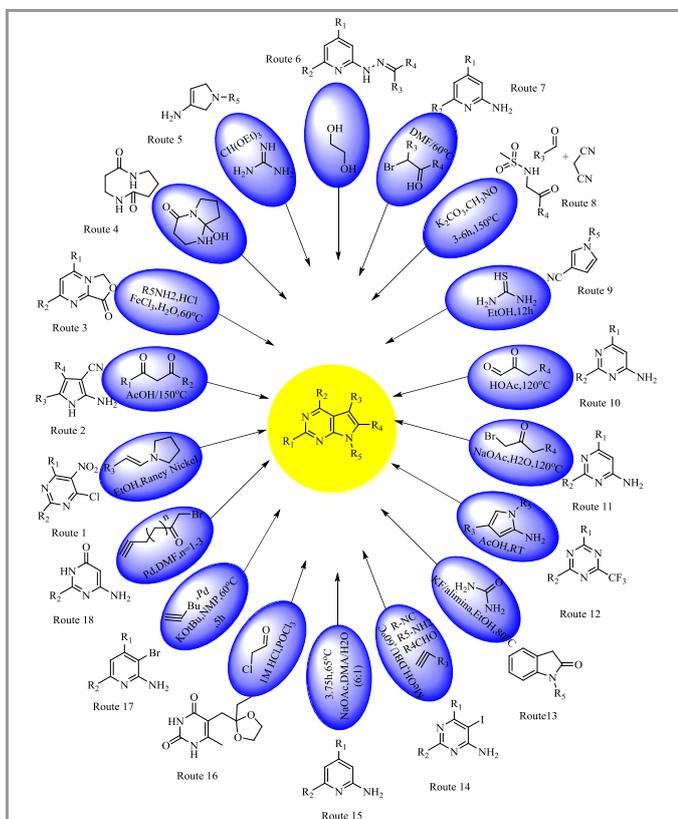
Various methods have been reported for synthesis of AGs in the literature. One of the most important

methods for their preparation is by oxidation of aryl methyl ketones by SeO₂. A compilation of methods to synthesize AGs, along with experimental procedures are summarized in Table 1.



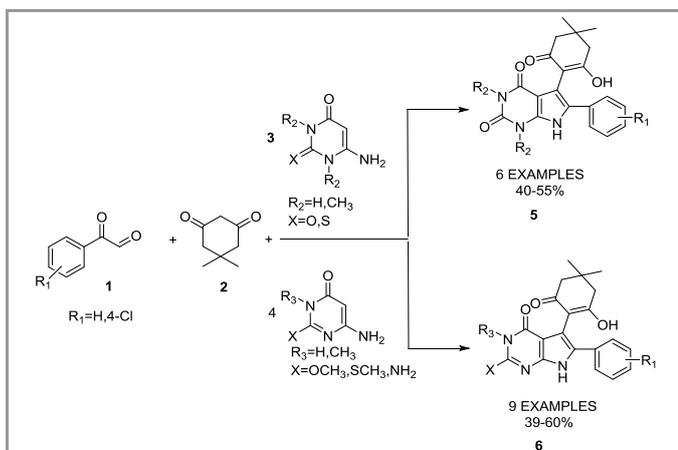
1.3. Pyrrolo[2,3-*d*]pyrimidine derivatives

Considerable effort has been made to synthesis a series of pyrrolo[2,3-*d*]pyrimidines and their derivatives, to optimise yield and purity. The major synthetic schemes are summarised in (Scheme 3) [55].



Scheme 3. Various synthetic schemes of Pyrrolo[2,3-*d*]pyrimidine derivatives

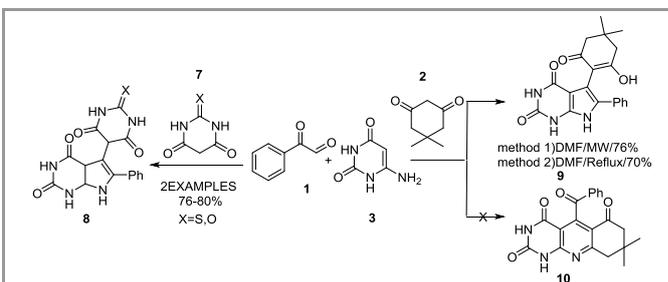
Quiroga and his coworkers reported the formation of several unexpected pyrrolo[2,3-*d*]pyrimidine derivatives **5** by a one-pot, three component reaction of aminopyrimidines **3** or **4** with dimedone **2**, and arylglyoxals **1** (Scheme 4) [76].



