

## DEVELOPMENT OF NEW PYRAZOLE HYBRIDS AS ANTITUBERCULAR AGENTS: SYNTHESIS, BIOLOGICAL EVALUATION AND MOLECULAR DOCKING STUDY

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### ABSTRACT

**Objective:** Synthesis of new 1, 3-diphenyl pyrazole derivatives 9(a-f) and 10(a-f) using molecular hybridization approach and evaluation of their antitubercular and cytotoxic studies.

**Methods:** The structures of synthesized compounds were confirmed by <sup>1</sup>H NMR, <sup>13</sup>C NMR and mass spectra. The antitubercular activity of compounds and standard drugs were assessed against *Mycobacterium tuberculosis* using Microplate alamar blue assay (MABA). The cytotoxic activities were performed by Sulforhodamine B (SRB) assay. The molecular docking and *in silico* ADME prediction studies were also performed by using Schrodinger.

**Results:** The results reveal that compounds 9c, 9d, 10c and 10d exhibited substantial antitubercular potential with MIC < 20 μM. The cytotoxic studies revealed that active compounds (9c, 9d, 10c and 10d) are non-toxic to HeLa cancer cell lines with the selectivity index > 10. The molecular docking study was performed to study the binding orientation and affinity of synthesized compounds for *InhA* enzyme.

**Conclusion:** The study explored that 1, 3-diphenyl pyrazole hybrid coupled with well-known antitubercular drugs could be a potential lead for antitubercular agents. *In silico* molecular docking, study helps to identify their corresponding intermolecular ligand-protein interactions with target enzyme. Also, ADME prediction studies revealed that the compounds were in acceptable range to have good pharmacokinetic parameters.

**Keywords:** Pyrazole, Antitubercular activity, Cytotoxicity, Molecular docking

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### INTRODUCTION

Tuberculosis (TB) is a re-emerging global health threat caused by an infectious bacillus called *Mycobacterium tuberculosis* and is the second largest killer disease caused by a single infectious agent after the human immunodeficiency virus (HIV) [1]. World Health Organization (WHO) Global Tuberculosis Report 2016 suggested that, in 2015, there were an estimated 10.4 million new tuberculosis cases worldwide. People living with HIV accounted for 1.2 million (11 %) of all new tuberculosis cases [2].

World Health Organization (WHO) promoted a comprehensive tuberculosis management program known as DOTS (Directly Observed Treatment, Shortcourse). DOTS therapy involves the use of four first-line agents as isoniazid, ethambutol, pyrazinamide, and rifampicin for six months [3].

Despite notable progress in antitubercular agents, multi-drug resistance tuberculosis (MDR-TB), extremely drug resistance tuberculosis (XDR-TB) and HIV co-infection are the major hurdles in control of tuberculosis infection [4]. Therefore, the discovery and development of new chemical entities with a novel mechanism of action, safe and efficacious drugs, and shorter duration of treatment are the desperate needs for infectious diseases research programs. Pioneering scientists had reported pyrazole and its derivatives for an extensive range of pharmacological activities such as antipyretic, analgesic, antimicrobial, anticancer, antitubercular, antiviral, antihypertensive, antioxidant, antidepressant, and anxiolytic [5-7]. Rangappa S. Keri *et al.* reviewed and discussed the possible structure-activity relationship of different types of pyrazole scaffold for designing of better antitubercular agents [8]. Synthetic analogues of pyrazole are known to exhibit significant antitubercular activity especially 1, 3-diphenyl pyrazole motif is known to be a potent antitubercular agent (fig. 1) [9]. Even though isoniazid (INH) has been the most widely used treatment

for tuberculosis and its latent infections, it suffers from two pitfalls as enzymatic acetylation of isoniazid by N-acetyltransferase (NATs) and its associated liver toxicity. So, blocking N-acetylation *via* chemical modification of N-terminal of isoniazid makes it to be more effective and less hepatotoxic than isoniazid [10]. Linezolid is recommended by the WHO to treat drug-resistant tuberculosis [11]. The study also found that it is potentially important for patients with XDR-TB. It acts by inhibiting protein synthesis at an early stage of translation [12]. 3-Fluoro-4-morpholinoaniline, a bioactive segment of linezolid plays a very crucial role in the antitubercular activity. 3-Fluoro group attached to phenyl ring enhances the potency while morpholine ring improves pharmacokinetic and water solubility of linezolid [13].

The molecular hybridization approach is one of the most valuable structural modification tools useful for the discovery of ligands and prototypes. Recently, the emerging interest to discover hybrid molecules resulting from the combination of pharmacophoric moieties of different known lead compounds has been developed with a new hope for the treatment of multifactorial diseases [14].

In the light of above consideration, it seems rational to combine the pharmacophoric potential of the two well-known antitubercular agents like isoniazid and linezolid with one core scaffolds, pyrazole (fig. 1). Finally, these hybridized molecules were evaluated for antitubercular and cytotoxicity studies. The molecular docking study was performed to determine the possible mechanism of action of titled compounds by which they exert antitubercular activity. The *in silico* molecular docking study helps to identify their corresponding intermolecular ligand-protein interactions with target enzyme to set a basis for mycobacterial inhibition. In addition to this, absorption, distribution, metabolism, and excretion (ADME) prediction studies were also performed to explore the results.

