

Supporting Information for:

**Mechanism and Activity of Ruthenium Olefin Metathesis Catalysts**

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*Arnold and Mabel Beckman Laboratories of Chemical Synthesis, California Institute of Technology, Pasadena, California 91125 (U.S.A.)***Table S1.** Rate Constants for and  $T_1$  values for Phosphine Exchange in 1–14.

Complex	$k_{\text{obs}}/\text{s}^{-1}$	eq $\text{PR}_3$	T/K	$T_{1\text{F}}/\text{s}$	$T_{1\text{C}}/\text{s}$
1	$0.116 \pm 0.002$	3 eq	313	9.2	2.3
	$0.381 \pm 0.010$	3 eq	323	9.8	2.7
	$1.21 \pm 0.02$	3 eq	333	8.9	3.1
	$3.56 \pm 0.06$	3 eq	343	7.8	3.6
	$9.57 \pm 0.06$	3 eq	353	7.3	4.3
	$1.22 \pm 0.04$	1.5 eq	333	8.3	3.3
	$1.13 \pm 0.04$	10 eq	333	10.0	2.7
	$1.11 \pm 0.03$	20 eq	333	9.0	2.3
2	$0.40 \pm 0.01$	3 eq	313	8.3	2.1
	$11.4 \pm 0.2$	3 eq	343	13.0	2.5
	$30.3 \pm 0.11$	3 eq	353	11.4	2.6
	$0.45 \pm 0.03$	10 eq	313	3.7	2.3
	$0.34 \pm 0.04$	20 eq	313	5.3	1.8
	3	$0.46 \pm 0.02$	3 eq	273	5.7
$1.56 \pm 0.08$		3 eq	283	6.9	1.4
$8.3 \pm 0.5$		3 eq	297	10	1.6
$16.5 \pm 0.5$		3 eq	303	10	1.7
$28 \pm 1$		3 eq	308	15	1.5
$44 \pm 2$		3 eq	313	5.8	3.0
$15.9 \pm 0.8$		10 eq	303	11	0.84
$15.5 \pm 0.6$		20 eq	303	8	0.74
5	$0.202 \pm 0.008$	3 eq	313	6.3	2.3
	$2.18 \pm 0.04$	3 eq	333	9.6	3.0
	$6.55 \pm 0.14$	3 eq	343	3.9	3.5
	$19.4 \pm 0.6$	3 eq	353	5.4	2.4
	$1.94 \pm 0.06$	10 eq	333	12.1	2.1

	1.81 ± 0.08	20 eq	333	10.4	1.9
6	0.037 ± 0.003	3 eq	333	8.7	3.0
	0.090 ± 0.004	3 eq	343	11.1	2.9
	0.326 ± 0.015	3 eq	353	10.6	3.3
	0.838 ± 0.016	3 eq	363	7.7	3.2
	2.16 ± 0.02	3 eq	373	7.0	2.9
	0.47 ± 0.02	10 eq	353	10.5	3.4
	0.41 ± 0.01	20 eq	353	9.5	2.7
7	0.163 ± 0.006	3 eq	333	7.8	5.2
	0.50 ± 0.01	3 eq	343	6.9	5.3
	1.42 ± 0.04	3 eq	353	5.8	5.2
	3.68 ± 0.12	3 eq	363	4.4	5.1
	0.43 ± 0.02	10 eq	343	10.8	4.7
	0.41 ± 0.06	20 eq	343	11.6	2.4
8	0.04 ± 0.01	1.5 eq	343	4.9	3.4
	0.126 ± 0.006	1.5 eq	353	8.4	3.6
	0.355 ± 0.016	1.5 eq	363	7.4	3.9
	1.02 ± 0.06	1.5 eq	373	12.6	3.4
	0.121 ± 0.008	5 eq	353	10.5	3.7
	0.12 ± 0.02	10 eq	353	9.2	3.3
9	0.52 ± 0.03	1.5 eq	353	9.6	3.8
	1.8 ± 0.1	1.5 eq	363	9.5	3.7
	4.3 ± 0.2	1.5 eq	373	13.5	4.0
	10.9 ± 0.5	1.5 eq	383	6.0	1.3
	0.52 ± 0.02	5 eq	353	8.3	3.7
	0.50 ± 0.02	10 eq	353	8.3	3.1
10	1.2 ± 0.1	1.5 eq	323	11.0	1.6
	5.0 ± 0.6	1.5 eq	333	11.6	1.9
	10 ± 1	1.5 eq	343	18.5	1.6
	30 ± 3	1.5 eq	353	12.4	1.7
	30 ± 3	5 eq	353	13.0	1.8
	29 ± 3	10 eq	353	13.2	2.0
11	0.049 ± 0.008	1.5 eq	303	17.8	4.1
	0.138 ± 0.005	1.5 eq	313	18.5	4.2
	0.420 ± 0.008	1.5 eq	323	20.0	4.5
	1.22 ± 0.02	1.5 eq	333	16.7	5.5

	0.11 ± 0.05	5 eq	313	20.0	3.5
	0.14 ± 0.04	10 eq	313	19.2	3.3
12	0.165 ± 0.006	1.5 eq	353	6.4	3.6
	0.47 ± 0.01	1.5 eq	363	7.1	3.9
	1.26 ± 0.04	1.5 eq	373	7.7	4.0
	3.37 ± 0.12	1.5 eq	383	12.5	4.3
	0.17 ± 0.05	5 eq	353	9.6	2.9
	0.18 ± 0.06	10 eq	353	8.5	3.8
14	0.09 ± 0.01	1.5 eq	363	7.1	1.8
	0.21 ± 0.02	1.5 eq	373	8.3	3.0
	0.54 ± 0.03	1.5 eq	383	6.2	3.9
	1.41 ± 0.08	1.5 eq	393	5.2	3.7
	1.2 ± 0.1	5 eq	393	5.1	3.2
	1.0 ± 0.2	10 eq	393	4.7	3.1

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**Table S2.**  $T_1$  Analysis for complexes **1-14** in  $C_7D_8$  solution.

