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Studies on Fungal Products. VII.¹⁾ The Structures of Meleagrins and 9-*O*-*p*-Bromobenzoylmeleagrins

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The structure of meleagrins (**1**), isolated from *Penicillium meleagrins*, was determined by X-ray crystallographic analysis of 9-*O*-*p*-bromobenzoylmeleagrins (**7**) monohydrate. The crystal structure of **7** monohydrate was determined by the heavy atom method; the final *R* value without hydrogen atoms was 0.094. The structure of meleagrins was established as the 9-*O*-demethyl compound of oxaline (**4**), isolated from *Penicillium oxalicum*. The absolute configurations at N₁, C₂, and C₃ are *S*, *S*, and *R*, respectively.

Keywords—meleagrins; *p*-bromobenzoylmeleagrins; oxaline; *Penicillium meleagrins*; X-ray analysis

In the course of screening *Penicillium* species for antimicrobial metabolites, meleagrins (**1**), C₂₃H₂₃N₅O₄, was isolated as a main component from the culture filtrate of *Penicillium meleagrins* BIORGE IFO 8143.¹⁾ It was clarified that meleagrins showed structural similarity with the tremorgenic mycotoxins formed from two amino acids and one isoprenyl moiety, but its structure could not be established. In this paper, we determined the structure of meleagrins as **1**.

On acetylation, **1** afforded a monoacetate (**2**), C₂₅H₂₅N₅O₅, which showed a proton nuclear magnetic resonance (¹H-NMR) signal at δ 2.27 (3H, s) assigned to the methyl protons of an acetoxyl group. The ¹H-NMR signal of one of the olefinic protons at δ 5.50 (s) in **1** shifted to δ 5.97 (s) after acetylation. This suggested the presence of an enol group in **1**. The signal of methoxy protons appeared at δ 3.73 (s) in **1**. Thus two of the four oxygen functions in **1** have been assigned to one hydroxyl and one methoxyl. The remaining two oxygens must be the carbonyls of amide groups in view of the signals at δ 159.9 and 166.7 in the carbon 13 nuclear magnetic resonance (¹³C-NMR) spectrum. On methylation with diazomethane, **1** gave a monomethyl derivative (**3**), C₂₄H₂₅N₅O₄, which showed a ¹H-NMR signal at δ 2.45 (s) assigned to methyl protons of an *N*-methyl group. No other significant changes were observed at **3** compared with **1**. The vinyl proton signals at δ 5.06 (1H, d, *J* = 10.6 Hz), 5.10 (1H, d, *J* = 17.2 Hz) and 6.12 (1H, dd, *J* = 17.2 and 10.6 Hz) disappeared on hydrogenation of **1** to the dihydro compound (**5**), in which ethyl group signals appeared at δ 0.88 (3H, t, *J* = 9 Hz) and 1.28 (2H, q-like). At the same time, two methyl signals at δ 1.24 and 1.35 in **1** shifted to higher field (δ 1.12). The presence of the vinyl group was confirmed by a decoupling experiment with irradiation at δ 6.12. The doublets at δ 5.06 and 5.10 changed to singlets. In addition to the above evidence, the base peak at *m/z* 364 in the mass spectrum of **1** attributable to the loss of a C₅H₉ radical suggested the presence of a 1,1-dimethylallyl group in the molecule of **1**. On hydrogenation with zinc and acetic acid, **1** afforded another dihydro compound (**6**), which yielded histidine on subsequent acid hydrolysis. The above result suggested the presence of a dehydrohistidine moiety in **1**, and therefore the remaining nitrogen atoms were considered to

TABLE I. Crystal Data for 7 Monohydrate

Monoclinic	$C_{30}H_{26}BrN_5O_5 \cdot H_2O$
Space group $C2$	$F_w = 634.49$
$a = 37.825 (20) \text{ \AA}$	$F(000) = 1404$
$b = 7.833 (4) \text{ \AA}$	$Z = 4$
$c = 11.086 (6) \text{ \AA}$	$\mu(\text{CuK}\alpha) = 21.7 \text{ cm}^{-1}$
$\beta = 104.72 (5)^\circ$	$D_c = 1.326 \text{ g} \cdot \text{cm}^{-3}$
$V = 3177 (5) \text{ \AA}^3$	$D_m = 1.330 \text{ g} \cdot \text{cm}^{-3}$