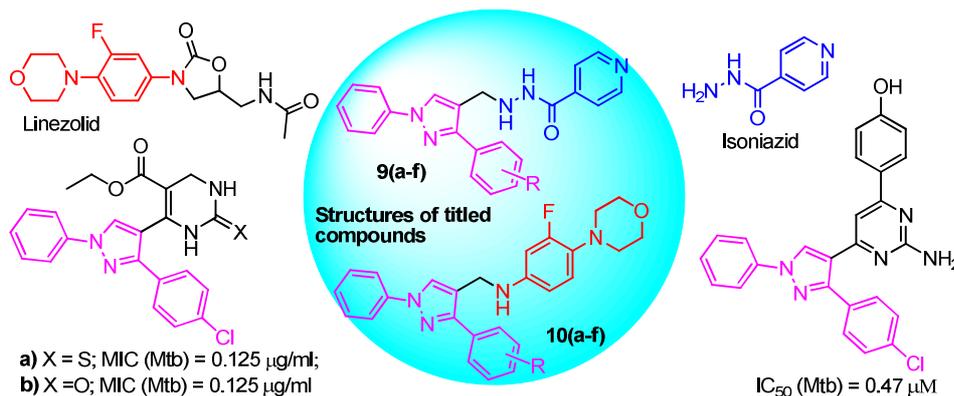


Fig. 1: Pharmacophores derived from structurally related and well known antitubercular agents

## MATERIALS AND METHODS

### Materials

All the solvents and reagents were used as obtained from the supplier or recrystallized/redistilled unless otherwise stated. The progress of the reaction was monitored by thin layer chromatography (TLC) using silica gel-G pre-coated aluminium plates (Merck) and visualized under UV light. Melting points were measured in open capillary tubes and uncorrected. The synthesized compounds were characterized by spectral analysis like  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR and Mass spectroscopy. The nuclear magnetic resonance  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra were recorded with a Bruker DRX-400 MHz and Bruker DRX-100 MHz NMR spectrometer. Chemical shifts ( $\delta$ ) are reported in parts per million (ppm). The mass spectra were recorded under ESI mode on Waters Micromass equipment (model Q-TOF micro).

### General procedure for synthesis of *N'*-((3-(4-substituted phenyl)-1-phenyl-1*H*-pyrazol-4-yl) methyl) isonicotinohydrazide 7(a-f) and *N'*-((3-(4-substituted phenyl)-1-phenyl-1*H*-pyrazol-4-yl) methylene)-3-fluoro-4-morpholinoaniline 8(a-f)

A mixture of substituted 3-aryl-1-phenylpyrazol-4-carbaldehydes 4(a-f) (1.0 mmol) and isoniazid 5 (1.0 mmol) or 3-fluoro-4-morpholinoaniline 6 (1.0 mmol) was dissolved in absolute ethanol (20 ml) and heated under reflux for 6–8 h in the presence of glacial acetic acid (2.0 mmol) as a catalyst. Then, the reaction mixture was poured in ice-cold water and filtered under suction; the precipitate thus obtained was washed with water and recrystallized from ethanol.

### General procedure for synthesis of *N'*-((3-(4-substituted phenyl)-1-phenyl-1*H*-pyrazol-4-yl) methyl) isonicotinohydrazide 9(a-f) and *N'*-((3-(4-substituted phenyl)-1-phenyl-1*H*-pyrazol-4-yl) methyl)-3-fluoro-4-morpholinoaniline 10(a-f)

Corresponding imines 7(a-f) and 8(a-f) (0.01 mmol) were dissolved in absolute methanol (10–15 ml) and placed in a two-necked flask fitted with two glass stoppers and a reflux condenser. This solution was warmed with continuous stirring. To this was added an equimolar amount of solid sodium borohydride ( $\text{NaBH}_4$ ). The portion-wise addition was made through one of the necks of the flask. If the reaction became too vigorous, the flask was moved to room temperature. When the initial reaction had subsided, the contents of the flask were refluxed for 15 min and then cooled. The precipitate of secondary amine thus formed was collected, washed with water, and dried. Then the compounds were recrystallized using methanol.

### *N'*-((1,3-diphenyl-1*H*-pyrazol-4-yl)methyl)isonicotinohydrazide (9a)

Pale yellow solid; Yield=86%; m. p. =152–154  $^\circ\text{C}$ ;  $R_f$  value=0.56;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 2.98 (s, 1H,  $-\text{CH}_2-\text{NH}-$ ), 3.92 (s, 2H,  $-\text{CH}_2-$ ), 7.52 (s, 1H,  $-\text{CH}$ -of Pyrazole), 7.47–7.83 (m, 10H, Ar-H), 7.72 (s, 1H,  $-\text{NH}-\text{C}=\text{O}$ ), 7.90–8.79 (m, 4H, pyridine-H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 49.63, 120.17, 121.05, 125.27, 126.55, 128.53, 128.97,

129.48, 130.93, 131.04, 131.22, 137.87, 137.87, 140.31, 150.63, 154.73, 164.03;  $m/z=370.25$  (M+1) (% Mol. Wt.: 99.70).