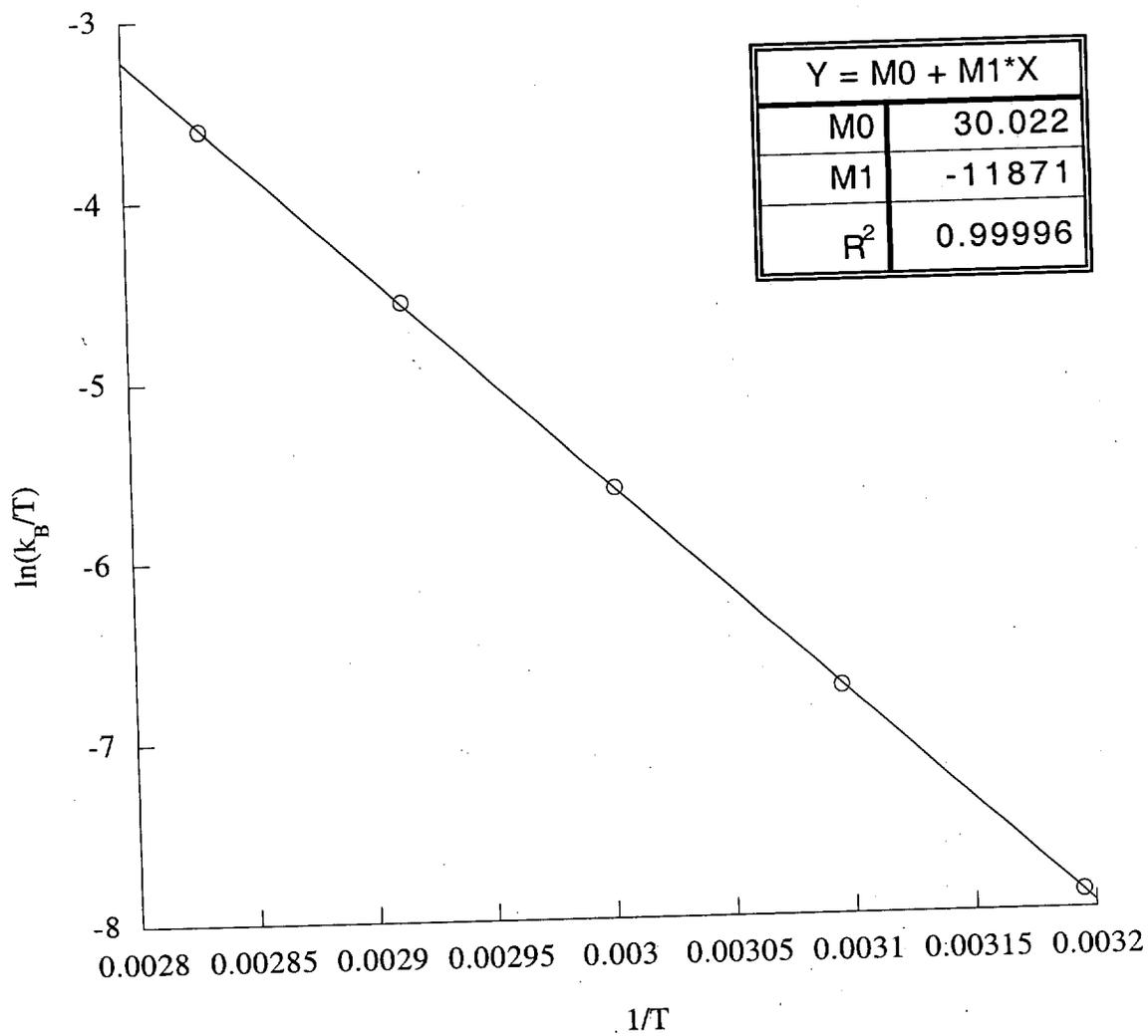
Complex	$T_1$ ( $s^{-1}$ )
PCy <sub>3</sub>	11.1 ± 0.2
PPh <sub>3</sub>	23.7 ± 0.6
PBn <sub>3</sub>	3.60 ± 0.09
<b>1</b>	2.46 ± 0.02
<b>2</b>	2.31 ± 0.02
<b>3</b>	1.91 ± 0.11
<b>5</b>	2.82 ± 0.02
<b>6</b>	2.78 ± 0.04
<b>7</b>	4.06 ± 0.03
<b>8</b>	3.07 ± 0.06
<b>9</b>	2.09 ± 0.03
<b>10</b>	1.58 ± 0.09
<b>11</b>	3.5 ± 0.2
<b>12</b>	1.6 ± 0.1
<b>14</b>	1.81 ± 0.02