Scheme 4. Generation of pyrrolo[2,3-*d*]pyrimidines via multicomponent reaction of 6-aminopyrimidines, dimedone, and arylglyoxal

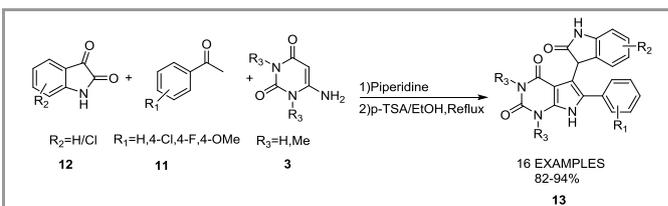
Shaker and his group designed the synthesis of new pyrrolo[2,3-*d*]pyrimidine-2,4-diones **8** or pyrimido[5,4-*b*]quinoline-2,4,9-(1*H*,3*H*,5*H*)-triones **9**

by reaction of 5-aminouracil **3**, dimedone **2** or barbituric acid **7** with phenylglyoxal hydrate **1** in DMF under controlled microwave heating for 20 min at 160 °C (Scheme 5) [77].



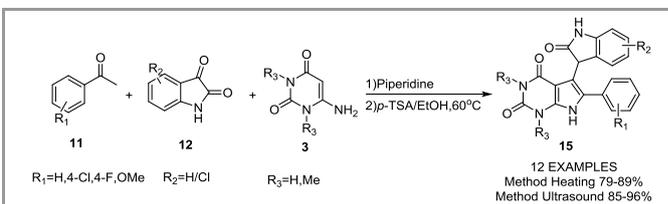
Scheme 5. One-pot Synthesis of 1*H*-Pyrrolo[2,3-*d*]pyrimidine-2,4-(3*H*,7*H*)-dione Derivatives Using Controlled Microwave Heating

Rad-Moghadam investigated the synthesis of the oxindolylpyrrolo[2,3-*d*]pyrimidines **13**, by a three-component reaction of the model substrates 6-amino-1,3-dimethyluracil **3** and acetophenone **11** instead of AG, the best yield of the product being obtained by the sequential use of piperidine (10 mol %) and *p*-toluenesulfonic acid (*p*-TSA, 40 mol %) in refluxing ethanol at 80 °C, which afforded the products **13** in good yields (Scheme 6) [78].



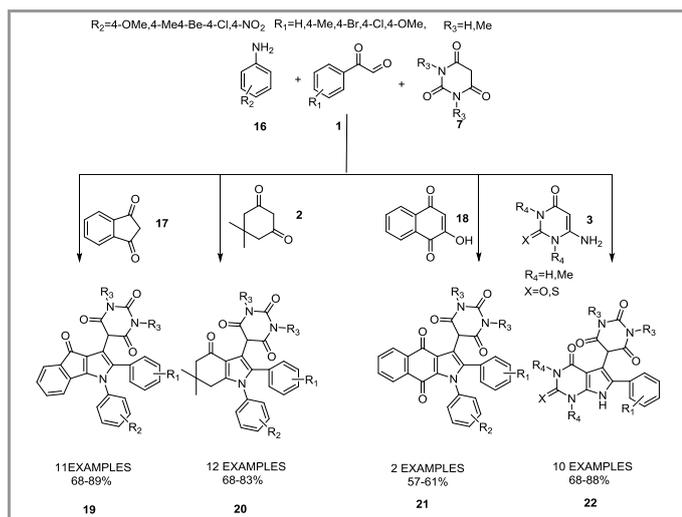
Scheme 6. Synthesis of oxindole substituted pyrrolo[2,3-*d*]pyrimidines under ultrasound irradiation

In another study, Azimi and his coworkers developed the synthesis of 1,3-dimethyl-5-(2-oxoindolin-3-yl)-6-phenyl-1*H*-pyrrolo[2,3-*d*]pyrimidine-2,4-(3*H*,7*H*)-dione analogues **15** under various reaction conditions and catalysts, by the reaction of acetophenone **11** and isatin **12** and 6-amino-1,3-dimethyluracil **3** as a model reaction (Scheme 7) [79].