### *N'*-((3-(4-chlorophenyl)-1-phenyl-1*H*-pyrazol-4-yl)methyl) isonicotinohydrazide (9b)

Yellow solid; Yield=92%; m. p =185–187  $^\circ\text{C}$ ;  $R_f$  value=0.60;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 3.14 (s, 1H,  $-\text{CH}_2-\text{NH}-$ ), 3.90 (s, 2H,  $-\text{CH}_2-$ ), 7.43–7.83 (m, 9H, Ar-H), 7.56 (s, 1H,  $-\text{CH}$ -of pyrazole), 7.73 (s, 1H,  $-\text{NH}-\text{C}=\text{O}$ ), 7.89–8.79 (m, 4H, pyridine-H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 49.63, 120.17, 121.05, 125.27, 126.55, 129.15, 129.48, 130.89, 131.02, 131.22, 134.00, 137.87, 140.31, 150.63, 154.73, 164.03;  $m/z=404.60$  (M+1) (% Mol. Wt.: 99.63).

### *N'*-((3-(4-fluorophenyl)-1-phenyl-1*H*-pyrazol-4-yl)methyl) isonicotinohydrazide (9c)

Dark-yellow solid; Yield=90%; m. p =208–210  $^\circ\text{C}$ ;  $R_f$  value=0.54;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 2.87 (s, 1H,  $-\text{CH}_2-\text{NH}-$ ), 3.92 (s, 2H,  $-\text{CH}_2-$ ), 7.38–7.84 (m, 9H, Ar-H), 7.51 (s, 1H,  $-\text{CH}$ -of pyrazole), 7.57 (s, 1H,  $-\text{NH}-\text{C}=\text{O}$ ), 7.90–8.79 (m, 4H, pyridine-H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 49.63, 116.69, 116.89, 120.17, 121.07, 125.27, 126.55, 128.41, 128.44, 129.48, 131.22, 132.32, 132.40, 137.87, 140.31, 150.63, 154.73, 162.94, 164.03, 165.46;  $m/z=388.53$  (M+1) (% Mol. Wt.: 99.64).

### *N'*-((3-(4-bromophenyl)-1-phenyl-1*H*-pyrazol-4-yl)methyl) isonicotinohydrazide (9d)

Brown solid; Yield=91%; m. p =292–294  $^\circ\text{C}$ ;  $R_f$  value=0.71;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 3.86 (s, 1H,  $-\text{CH}_2-\text{NH}-$ ), 3.91 (s, 2H,  $-\text{CH}_2-$ ), 7.44–7.64 (m, 9H, Ar-H), 7.56 (s, 1H,  $-\text{CH}$ -of pyrazole), 7.81–8.79 (m, 4H, pyridine-H), 8.32 (s, 1H,  $-\text{NH}-\text{C}=\text{O}$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 49.63, 120.17, 121.05, 123.85, 125.27, 126.55, 128.50, 129.48, 129.95, 130.84, 137.87, 140.31, 150.63, 154.73, 164.03;  $m/z=448.64$  (M+1) (% Mol. Wt.: 99.52).

### *N'*-((3-(4-nitrophenyl)-1-phenyl-1*H*-pyrazol-4-yl)methyl) isonicotinohydrazide (9e)

Dark grey solid; Yield=82%; m. p =276–278  $^\circ\text{C}$ ;  $R_f$  value=0.55;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 3.90 (s, 1H,  $-\text{CH}_2-\text{NH}-$ ), 4.12 (s, 2H,  $-\text{CH}_2-$ ), 7.44–7.50 (m, 9H, Ar-H), 7.51 (s, 1H,  $-\text{CH}$ -of pyrazole), 7.81–8.79 (m, 4H, pyridine-H), 8.50 (s, 1H,  $-\text{NH}-\text{C}=\text{O}$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 49.63, 120.17, 121.05, 123.63, 125.27, 126.55, 127.40, 129.48, 131.22, 137.87, 140.31, 148.87, 150.63, 154.73, 164.03;  $m/z=415.34$  (M+1) (% Mol. Wt.: 99.73).

### *N'*-((3-(4-methoxyphenyl)-1-phenyl-1*H*-pyrazol-4-yl)methyl) isonicotinohydrazide (9f)

Yellow solid; Yield=85%; m. p =162–164  $^\circ\text{C}$ ;  $R_f$  value=0.63;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 3.81 (s, 3H,  $-\text{OCH}_3$ -of phenyl), 3.90 (s, 2H,  $-\text{CH}_2-$ ), 7.02–7.49 (m, 9H, Ar-H), 7.50 (s, 1H,  $-\text{CH}$ -of pyrazole), 7.76–8.79 (m, 4H, pyridine-H), 7.77 (s, 1H,  $-\text{NH}-\text{C}=\text{O}$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 49.63, 55.39, 114.70, 120.17, 121.05, 125.27, 126.55, 129.48, 130.82, 131.22, 137.87, 140.31, 150.63, 154.73, 161.23, 164.03;  $m/z=400.4$  (M+1) (% Mol. Wt.: 99.45).

**N-((1,3-diphenyl-1H-pyrazol-4-yl)methyl)-3-fluoro-4-morpholinoaniline (10a)**

Brown solid; Yield=94%; m. p =226-228 °C;  $R_f$  value=0.64;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 2.93-2.96 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 3.72-3.75 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 4.64 (s, 2H, -CH<sub>2</sub>-), 6.32 (s, 1H, NH), 6.15-7.92 (m, 13H, Ar-H); 7.49 (s, 1H, -CH-of pyrazole),  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 40.66, 51.10, 66.81, 102.60, 102.80, 113.73, 113.76, 120.17, 121.19, 121.27, 125.01, 126.55, 128.53, 128.97, 129.48, 130.93, 131.04, 131.17, 134.12, 134.32, 140.31, 144.01, 144.09, 154.73, 155.09, 157.61; MS:  $m/z=429.8$  (M+1) (% Mol. Wt.: 99.24).