**Table S3.** Values of  $k_1$  from  $1/k_{\text{obs}}$  versus  $[\text{PR}_3]/[\text{olefin}]$ .

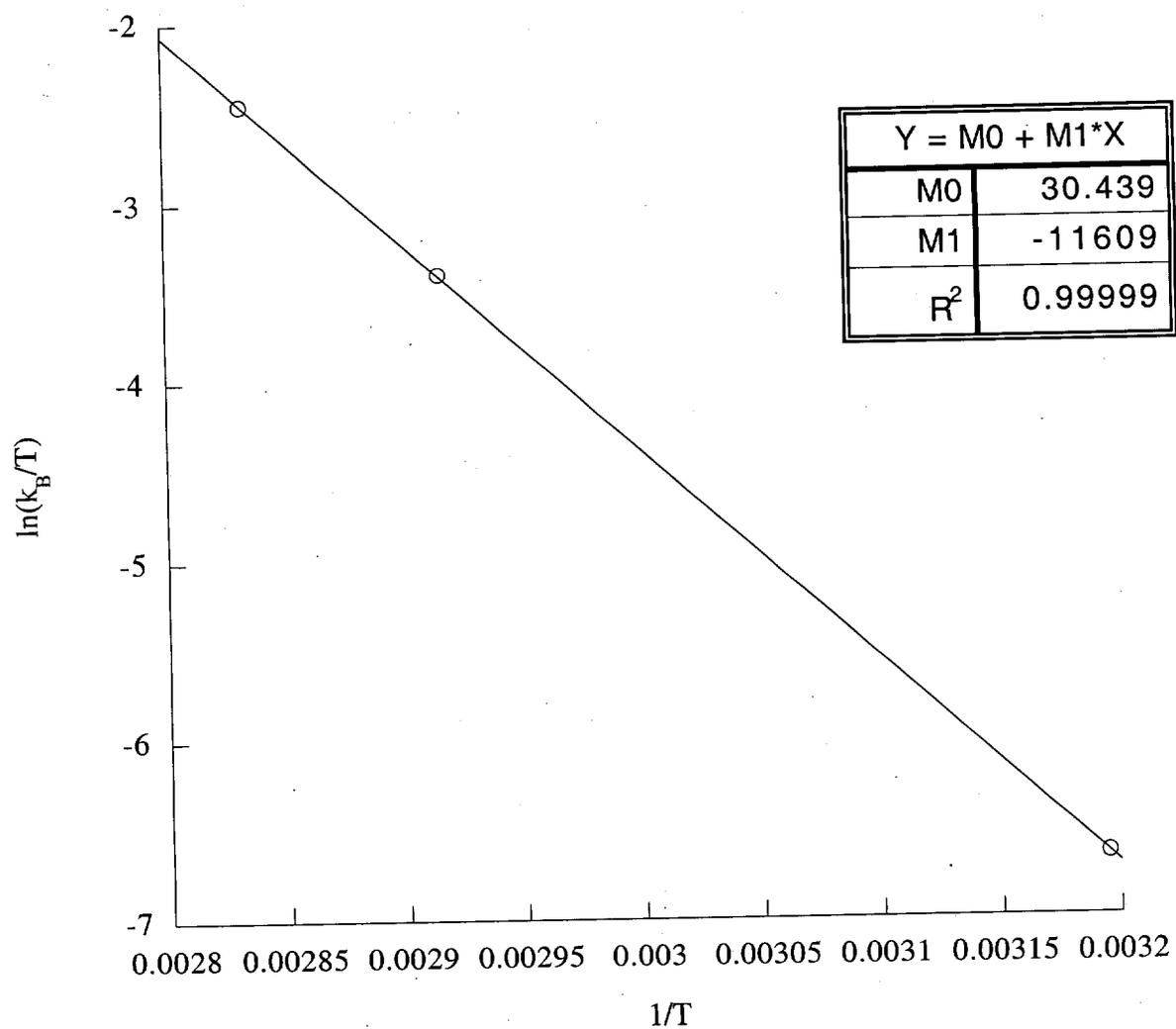
Complex	$k_1$ ( $\text{s}^{-1}$ )	$k_1$ (predicted) ( $\text{s}^{-1}$ )
1	$2.2 \times 10^{-1}$	$7.7 \times 10^{-1}$
2	$-7.0 \times 10^{-3}$	2.6
3	$-1.0 \times 10^{-1}$	$2.4 \times 10^2$
6	$2.7 \times 10^{-2}$	$2.3 \times 10^{-2}$
8	$3.1 \times 10^{-2}$	$3.2 \times 10^{-2}$
10	$1.7 \times 10^{-2}$	1.4
11	$1.3 \times 10^{-2}$	$2.5 \times 10^{-2}$
12	$5.0 \times 10^{-3}$	$4.3 \times 10^{-3}$

Note:  $k_{-1}$  values are from  $1/\text{intercept}$  and therefore have a significant amount of error associated with them. The negative values for catalysts 2 and 3 reflect the fact that  $1/k_1$  in these systems is very close to zero, and the error inherent in the intercept is relatively large.

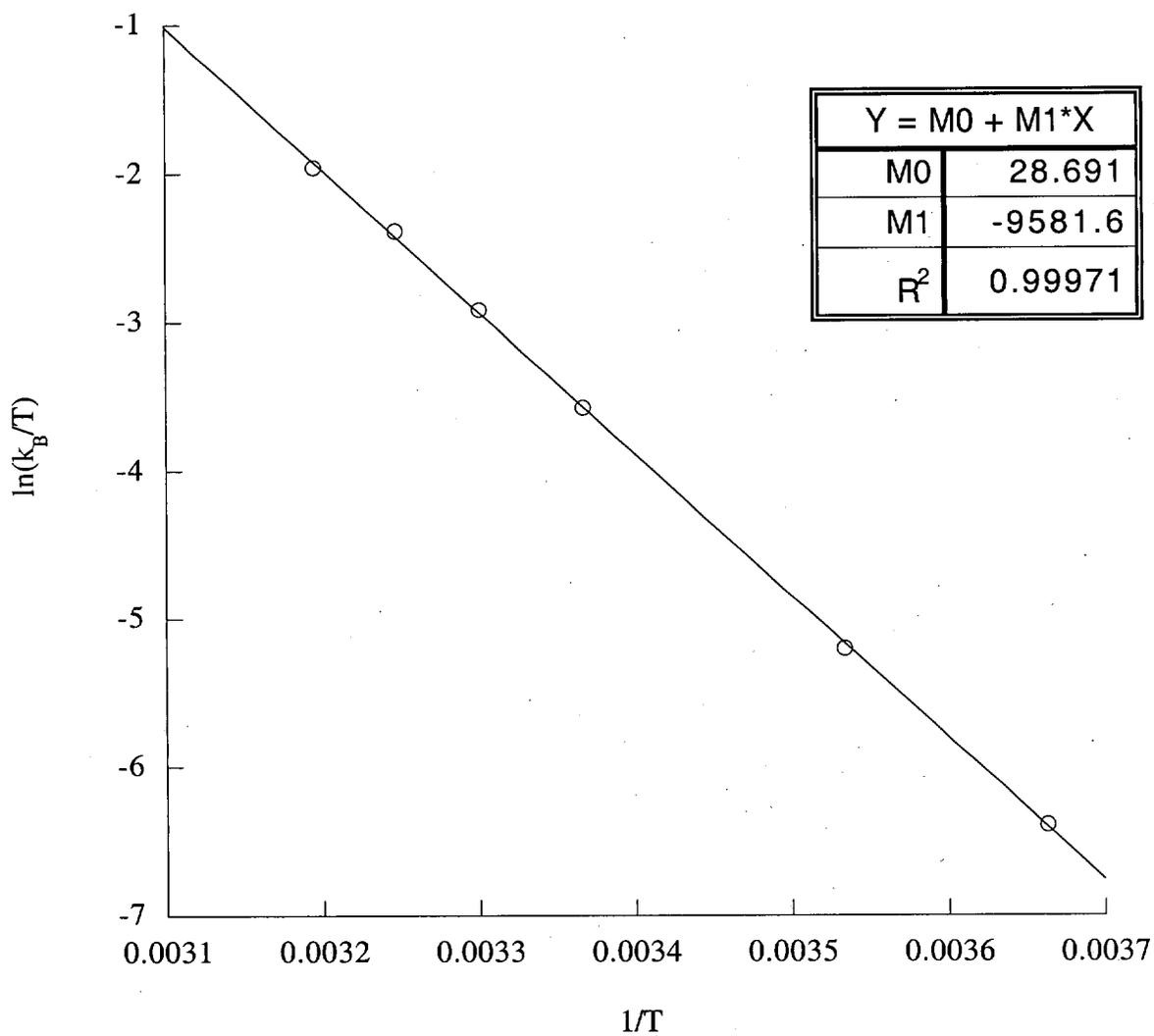
Figure S1. Eyring Plot for Phosphine Exchange in Catalyst 1