TABLE II. Final Atomic Parameters ($\times 10^4$, for Br $\times 10^5$) and Equivalent Thermal Parameters, with Estimated Standard Deviations in Parentheses

	x	y	z	$B_{\text{eq.}} (\text{\AA}^2)$
BR	6987 (7)	0 (0)	105419 (21)	7.71 (0.5)
N1	-1558 (3)	5978 (19)	4131 (12)	3.8 (0.2)
C2	-1412 (4)	7664 (24)	4656 (14)	3.6 (0.3)
C3	-1070 (4)	7940 (28)	4111 (16)	4.3 (0.3)
C3a	-1208 (5)	7030 (29)	2854 (15)	4.6 (0.3)
C4	-1058 (6)	7016 (34)	1814 (18)	6.4 (0.4)
C5	-1222 (6)	5929 (38)	849 (19)	7.8 (0.5)
C6	-1517 (6)	4920 (47)	879 (18)	8.6 (0.5)
C7	-1669 (5)	4827 (36)	1938 (16)	6.4 (0.4)
C7a	-1497 (5)	6001 (30)	2912 (17)	5.6 (0.4)
C8	-767 (5)	6778 (25)	4855 (17)	4.6 (0.3)
C9	-763 (4)	6178 (23)	5982 (16)	3.9 (0.3)
C10	-1030 (4)	6659 (23)	6671 (15)	3.5 (0.3)
N11	-1324 (3)	7583 (19)	6037 (11)	3.6 (0.2)
C12	-1585 (4)	8516 (22)	6527 (14)	3.2 (0.2)
C13	-1822 (4)	9430 (24)	5425 (13)	3.8 (0.3)
N14	-1681 (3)	8999 (18)	4419 (12)	3.5 (0.2)
C15	-1589 (5)	8574 (26)	7743 (15)	4.4 (0.3)
C16	-1835 (5)	9480 (26)	8295 (15)	4.9 (0.3)
N17	-2151 (4)	10374 (25)	7750 (12)	5.2 (0.3)
C18	-2301 (6)	10893 (33)	8592 (17)	6.3 (0.4)
N19	-2100 (5)	10319 (31)	9741 (14)	7.2 (0.4)
C20	-1803 (5)	9447 (32)	9564 (16)	5.9 (0.4)
C21	-923 (5)	9887 (31)	4166 (16)	5.0 (0.3)
C22	-902 (5)	10612 (24)	5476 (19)	5.5 (0.4)
C23	-1048 (6)	11974 (29)	5729 (23)	6.8 (0.5)
C24	-1174 (6)	10912 (31)	3082 (19)	6.3 (0.4)
C25	-508 (5)	9909 (40)	4043 (21)	7.4 (0.5)
C26	-1952 (6)	4268 (33)	4806 (23)	7.6 (0.5)
C27	-194 (5)	5682 (25)	7398 (17)	4.7 (0.3)
C28	31 (4)	4278 (24)	8095 (14)	3.5 (0.3)
C29	311 (4)	4773 (34)	9118 (15)	5.2 (0.3)
C30	509 (5)	3496 (31)	9830 (16)	5.2 (0.4)
C31	425 (5)	1776 (28)	9502 (16)	5.0 (0.4)
C32	153 (4)	1276 (27)	8496 (15)	4.3 (0.3)
C33	-37 (5)	2610 (25)	7786 (15)	4.1 (0.3)
O1	-1921 (3)	5782 (17)	4121 (11)	4.7 (0.2)
O2	-499 (3)	5039 (19)	6596 (9)	4.2 (0.2)
O3	-967 (3)	6336 (18)	7813 (10)	4.9 (0.2)
O4	-2072 (3)	10332 (18)	5393 (9)	4.3 (0.2)
O5	-129 (4)	7190 (19)	7458 (14)	6.6 (0.3)
Ow	-2203 (6)	10296 (34)	2180 (11)	13.1 (0.5)

be those of another amino acid moiety in the molecule. The $^1\text{H-NMR}$ signals of aromatic protons in **1** appeared at δ 6.96 (br d, $J=7.5$ Hz), 7.07 (br t, $J=7.5$ Hz), 7.30 (br t, $J=7.5$ Hz) and 7.58 (br d, $J=7.5$ Hz). This result suggests that the tryptophan moiety of **1** has no substituent on the aromatic ring. Hence the methoxyl group must be on the nitrogen (N_1).