**N-((3-(4-chlorophenyl)-1-phenyl-1H-pyrazol-4-yl)methyl)-3-fluoro-4-morpholinoaniline (10b)**

Yellowish-brown solid; Yield=94%; m. p =228-230 °C;  $R_f$  value=0.53;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 2.93-2.96 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 3.72-3.75 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 4.56 (s, 2H, -CH<sub>2</sub>-), 5.89 (s, 1H, NH), 5.88-7.92 (m, 12H, Ar-H); 7.46 (s, 1H, -CH-of pyrazole);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 40.66, 51.14, 66.81, 102.60, 102.80, 113.73, 113.76, 120.17, 121.19, 121.27, 125.01, 126.55, 129.15, 129.48, 130.89, 131.02, 131.17, 134.00, 134.12, 134.32, 140.31, 144.01, 144.09, 154.73, 155.09, 157.61; MS:  $m/z=463.3$  (M+1) (% Mol. Wt.: 99.84).

**3-Fluoro-N-((3-(4-fluorophenyl)-1-phenyl-1H-pyrazol-4-yl)methyl)-4-morpholinoaniline (10c)**

Brown solid; Yield=89%; m. p =200-202 °C;  $R_f$  value=0.62;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 2.93-2.96 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 3.72-3.75 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 4.60 (s, 2H, -CH<sub>2</sub>-), 6.24 (s, 1H, NH), 6.12-7.92 (m, 12H, Ar-H); 7.46 (s, 1H, -CH-of pyrazole);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 40.66, 51.14, 66.81, 102.60, 102.80, 113.73, 113.76, 116.69, 116.89, 120.17, 121.19, 125.01, 126.55, 128.41, 128.44, 129.48, 131.17, 132.32, 132.40, 134.12, 140.31, 144.01, 144.09, 154.73, 155.09, 157.61; MS:  $m/z=447.3$  (M+1) (% Mol. Wt.: 99.43).

**N-((3-(4-bromophenyl)-1-phenyl-1H-pyrazol-4-yl)methyl)-3-fluoro-4-morpholinoaniline (10d)**

Dark-brown solid; Yield=85%; m. p. =188-190 °C;  $R_f$  value=0.43;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 2.93-2.96 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 3.72-3.75 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 4.59 (s, 2H, -CH<sub>2</sub>-), 7.47 (s, 1H, NH), 7.45 (s, 1H, -CH-of pyrazole), 7.46-7.92 (m, 12H, Ar-H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 40.66, 51.10, 66.81, 102.60, 102.80, 113.73, 113.76, 120.17, 121.19, 121.27, 123.85, 125.01, 126.55, 128.50, 129.48, 129.95, 130.84, 131.17, 134.12, 134.32, 140.31, 144.09, 154.73, 155.09, 157.61; MS:  $m/z=507.7$  (M+1) (% Mol. Wt.: 99.40).

**3-Fluoro-4-morpholino-N-((3-(4-nitrophenyl)-1-phenyl-1H-pyrazol-4-yl)methyl)aniline (10e)**

Dark-brown solid; Yield=85%; m. p =188-190 °C;  $R_f$  value=0.56;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 2.93-2.96 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 3.72-3.75 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 4.59 (s, 2H, -CH<sub>2</sub>-), 6.15 (s, 1H, NH), 7.52 (s, 1H, -CH-of pyrazole), 5.98-8.24 (m, 12H, Ar-H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 40.66, 51.10, 66.81, 102.60, 102.80, 113.73, 113.76, 120.17, 121.19, 121.27, 123.63, 125.01, 126.55, 127.40, 129.48, 131.17, 134.12, 134.32, 137.87, 140.31, 144.09, 148.87, 154.73, 155.09, 157.61; MS:  $m/z=474.8$  (M+1) (% Mol. Wt.: 99.65).

**3-Fluoro-N-((3-(4-methoxyphenyl)-1-phenyl-1H-pyrazol-4-yl)methyl)-4-morpholinoaniline (10f)**

Pale yellow solid; Yield=90%; m. p =238-240 °C;  $R_f$  value=0.70;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz,  $\delta$  ppm): 2.93-2.96 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 3.72-3.75 (t, 4H, -CH<sub>2</sub>-of morpholine,  $J=7.2$  Hz), 3.81 (s, 3H, -OCH<sub>3</sub>-of Phenyl), 4.65 (s, 2H, -CH<sub>2</sub>-), 6.40 (s, 1H, NH), 7.47 (s, 1H, -CH-of pyrazole), 6.13-7.92 (m, 12H, Ar-H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$  ppm): 40.66, 51.14, 55.39, 66.81, 102.60, 102.80, 113.73, 113.76, 114.70, 120.17, 121.19, 121.27, 125.01, 125.75, 126.55, 129.48, 130.82, 131.17, 134.12, 134.32, 140.31, 144.09, 154.73, 155.09, 157.61, 161.23; MS:  $m/z=459.3$  (M+1) (% Mol. Wt.: 99.41).