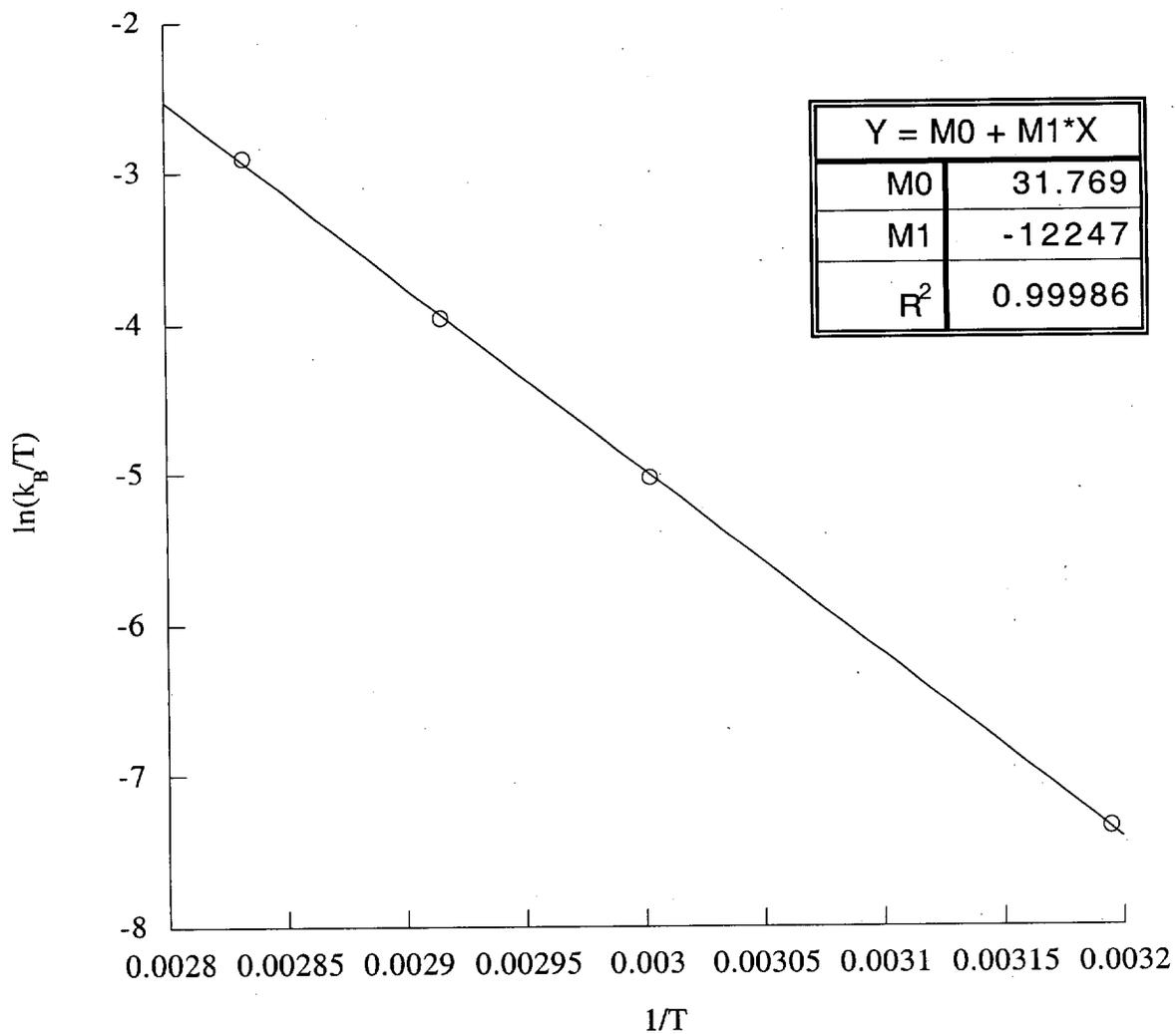
**Figure S2. Eyring Plot for Phosphine Exchange in Catalyst 2**



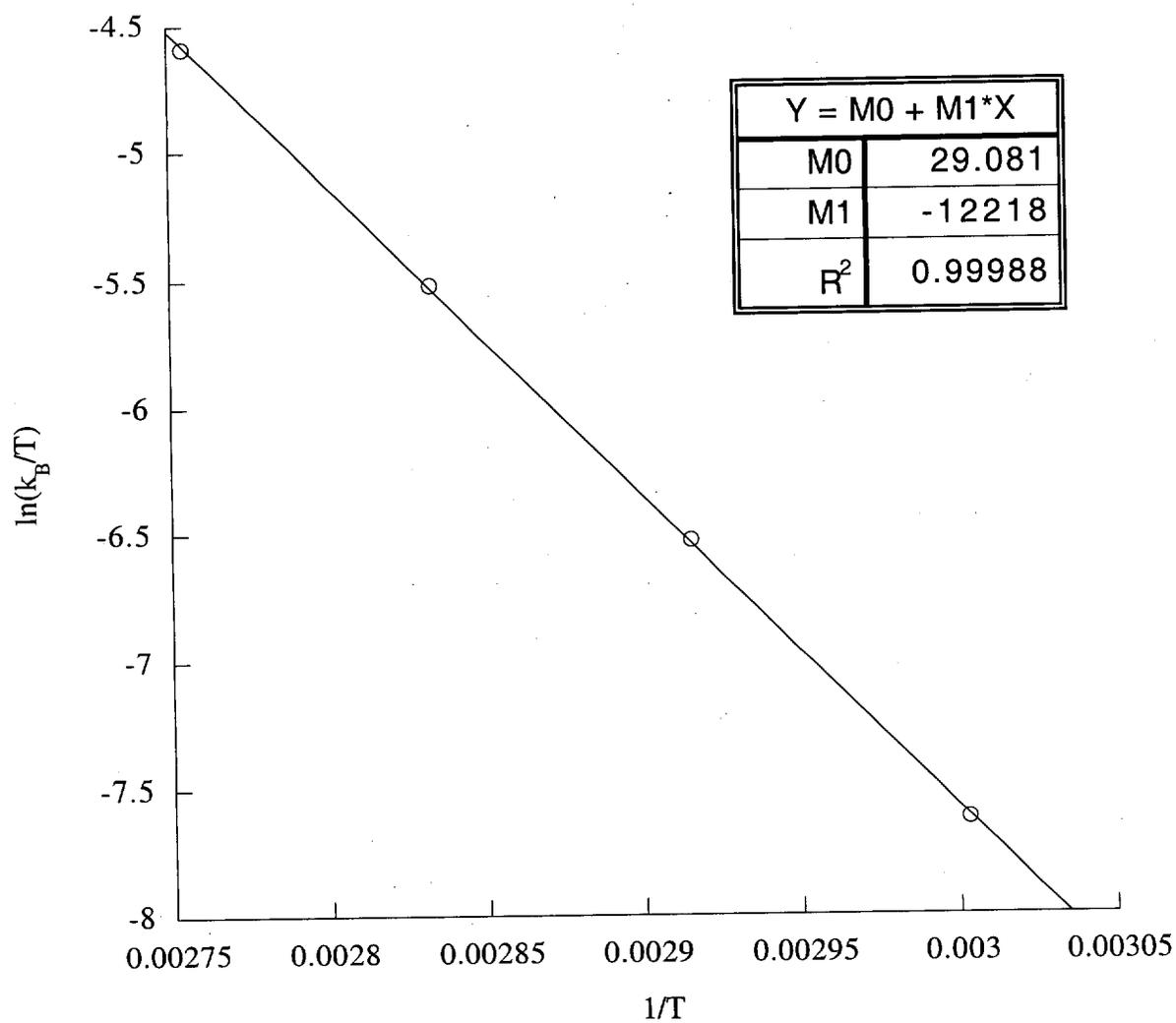
**Figure S3. Eyring Plot for Phosphine Exchange in Catalyst 3**