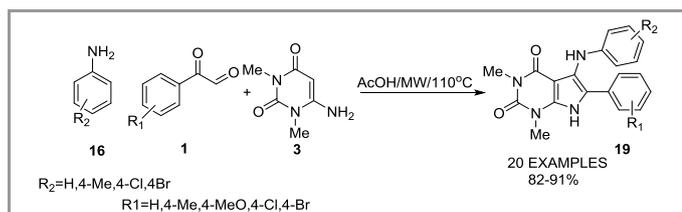
Scheme 7. Synthesis of bis-pyrrolo[2,3-*d*]pyrimidine derivatives with isatin

Dommaraju and coworkers developed a methodology based on a two-step sequence using 4-methoxy-aniline **16**, 1,3-dimethyl barbituric acid **7** and 4-methyl phenylglyoxal **1** with 1,3-indanedione **17** or dimedone **2** or 2-hydroxy-1,4-naphthaquinone **18** or 6-aminouracil **3** in equimolar quantities into a one-pot reaction in ethanol (Scheme 8) [80].



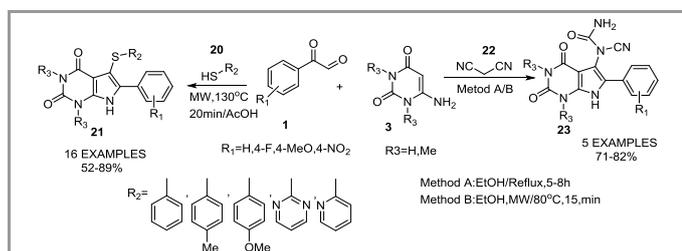
Scheme 8. Catalyst-free Multicomponent Reaction for the Synthesis of Pyrimidine Functionalized Pyrrolo-annulated Derivatives

Naidu and co-workers reported the use of microwave irradiation for the three component reaction of *N,N*-dimethyl-6-aminouracil **3**, phenylglyoxal **1**, and aniline **16** at 100 °C for 5 min, achieving an excellent yield of 84% when acetic acid was used as a solvent, without any added catalyst (Scheme 9) [81].



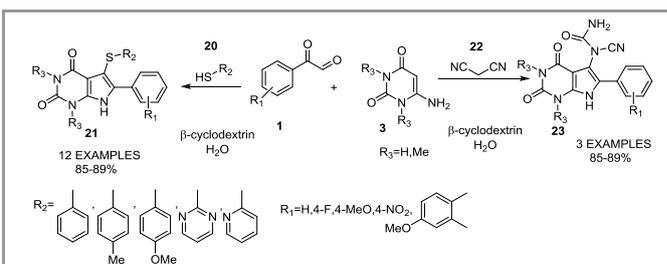
Scheme 9. Synthesis of bis-pyrrolo[2,3-d]pyrimidine derivatives under MW irradiations

Choudhury's group synthesized pyrrolo[2,3-*d*]pyrimidine derivatives **21** and **23** by the reaction of phenylglyoxal **1**, 6-amino-1,3-dimethyluracil **3** and 2-mercaptopyrimidine **20** or malononitrile **22** under microwave heating conditions (Scheme 10) [82].



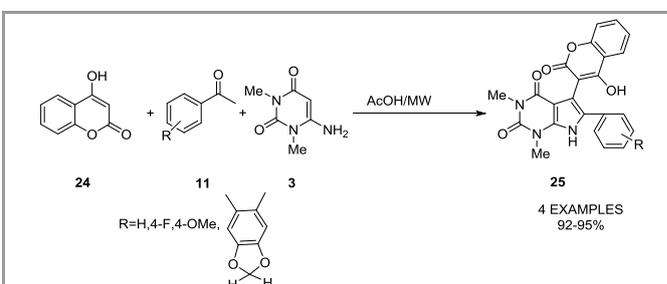
Scheme 10. Synthesis of Novel 5,6-Disubstituted Pyrrolo [2,3-d]Pyrimidine-2,4-Diones via One-Pot

Yadav and co-workers reported a one-pot, four component biomimetic protocol for the synthesis of pyrrolo[2,3-*d*]pyrimidine **23** for the first time, by employing 6-aminouracil **3**, malononitrile **22** and arylglyoxal monohydrates **1** in aqueous β -cyclodextrin (Scheme 11) [83].