**Biological activities****Antitubercular activity**

The antitubercular activity of the synthesized compounds and standard drugs were assessed against *Mycobacterium tuberculosis* using Microplate alamar blue assay (MABA) [15]. This methodology is non-toxic, uses a thermally stable reagent and shows good correlation with proportional and BACTEC radiometric method. Briefly, 200  $\mu\text{l}$  of sterile deionized water was added to all outer perimeter wells of sterile 96 wells plate to minimized evaporation of medium in the test wells during incubation. The 96 wells plate received 100  $\mu\text{l}$  of the Middlebrook 7H9 broth (HiMedia, Mumbai) and serial dilution of compounds was made directly on the plate. The final drug concentrations tested were made. Plates were covered and sealed with parafilm and incubated at 37 °C for five days. After this time, 25  $\mu\text{l}$  of freshly prepared 1:1 mixture of almar blue reagent and 10 % tween 80 was added to the plate and incubated for 24 h. The minimum inhibitory concentration (MIC) was defined as the minimum concentration of compound required to give complete inhibition of bacterial growth. The antitubercular screening was performed in triplet and the standard errors were all within 10 % of the mean.

**Cytotoxicity study**

The active compounds 9c, 9d, 10c and 10d were evaluated for their *in vitro* cytotoxic activity against HeLa human cancer cell line, by Sulforhodamine B (SRB) assay according to the reported procedures [16]. This assay gives Growth inhibition concentration (GI<sub>50</sub>) values which were taken as the lowest concentration of the compound killing 50 % of the cells.

**Selectivity index**

The selectivity index (SI) was calculated by dividing GI<sub>50</sub> for human cancer cell line (HeLa) by the MIC ( $\mu\text{g/ml}$ ) for *in vitro* activity against *Mycobacterium tuberculosis*. If the selectivity index (SI) is  $\geq 10$ , the compounds are processed further for drug development [17].

**Computational studies****Molecular docking study**

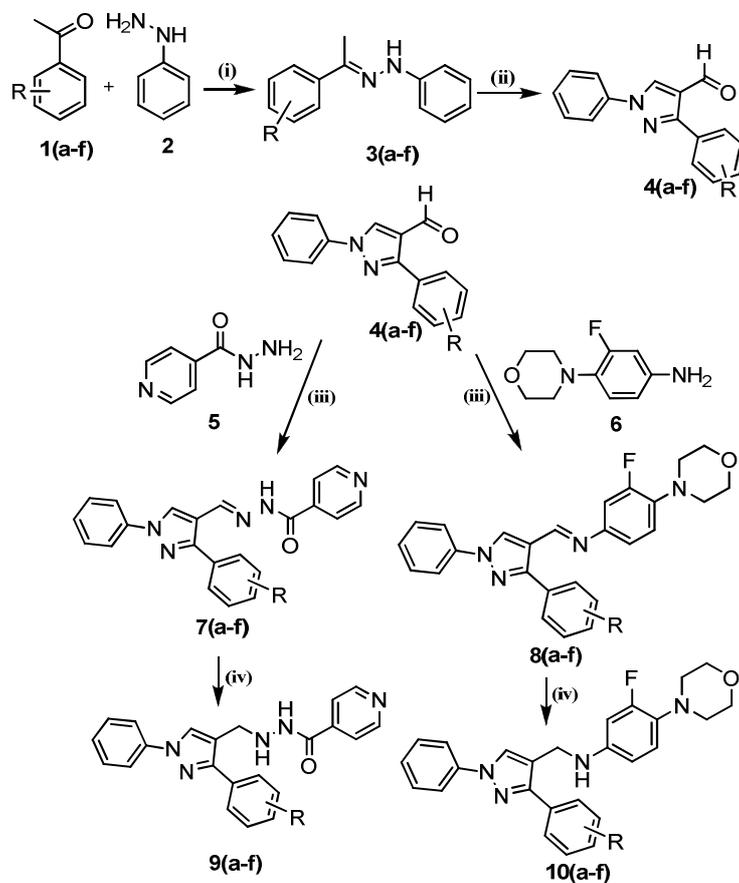
The molecular docking studies were performed in Maestro 9.1 using Glide (Schrodinger, LLC, New York, NY, 2015) [18]. This is an interactive molecular graphics program for docking calculations, for identification of the probable binding site of the biomolecules, and for visualizing ligand-receptor interactions. All compounds were built using Maestro build panel and optimized to lower energy conformers using Ligprep v2.4 which uses an OPLS\_2005 force field. Epik v2.1 was used to generate an ionized state of all compounds at target pH 7.0 $\pm$ 2.0. The coordinate for *InhA* enzyme (PDB: 2X23) [19] were taken from RCSB Protein Data Bank and prepared for docking using 'protein preparation wizard' in Maestro 9.1. Water molecules in the structures were removed and termini were capped by adding ACE and NMA residue. The bond orders and formal charges were added for hetero groups and hydrogens were added to all atoms in the structure. Side chains that are not close to the binding cavity and do not participate in salt bridges were neutralized. After preparation, the structure was refined to optimize the hydrogen bond network using an OPLS\_2005 force field. This helps in the reorientation of side chain hydroxyl group. The minimization was terminated when the energy converged or the RMSD reached a maximum cutoff of 0.30 Å. Grids were then defined around refined structure by centring on ligand using default box size. The extra precision (XP) docking mode for all compounds was performed on a generated grid of protein structure. The final evaluation of ligand-protein binding was done with glide score (docking score). The extra precision (XP) docking mode for all compounds was performed on a generated grid of protein structure. The final evaluation of ligand-protein binding was done with glide score (docking score).