**Figure S4. Eyring Plot for Phosphine Exchange in Catalyst 5**



**Figure S5. Eyring Plot for Phosphine Exchange in Catalyst 7.**



**Figure S6. Eyring Plot for Phosphine Exchange in Catalyst 8.**

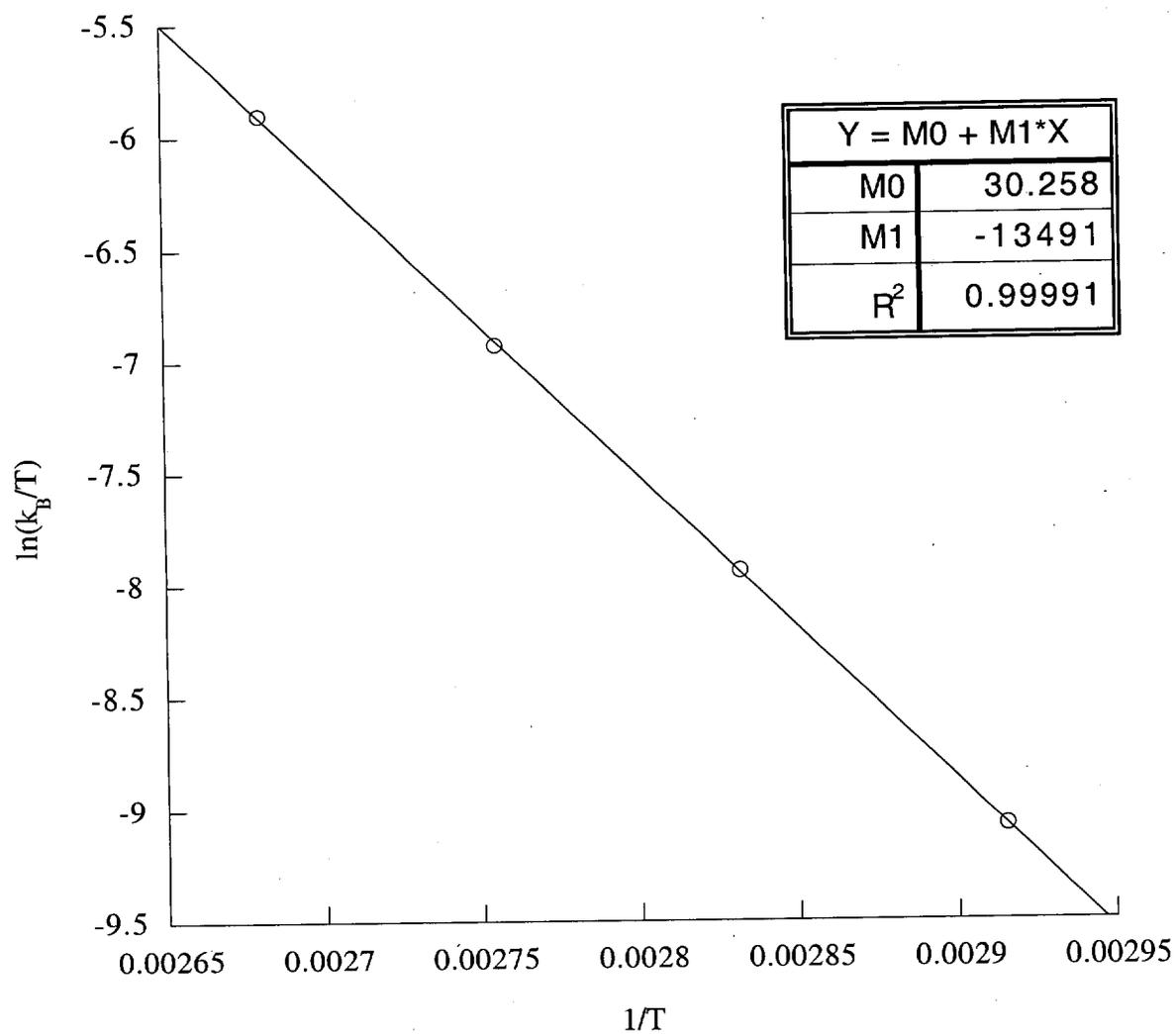


Figure S7. Eyring Plot for Phosphine Exchange in Catalyst 9.