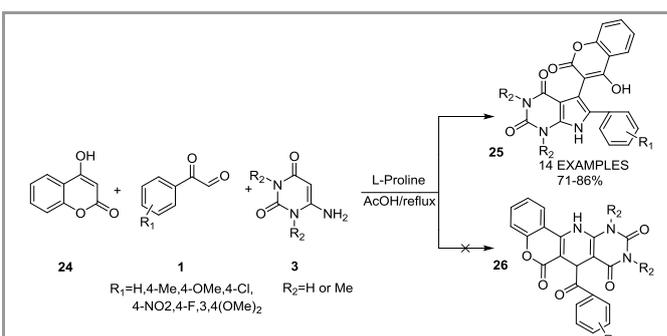
Scheme 11. Synthesis of Pyrrolo [2,3-d]Pyrimidine-2,4-Diones via One-Pot Three-Component Reactions

A simple and efficient method for the synthesis of a series pyrrolo[2,3-*d*]pyrimidine derivatives **25** with excellent yields was reported by Choudhury and co-workers. The reaction was performed between acetophenone **11** instead of AG **1**, 4-hydroxycoumarin **24**, and 6-aminouracil **3** under MW conditions in AcOH for the synthesis of to pyrrolo[2,3-*d*]pyrimidine derivatives **25** (Scheme 12) [84].



Scheme 12. Synthesis of bis-pyrrolo[2,3-d]pyrimidine derivatives under MW conditions

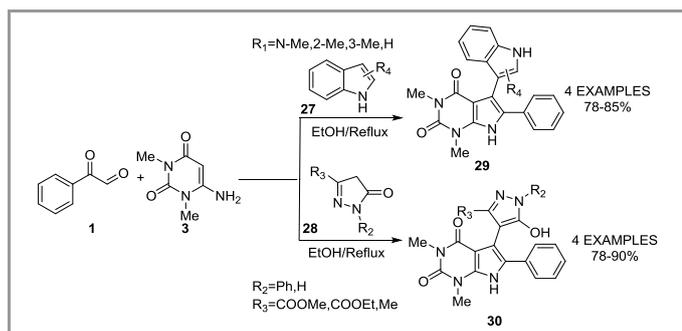
Javahershenas and Khalafy reported a new method for the synthesis of pyrrolo[2,3-*d*]pyrimidine derivatives **25** through the one-pot, three-component reaction of 4-hydroxycoumarin **24**, arylglyoxal **1** and 6-aminouracil **3** or 1,3- dimethyl-6-aminouracil **3** catalyzed by *L*-proline as catalyst (Scheme 13) [85].



Scheme 13. Synthesis of pyrrolo[2,3-d]pyrimidine derivatives catalyzed by *L*-proline

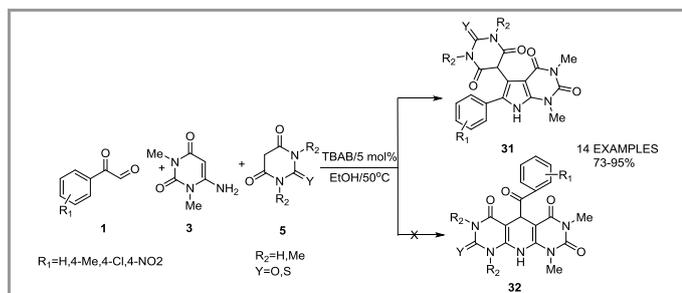
Khurana and coworkers reported the microwave assisted synthesis of some novel 5-substituted 6-phenyl pyrrolo[2,3-*d*]pyrimidine derivatives **29** and **30** by one-pot three-component condensation of arylglyoxal **1**, 6-amino-1,3dimethyluracil **3** and indole derivatives **27** or 1*H*-pyrazol-5(4*H*)-one derivatives **28** by refluxing in ethanol under catalyst-free conditions (Scheme 14) [86].





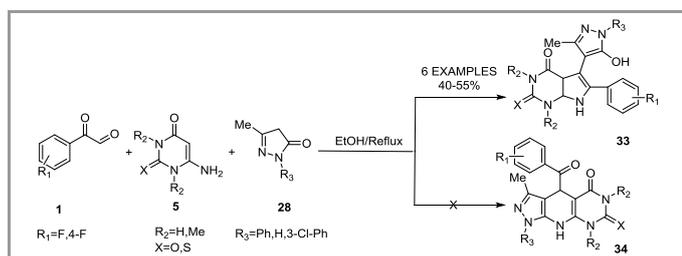
Scheme 14. Synthesis of [2,3-*d*]pyrimidine Derivatives via One-pot Three-component Reactions Under Catalyst-Free Condition

In another study, Javahershenas and Khalafy reported an efficient procedure for the reaction of arylglyoxals **1** with 6-amino-1,3-dimethyluracil **3** and barbituric acid derivatives **5** in the presence of TBAB (5 mol%) in ethanol at 50 °C, affording polyfunctionalized pyrrolo[2,3-*d*]pyrimidine derivatives **31** in high yields with no sign of any dihydropyrido[2,3-*d*:6,5-*d'*]dipyrimidine derivatives **32** (Scheme 15) [87].