**RESULTS AND DISCUSSION****Chemistry**

The pyrazolyl derivatives 9(a-f) and 10(a-f) were synthesized as presented in Scheme 1. Initially, 3-aryl-1-phenylpyrazole-4-

carbaldehydes 4(a-f) were synthesized from phenylhydrazones 3(a-f) by Vilsmeier-Haack formylation as reported by Raquib Alam *et al.* [20]. Then 3-aryl-1-phenylpyrazole-4-carbaldehydes 4(a-f) were condensed with isoniazid 5 and 3-fluoro-4-morpholinoaniline 6

which gives corresponding imines 7(a-f) and 8(a-f) respectively. The last step involves reduction amination of imines 7(a-f) and 8(a-f) using sodium borohydride ( $\text{NaBH}_4$ ) in methanol to afford corresponding titled compounds 9(a-f) and 10(a-f).



a; R = 4-H, b; R = 4-Cl, c; R = 4-F, d; R = 4-Br, e; R = 4-NO<sub>2</sub>, f; R<sub>1</sub> = 4-OCH<sub>3</sub>

**Scheme 1:** General procedure for synthesis of pyrazolyl derivatives 9(a-f) and 10(a-f); (i) EtOH, H<sub>2</sub>SO<sub>4</sub>, reflux; (ii) a. POCl<sub>3</sub>/DMF, 80 °C; b. NaHCO<sub>3</sub>/H<sub>2</sub>O; (iii) EtOH, AcOH, reflux; (iv) Methanol, NaBH<sub>4</sub>

## Biological activity

### *In vitro* antitubercular activity

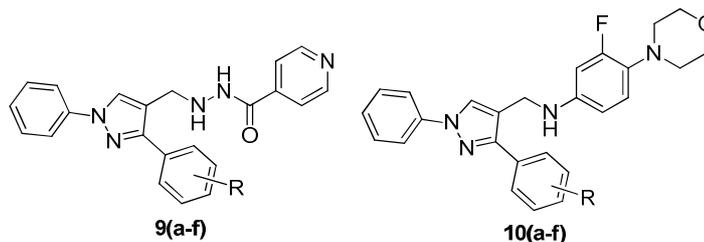
All pyrazole-isoniazid 9(a-f) and pyrazole-linezolid like conjugates 10(a-f) were tested for their *in vitro* antitubercular activity against *Mycobacterium tuberculosis* H37Ra (ATCC 27294). The antitubercular activity in terms of minimum inhibitory concentration (MIC) of titled compounds and standard drugs were given in table 1. As evident from the data, among the pyrazole-isoniazid 9(a-f) conjugates two compounds 9c and 9d were found to be most active against *Mycobacterium tuberculosis* H37Ra having MIC 16.06  $\mu\text{M}$  and 13.91  $\mu\text{M}$ , respectively. The compound 9f showed good antitubercular activity with MIC 31.15  $\mu\text{M}$ . Other compounds 9b (MIC = 61.70  $\mu\text{M}$ ) and 9g (MIC = 60.07  $\mu\text{M}$ ) also showed moderate antitubercular activity. Further, pyrazole-linezolid like conjugates 10(a-f) was found to possess excellent potency against *Mycobacterium tuberculosis* H37Ra. From this series, compound 10c and 10d exhibited excellent antitubercular effect with MIC 13.94  $\mu\text{M}$  and 12.30  $\mu\text{M}$ , respectively. However, rest of compounds 10b, 10e, 10a, and 10f were also found to have moderate inhibitory effect against *Mycobacterium tuberculosis*. As a result of this discussion, compound 10d (MIC = 12.30  $\mu\text{M}$ ) was found to be most potent in comparison of both the series.

### Structure-activity relationship (SAR)

Observing the results, we deduce the valuable data about structure-activity correlations of the tested compounds. From the two series of

compounds 9(a-f) and 10(a-f), the pyrazole-linezolid like conjugates 10(a-f) exhibited superior inhibitory activity than pyrazole-isoniazid 9(a-f) derivatives. The basic skeleton 1, 3-diphenyl pyrazole scaffold is supposed to have significant antitubercular activity. Further, we investigated the impact of substitution of *p*-position on the 3-phenyl of pyrazole ring. The *p*-position on the 3-phenyl of pyrazole ring substituted with electron withdrawing (Cl, F, Br, and NO<sub>2</sub>) or electron donating groups (-OCH<sub>3</sub>) were investigated. In the pyrazole-isoniazid 9(a-f) derivatives, the introduction of electron withdrawing groups like *p*-fluoro 9c (MIC = 16.06  $\mu\text{M}$ ) and *p*-bromo 9d (MIC = 13.91  $\mu\text{M}$ ) on 3-phenyl of pyrazole leads to increase activity by two folds when compared with electron donating group *p*-OCH<sub>3</sub> 9f (MIC 31.15  $\mu\text{M}$ ).

Other electron withdrawing groups like *p*-chloro 9b (MIC = 61.70  $\mu\text{M}$ ) and *p*-nitro 9e (MIC = 60.07  $\mu\text{M}$ ) on 3-phenyl of pyrazole showed least antitubercular activity. In pyrazole-linezolid like conjugates 10(a-f), the presence of electron withdrawing fluoro group on phenyl ring largely influences the antitubercular activity (10c; MIC = 13.94  $\mu\text{M}$ ). Along with this fluoro group, the introduction of strong electron withdrawing bromo group at *p*-position on the 3-phenyl of pyrazole ring (10d; MIC = 12.30  $\mu\text{M}$ ) leads to significant increase in antitubercular effect. Introduction of least electron withdrawing groups like *p*-chloro 10b (MIC = 26.92  $\mu\text{g/ml}$ ) and *p*-fluoro 10e (MIC = 26.30  $\mu\text{M}$ ) also responsible for enhancement of antitubercular activity. The compounds with unsubstituted phenyl ring and presence of electron donating *p*-OCH<sub>3</sub> group resulted in a decrease in activity (10a, MIC = 134.70  $\mu\text{M}$ ; 10a, MIC = 116.22  $\mu\text{g/ml}$ ).