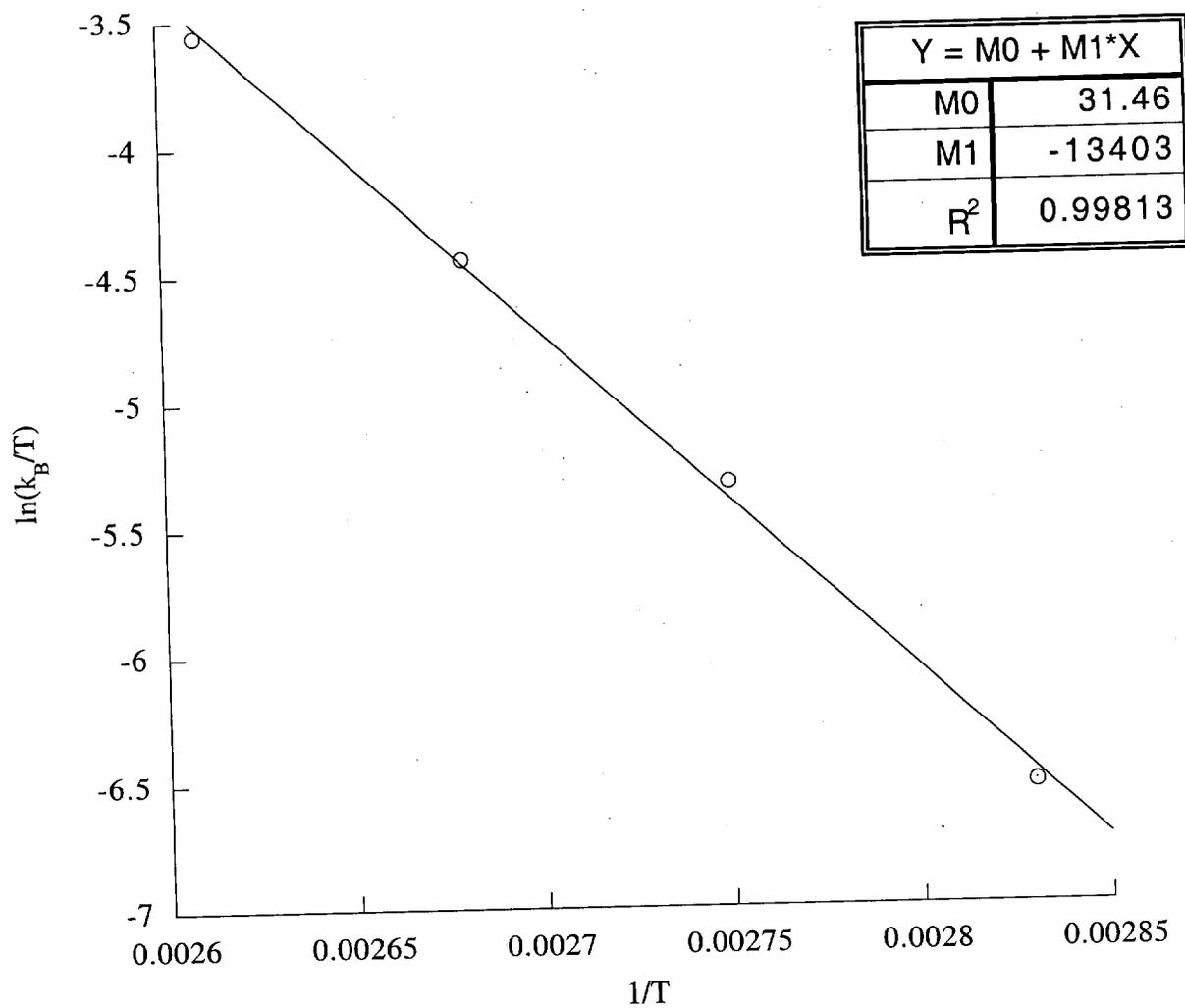
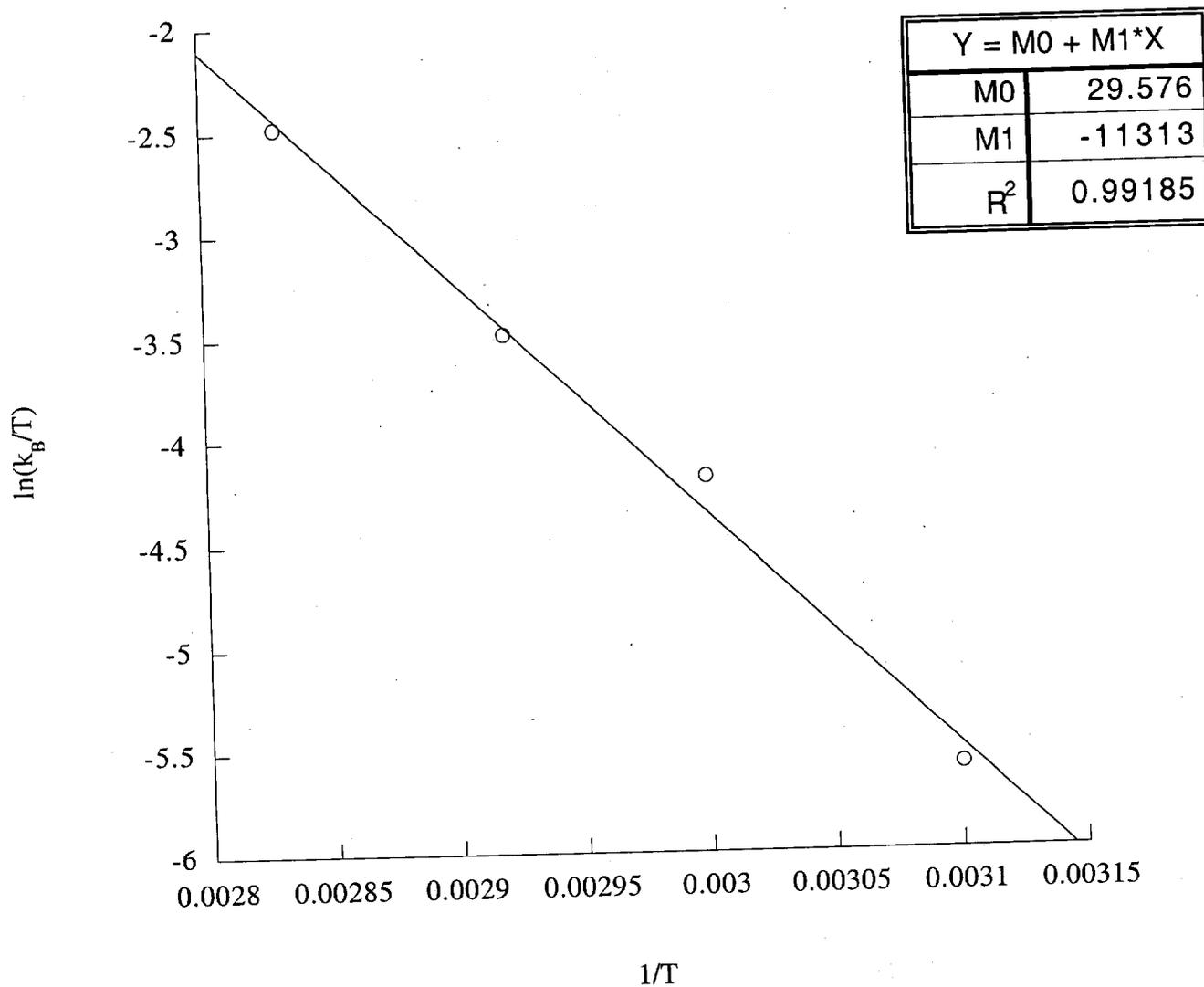