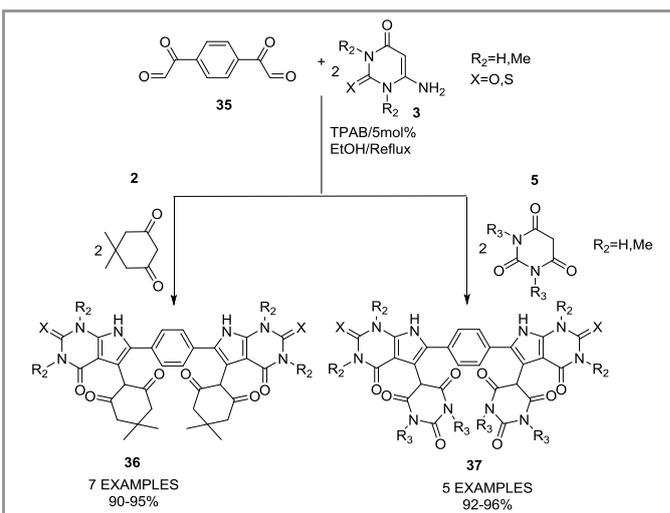
Scheme 15. TBAB catalyzed synthesis of bis-pyrrolo[2,3-*d*]pyrimidine derivatives

The new indole derivatives pyrrolo[2,3-*d*]pyrimidine **33** have been synthesized from arylglyoxals **1** with 6-amino-uracil derivatives **5** and 5-methyl-2,4-dihydro-3*H*-pyrazol-3-one derivatives **28** in the presence of sulfamic acid as an efficient catalyst by Bayat and coworkers (Scheme 16) [88].



Scheme 16. Generation of pyrrolo[2,3-*d*]pyrimidines via multicomponent reaction of 6-aminopyrimidines, dimedone, and arylglyoxal

Khalafy and coworkers used TPAB as catalyst in the reaction between 1,4-phenylene-bis-glyoxal **35**, 6-aminouracil derivatives **3**, and barbituric acid derivatives **5** or dimedone **2** in a one-pot, three-component reaction in EtOH under reflux conditions for the synthesis of bis-pyrrolo[2,3-*d*]pyrimidine derivatives **36** and **37** in high yields (Scheme 17) [89].



Scheme 17. Synthesis of pyrazolo substituted pyrrolo[2,3-*d*]pyrimidines

2. Conclusion

This study presented an overview of the recent literature on application of the arylglyoxals for synthesis of the pyrrolo[2,3-*d*]pyrimidines via multicomponent reactions over the last decades. In the light of our studies, we found that, arylglyoxals have been frequently utilized for synthesis of various organic compounds including, pyrrolo[2,3-*d*]pyrimidine derivatives, which are influential because of their biological and medicinal characteristics.

Acknowledgment

We are grateful for financial support from the Research Council of Urmia University.

Abbreviations

AcOH	acetic acid
AGs	arylglyoxals
AcO	acetate
aq	aqueous
DMF	<i>N,N</i> -dimethylformamide
DMDO	dimethyldioxirane
DCE	1,2-dichloroethane
DBU	1,8-diazabicyclo[5.4.0]undec-7-ene
DMA	dimethylacetamide
DCM	dichloromethane
DMSO	dimethylsulfoxide
HMPA	hexamethylphosphoramide
HIV	human immunodeficiency virus
NMP	<i>N</i> -methyl-2-pyrrolidone
NBS	<i>N</i> -bromosuccinimide
MW	microwave
MCR	multicomponent reaction
PG	phenylglyoxal



<i>p</i> -TsOH	<i>p</i> -toluenesulfonic acid
rt	room temperature
TBAB	tetrabutylammonium bromide
TPAB	tetrapropylammonium bromide

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