Meleagrins (**1**) is closely similar in all of the spectroscopic data, except for the absence of one (*O*)-methyl, to oxaline (**4**), which was recently isolated from *Penicillium oxalicum*.^{3,4} From a comparison of the $^1\text{H-NMR}$ spectra of **1** and **4**, meleagrins was suggested to be represented by the chemical formula **1**, which is 9-*O*-demethyloxaline. We tried to obtain oxaline (**4**) by methylation of **1**, but only N_{14} -methylmeleagrins (**3**) was obtained. In order to determine the exact structure of **1** including the absolute stereochemistry, an X-ray structure analysis of 9-*O*-*p*-bromobenzoylmeleagrins (**7**) monohydrate was undertaken.

Crystals of **7** monohydrate were grown in chloroform-ethanol solution as pale yellow prisms. The size of the X-ray specimen was about $0.3 \times 0.07 \times 0.15$ mm. The lattice constants and intensity data were collected on a Philips PW 1100 diffractometer using $\text{CuK}\alpha$ radiation monochromated by a graphite plate. The crystal data are summarized in Table I. A total of 2455 reflections were measured in a 2θ range of $6-130^\circ$ as above the 2σ (I) level; 296 of these were Friedel reflections. Intensities were corrected for Lorenz and polarization factors, but no absorption correction was applied.

The crystal structure was solved by the heavy atom method, and the refinement was carried out by the block-diagonal least-squares method using HBLS IV.⁵ To determine the absolute configuration, the crystal structure was first refined with anisotropic thermal parameters to an R value of 9.5% excluding hydrogen atoms, and then the anomalous dispersion corrections were applied to the bromine atoms. A comparison between the observed and calculated amplitude ratios of $|F(hkl)|/|F(\bar{h}\bar{k}\bar{l})|$ for 296 Friedel pairs indicated that 88 out of 96 pairs for which the observed and calculated ratios differed by more than 5% from unity showed the same absolute configuration. The final atomic parameters are listed in Table II.⁶

The absolute structure of 9-*O*-*p*-bromobenzoylmeleagrins was determined to be as shown

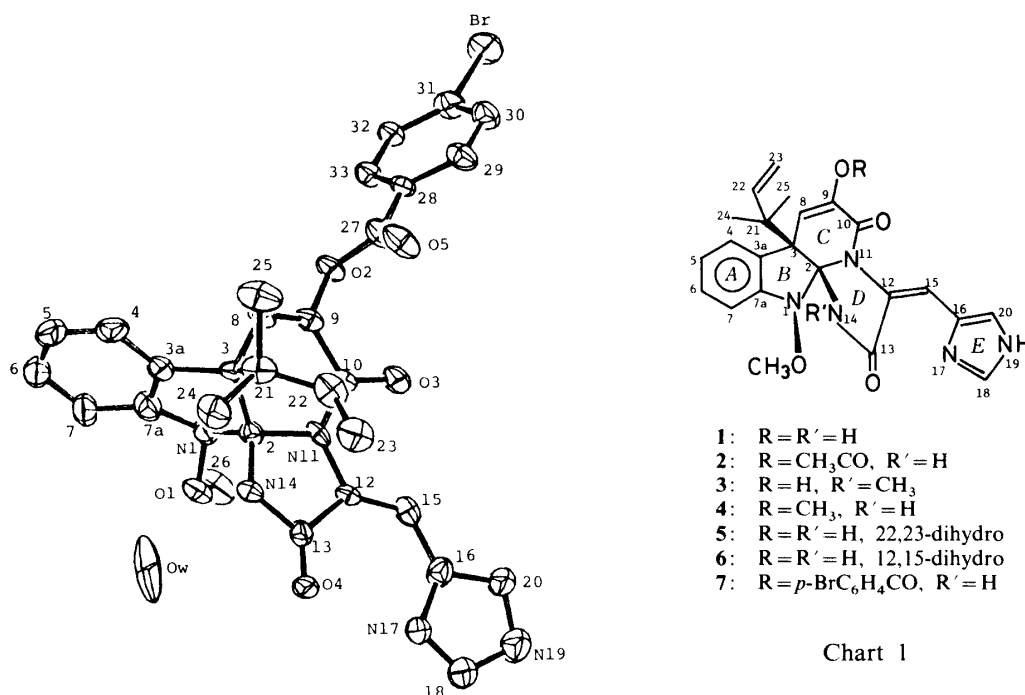


Fig. 1. The Molecular Structure of 9-*O*-*p*-Bromobenzoylmeleagrins (**7**) monohydrate