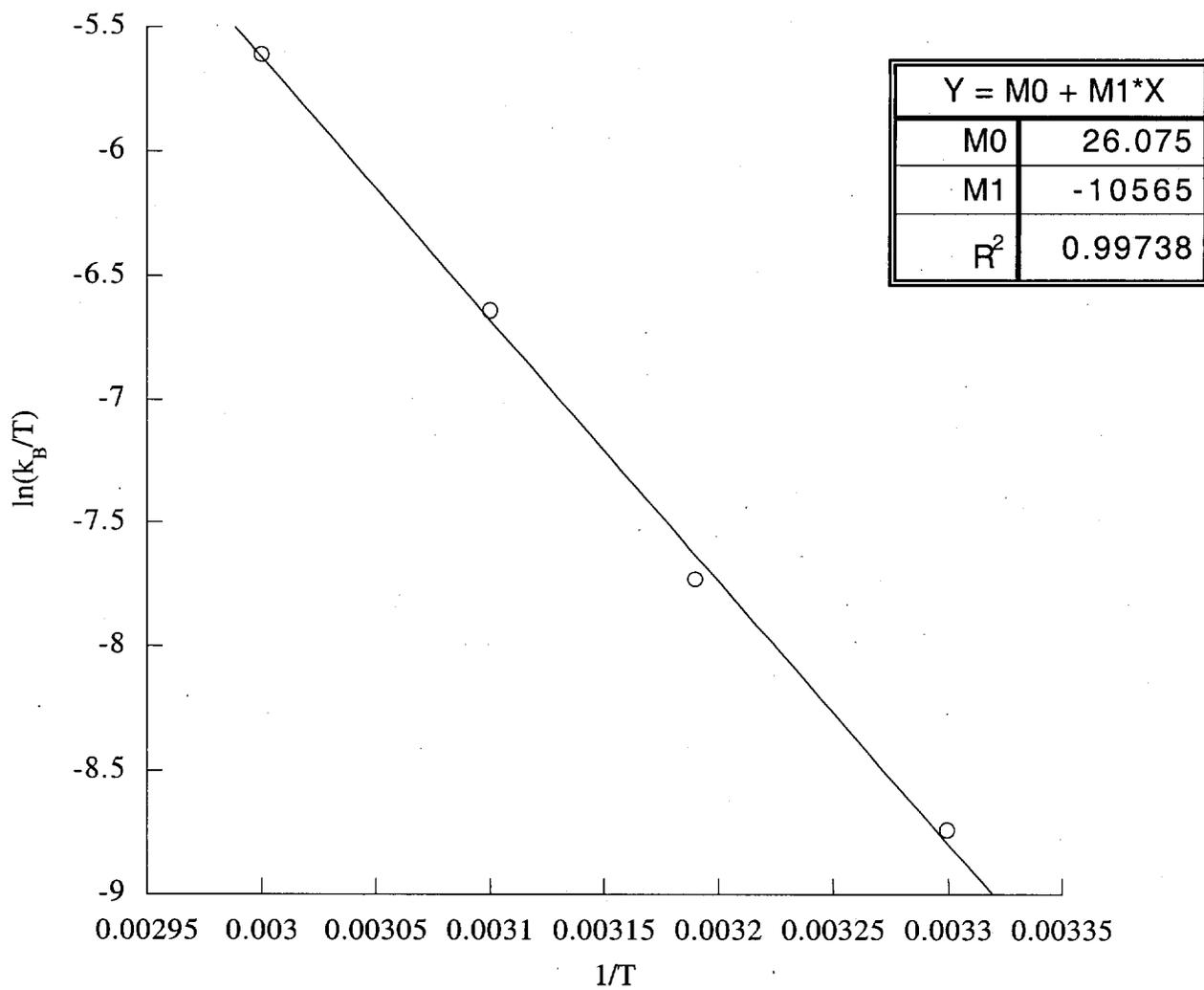
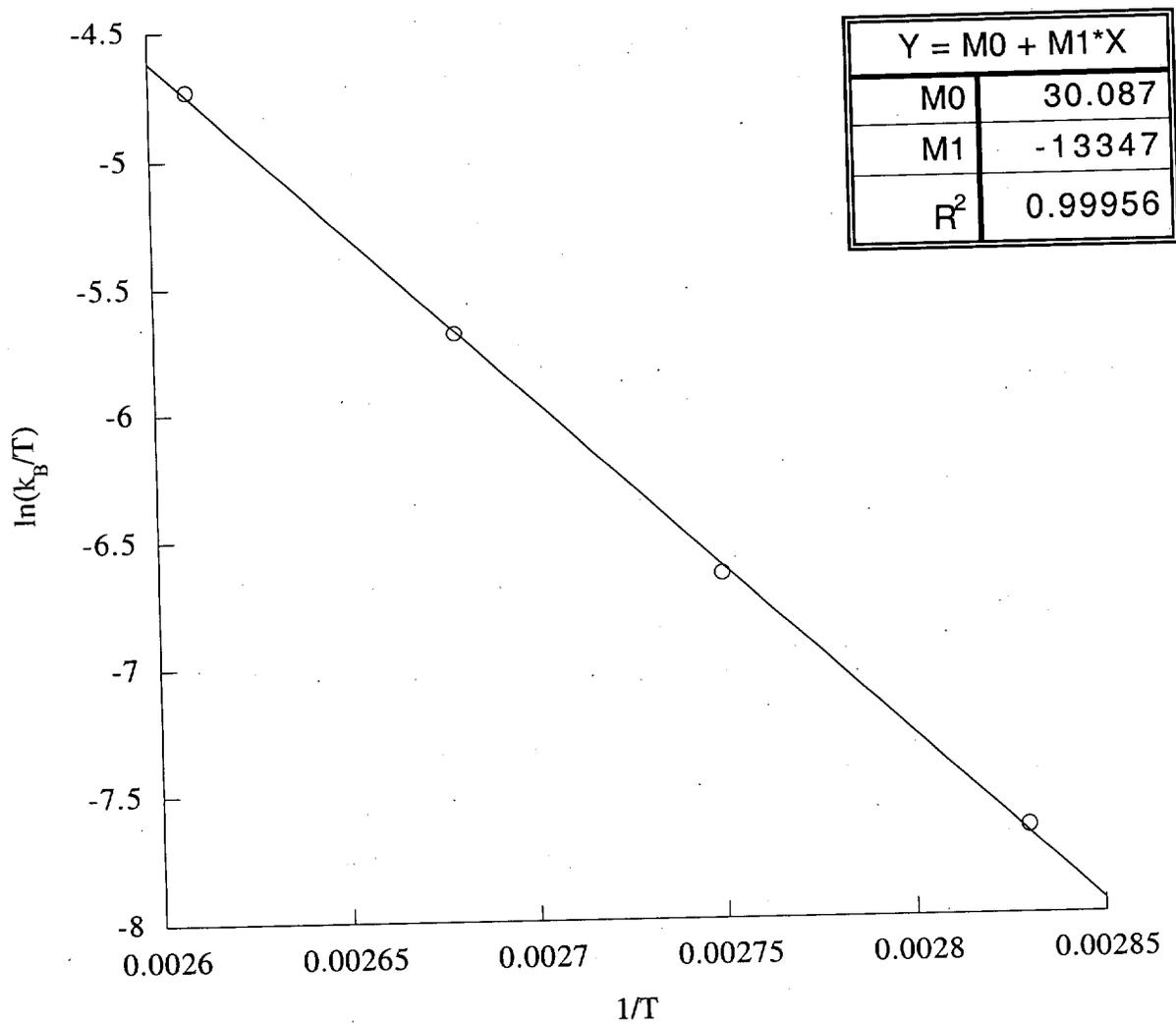