Table 1: *In vitro* antitubercular activity, cytotoxicity study and selectivity index of compounds 9(a-f) and 10(a-f)

Entry	R	Anti-TB MIC ( $\mu\text{M}$ )	Cytotoxicity HeLa $\text{GI}_{50}$ ( $\mu\text{M}$ )	Selectivity Index
9a	H	134.70	-	-
9b	4-Cl	61.70	-	-
9c	4-F	16.06	>256.96	>16
9d	4-Br	13.91	>222.67	>16
9e	4-NO <sub>2</sub>	60.07	-	-
9f	4-OCH <sub>3</sub>	31.15	-	-
10a	H	116.22	-	-
10b	4-Cl	26.92	-	-
10c	4-F	13.94	>223.11	>16
10d	4-Br	12.30	>201.80	>16
10e	4-NO <sub>2</sub>	26.30	-	-
10f	4-OCH <sub>3</sub>	108.64	-	-
<sup>a</sup> Isoniazid		0.87	NA	NA
<sup>a</sup> Ciprofloxacin		9.41	NA	NA
<sup>a</sup> Rifampicin		0.70	NA	NA
<sup>b</sup> Paclitaxel		NA	0.004	NA

<sup>a</sup>Standard antitubercular drug; standard anticancer drug; DPPH: 1,1-Diphenyl-1-picrylhydrazyl; NA: Not applicable; NT: Not tested; Selectivity index (S. I.) = ( $\text{GI}_{50}/\text{MIC}$ );  $\text{GI}_{50}$  = concentration inhibits 50 % growth of cells. (Biological activities performed in the triplet. The results were with 10 % of standard error mean, SEM)

### *In vitro* cytotoxicity study

The safety profile of antitubercular active compounds with  $\text{MIC} < 20 \mu\text{M}$  was assayed for *in vitro* cytotoxicity against HeLa cell line by Sulforhodamine B (SRB) assay method. None of the synthesized compounds 9c, 9d, 10c and 10d was cytotoxic at concentration up to 200  $\mu\text{M}$ .

### Selectivity index

Table 1 observations interpret that, the screened compounds 9c, 9d, 10c and 10d were shown selectivity index >10. This indicates that the screened compounds were further investigated for future drug development process.

### Computational studies

#### Molecular docking study

In order to rationalize the observed antitubercular results and to get insight into the inhibition pattern, interactions of synthesized compounds were analyzed and depicted using molecular docking

studies. Visual inspection of the minimum energy docked poses revealed that these derivatives snugly fitted into the binding pocket of *InhA* enzyme (PDB: 2X23) making close contacts with the surrounding residues. The synthesized compounds showed good docking scores. A detailed per-residue interaction analysis between the protein and the most active compound 10d only is elucidated for the sake of brevity through which we can speculate regarding the binding patterns in the cavity. The binding interaction of 10d is presented in fig. 2. It showed multiple interactions with the residues in the active site, however for visibility and clarity only selected interacting residues are exhibited. The compound 10d helped in the stabilization of compound by making hydrophobic interactions with amino acid residues like ILE95, PHE97, ILE202, MET103, MERT161, MET199, VAL203, LEU218, PRO193, PHE149, ILE194, ALA191, ILE21 and MET147 of *InhA* enzyme. Strong binding of 10d with the active site of *InhA* enzyme is also contributed by its position in the pi-interaction in space of TYR158. On the basis of activity data and docking result, it was found that the compound 10d has potential to inhibit *InhA* enzyme and can be processed further to develop a lead compound.