(247) (positive maximum), $+5.09 \times 10^4$ (257) (negative maximum), $+6.14 \times 10^4$ (275) (positive maximum), -3.94×10^4 (342) (negative maximum). $^1\text{H-NMR}$ (CDCl_3) δ : 1.24 and 1.35 (3H, each, s, 24- and 25- CH_3 's), 3.73 (3H, s, 1- OCH_3), 5.06 (1H, d, $J=10.6$ Hz, 23-H), 5.10 (1H, d, $J=17.2$ Hz, 23-H), 5.50 (1H, s, 8-H), 6.12 (1H, dd, $J=17.2$ and 10.6 Hz, 22-H), 6.30 (1H, br, NH or OH), 6.96 (1H, br d, $J=7.5$ Hz), 7.07 (1H, br t, $J=7.5$ Hz), 7.25 (1H, s, 20-H), 7.30 (1H, br t, $J=7.5$ Hz), 7.58 (1H, br d, $J=7.5$ Hz), 7.61 (1H, s, 18-H), 8.27 (1H, s, 15-H), 12.72 (1H, br NH). $^1\text{H-NMR}$ (in acetone- d_6) δ : 1.32 and 1.38 (3H each, s, 24- and 25- CH_3 's), 3.74 (3H, s, 1- OCH_3), 5.05 (1H, d, $J=10$ Hz, 23-H), 5.10 (1H, d, $J=16$ Hz, 23-H), 5.50 (1H, s, 8-H), 6.18 (1H, br, 22-H), 6.97 (1H, d, $J=8$ Hz), 7.07 (1H, t, $J=8$ Hz), 7.26 (1H, t, $J=8$ Hz), 7.46 (1H, s, 20-H), 7.65 (1H, d, $J=8$ Hz), 7.95 (1H, s, 18-H), 8.50 (1H, s, 15-H).

9-O-Acetylmeleagrins (2)—**1** (50 mg) was acetylated with acetic anhydride (1 ml) and pyridine (0.5 ml) at room temperature for 48 h to give the monoacetate (**2**) (40 mg) as a pale yellow crystalline powder, mp 247°C from benzene-petroleum ether. IR $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$: 3400, 3300 (NH), 1770 (COO), 1708, 1690 ($\text{C}=\text{O}$), 1197, 1112. UV $\lambda_{\text{max}}^{\text{EtOH}} \text{nm}$ (log ϵ): 204 (4.68), 223 sh (4.45), 295 sh (4.17), 347 (4.50). MS m/z : 475.1871 (M^+ , 475.1876 required for $\text{C}_{25}\text{H}_{25}\text{N}_5\text{O}_5$), 444, 433, 407 ($\text{M}^+ - \text{C}_5\text{H}_8$, base peak), 406. $^1\text{H-NMR}$ (CDCl_3) δ : 1.31 and 1.35 (3H, each, s, 24- and 25- CH_3 's), 2.27 (3H, s, 9-OAc), 3.77 (3H, s, 1- OCH_3), 5.04 (1H, d, $J=10$ Hz, 23-H), 5.10 (1H, d, $J=17$ Hz, 23-H), 5.97 (1H, s, 8-H), 6.09 (1H, dd, $J=17$ and 10 Hz, 22-H), 6.6–7.6 (4H, m, aromatic-H's), 7.34 (1H, s, 20-H), 7.50 (1H, s, 18-H), 8.28 (1H, s, 15-H).

N(14)-Methylmeleagrins (3)—**1** (100 mg) was methylated in ether with CH_3N_2 overnight at room temperature. The reaction mixture was chromatographed on silica gel with 1% $\text{MeOH}-\text{CHCl}_3$ to give the monomethyl derivative (**3**) (65 mg) as pale yellow needles, mp 212°C (dec.) from MeOH. IR $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$: 3400, 3150 (NH, OH), 1715, 1680 ($\text{C}=\text{O}$). UV $\lambda_{\text{max}}^{\text{MeOH}} \text{nm}$ (log ϵ): 229 (4.02), 285 sh (3.63), 347 (4.06). MS m/z : 447.1902 (M^+ , 447.1905 required for $\text{C}_{24}\text{H}_{25}\text{N}_5\text{O}_4$). CD ($c=0.0088$, MeOH) $[\theta]^{20} \text{nm}$: -1.47×10^5 (226) (negative maximum), $+4.88 \times 10^4$ (247) (positive maximum), $+4.37 \times 10^4$ (256) (negative maximum), $+5.99 \times 10^4$ (280) (positive maximum), -4.78×10^4 (345) (negative maximum). $^1\text{H-NMR}$ (CDCl_3) δ : 1.21 and 1.32 (3H each, s, 24- and 25- CH_3 's), 2.45 (3H, s, NCH_3), 3.75 (3H, s, 1- OCH_3), 5.02 (1H, d, $J=17.2$ Hz, 23-H), 5.03 (1H, d, $J=10.6$ Hz, 23-H), 5.44 (1H, s, 8-H), 6.11 (1H, dd, $J=17.2$ and 10.6 Hz, 22-H), 6.8–7.2 (3H, m, aromatic-H's), 7.37 (1H, s, 20-H), 7.53 (1H, d, $J=7.7$ Hz), 7.70 (1H, s, 18-H), 8.31 (1H, s, 15-H), 13.05 (1H, br, NH). The reaction of **1** with Ag_2O and MeI gave a monomethyl derivative, which was identified as **3**.

22,23-Dihydromeleagrins (5)—Catalytic reduction of **1** (80 mg) with 5% Pd/C (20 mg) in AcOEt (20 ml) afforded **5** (55 mg) as pale yellow needles, mp 210°C (dec.) from chloroform. IR $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$: 3400, 3150 (NH, OH), 1700, 1670 ($\text{C}=\text{O}$). UV $\lambda_{\text{max}}^{\text{EtOH}} \text{nm}$ (log ϵ): 232 (4.06), 348 (4.08). MS m/z : 435.1936 (M^+ , 435.1906 required for $\text{C}_{23}\text{H}_{25}\text{N}_5\text{O}_4$), 366, 365 (base peak). $^1\text{H-NMR}$ (CDCl_3) δ : 0.88 (3H, t, $J=9$ Hz, 23- CH_3), 1.12 (6H, s, 24- and 25- CH_3 's), 1.28 (2H, q-like, 22- CH_2), 3.74 (3H, s, 1- OCH_3), 5.50 (1H, s, 8-H), 6.20 (1H, br, NH or OH), 6.9–7.6 (4H, m, aromatic-H's), 7.30 (1H, s, 20-H), 7.61 (1H, s, 18-H), 8.31 (1H, s, 15-H), 12.65 (1H, br, NH).

12,15-Dihydromeleagrins (6)—**1** (5 mg) was treated with zinc powder (40 mg) in acetic acid (0.5 ml) at 70°C for 3 h to give a dihydromeleagrins (**6**) (3 mg). MS m/z : 435.1920 (M^+ , 435.1907 required for $\text{C}_{23}\text{H}_{25}\text{N}_5\text{O}_4$), 405.1822 ($\text{M}^+ - \text{OCH}_2$, 405.1801 requires for $\text{C}_{22}\text{H}_{23}\text{N}_5\text{O}_3$).

9-O-*p*-Bromobenzoylmeleagrins (7)—**1** (80 mg) was treated with *p*-bromobenzoyl chloride (200 mg) and pyridine (2 ml) at room temperature for 24 h. The reaction mixture was chromatographed on silica gel with benzene-acetone (10:1) to afford the *p*-bromobenzoate (**7**) (75 mg) as pale yellow plates, mp 190°C (dec.) from CHCl_3 -MeOH or CHCl_3 -EtOH. IR $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$: 3200–2800 (NH), 1730 (COO), 1700, 1670 ($\text{C}=\text{O}$). MS m/z : 615 and 617 (M^+). Anal. Calcd for $\text{C}_{30}\text{H}_{26}\text{BrN}_5\text{O}_5 \cdot \text{H}_2\text{O}$: C, 56.79; H, 4.53; N, 11.04. Found: C, 56.32; H, 4.53; N, 10.86. $^1\text{H-NMR}$ (CDCl_3) δ : 1.36 and 1.41 (3H each, s, 24- and 25- CH_3 's), 3.84 (3H, s, 1- OCH_3), 5.11 (1H, d, $J=10.8$ Hz, 23-H), 5.18 (1H, d, $J=17.2$ Hz, 23-H), 6.13 (1H, s, 8-H), 6.18 (1H, dd, $J=17.2$ and 10.8 Hz, 22-H), 6.90–7.37 (4H, m, aromatic-H's), 7.19 (1H, s, 20-H), 7.62 (2H, d, $J=8.5$ Hz), 7.63 (1H, s, 18-H), 8.00 (2H, d, $J=8.5$ Hz), 8.34 (1H, s, 15-H).

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