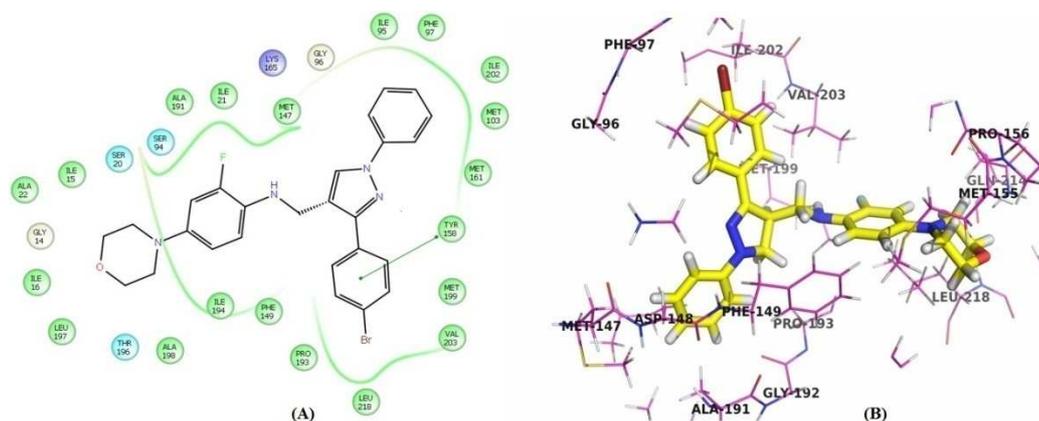


Fig. 2: (A) 2D docking pose of compound 10d; (B) 3D docking pose of compound 10d

### Prediction of physicochemical and pharmacokinetic properties

All the synthesized compounds 9(a-f) and 10(a-f) were evaluated for prediction of physicochemical and pharmacokinetic properties for drug-likeness using QikProp module of Schrodinger 9.0 [21]. As per Lipinski's rule-of-five, a molecule likely to be developed as an orally active drug candidate should show no more than one violation of the four parameters such as molecular weight (MW<500), hydrogen bond donor (HBD; 0-6), hydrogen bond acceptor (HBA; 2-20), and partition coefficient (QlogP; 1-5) [22]. Except for partition coefficient (3.52-6.88), other physicochemical properties like molecular weight (371.441-491.420), hydrogen bond donor (1-2) and hydrogen bond acceptor (4.7-6.75) found the results within their acceptable range. So, the synthesized compound has shown no more than one violation and can be developed as a good oral drug

candidate. Other parameters as solvent accessible surface area (SASA; 709.441-829.983) and a number of rotatable bonds (#rotor; <15; 3-5) also have a great influence on the oral bioavailability of the drug molecules and found in acceptable range. Investigation of intestinal absorption, permeation, and metabolism is also greatly contributed to minimizing the clinical trial failures. Hence, predicted Caco-2 cell permeability (QPPCaco), apparent MDCK cell permeability (QPPMDCK), predicted aqueous solubility (QPlogS), percent scale human oral absorption (%HOA), and number of likely metabolic reactions (#metab) were also analyzed and displayed a reliable results falling in the prescribed range as shown in table 2. From all *in silico* prediction of physicochemical and pharmacokinetic properties (ADME) study, these compounds could be developed as an oral active drug, making them potentially capable for antitubercular treatment.

**Table 2: Physicochemical properties for the titled compounds 9(a-f) and 10(a-f) predicted by QikProp**

Entry	MW (<500)	HBD (0-6)	HBA (2-20)	QlogPo/w (1-5)	SASA (300-1000)	#rotor (0-15)	No. of violations (<1)	QPPCaco (>500 great)	QPPMDCK (>500)	QPlogS (-6.5-0.5)	% HOA (>80%)	#metab (1-8)
9a	371.441	2	6	4.22	709.411	4	0	857.23	418.833	-5.576	100	3
9b	405.886	2	6	4.69	732.779	4	0	879.61	1060.693	-6.259	100	3
9c	389.431	2	6	4.43	718.268	4	0	879.46	776.501	-5.932	100	3
9d	450.337	2	6	4.76	738.357	4	0	879.10	1134.407	-7.153	100	3
9e	416.438	2	7	3.52	754.269	5	0	99.065	40.65	-6.09	83.30	4
9f	401.467	2	6.75	4.98	829.983	5	0	1318.83	667.204	-5.888	100	4
10a	430.524	1	4.7	6.30	742.260	3	1	5954.59	5231.022	-7.447	100	4
10b	464.969	1	4.7	6.79	766.793	3	1	5884.19	10000	-8.159	100	4
10c	448.514	1	4.7	6.52	748.262	3	1	5910.85	9364.362	-7.818	100	4
10d	491.420	1	4.7	6.88	775.253	3	1	5982.35	10000	-9.081	100	4
10e	475.521	1	5.7	5.59	782.232	4	1	701.45	518.474	-7.965	100	5
10f	460.550	1	5.45	6.36	779.035	4	1	5750.95	5026.507	-7.759	100	5

MW-Molecular weight; HBD-Hydrogen bond donor; HBA-Hydrogen bond acceptor; QlogPo/w-Predicted octanol/water partition coefficient; SASA-Solvent accessible surface area; #rotor-Number of rotatable bonds; QPPCaco-Possible CaCO cell permeability; QPPMDCK-Predicted apparent MDCK cell permeability; QPlogS-Predicted aqueous solubility; % HOA-Percent human oral absorption; #metab-Number of likely metabolic reactions.

### CONCLUSION

In conclusion, some new structural hybrids of 1, 3-diphenyl pyrazole bearing isoniazid and linezolid like moieties were synthesized and investigated for their *in vitro* antitubercular activity with an anticipation of generating new structural leads. Four of the synthesized compounds (9c, 9d, 10c and 10d) possessing electron withdrawing groups such as *fluoro* and *bromo* at para position were identified as the most potent antitubercular agents. The potent antitubercular activity of the most potent compounds was accompanied with a relatively low level of cytotoxicity and higher selective index. Additionally, the molecular docking study of compounds helps to estimate possible binding interactions with the active site of *InhA* enzyme-enzyme. The analysis of the physicochemical and pharmacokinetic properties for the titled compounds has shown that these compounds have good oral-like properties and can be developed as oral drug candidates.

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### AUTHORS' CONTRIBUTION

Sameer I. Shaikh researched and wrote this overview. Zahid Zaheer and Santosh N. Mokale provided guidance, critical review and revision. Deepak K. Lokwani carried out all computational studies (Molecular docking and *in silico* properties prediction). All authors' read and approved the final version of this manuscript.

### CONFLICT OF INTERESTS

The authors declare that there is no conflict of interest

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