Figure S8. Eyring Plot for Phosphine Exchange in Catalyst 10



**Figure S9. Eyring Plot for Phosphine Exchange in Catalyst 11**

**Figure S10. Eyring Plot for Phosphine Exchange in Catalyst 12**



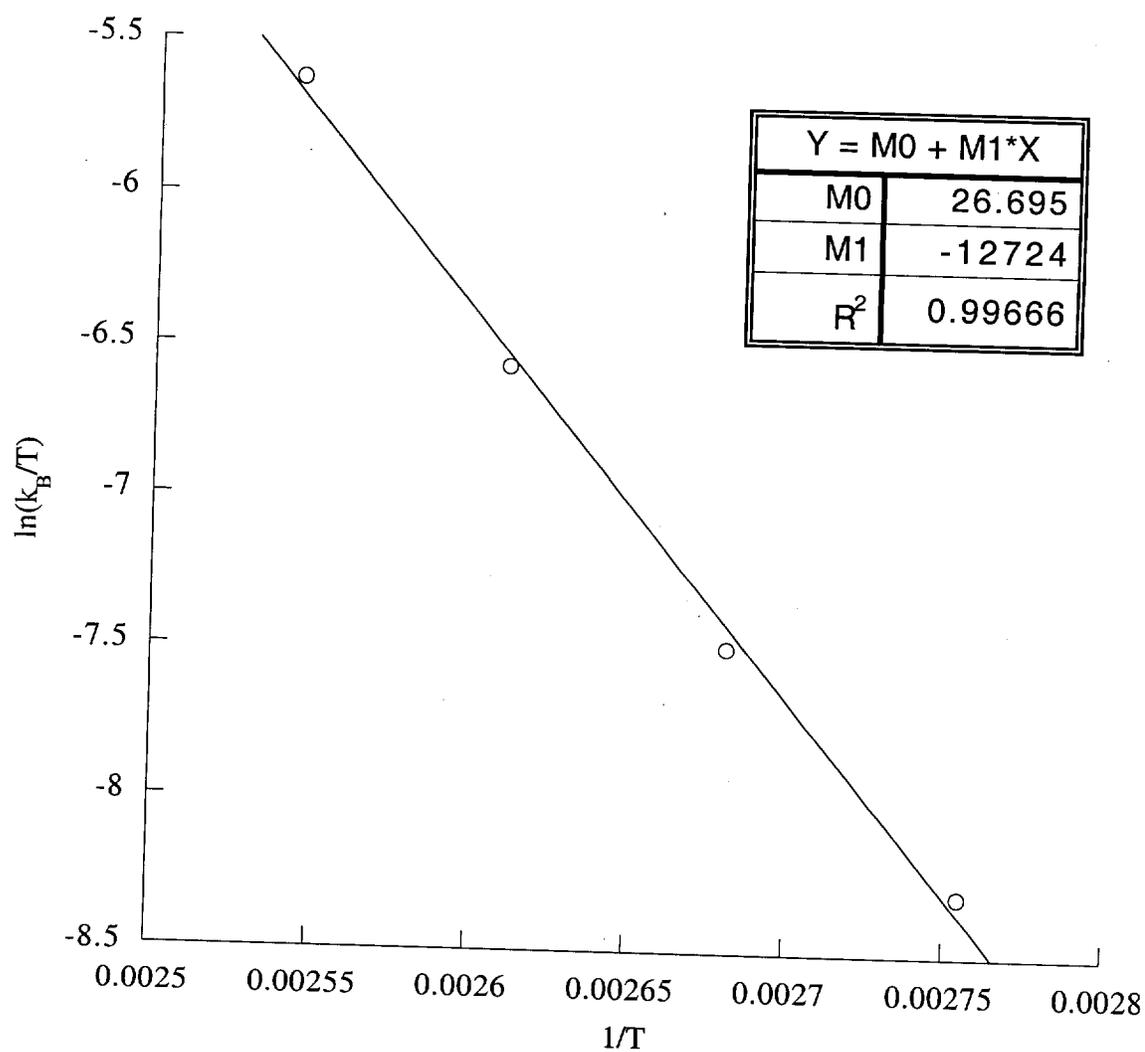
**Figure S11. Eyring Plot for Phosphine Exchange in Catalyst 14.**

Figure S12.  $k_{\text{Init}}$  versus [olefin] for Catalyst 2

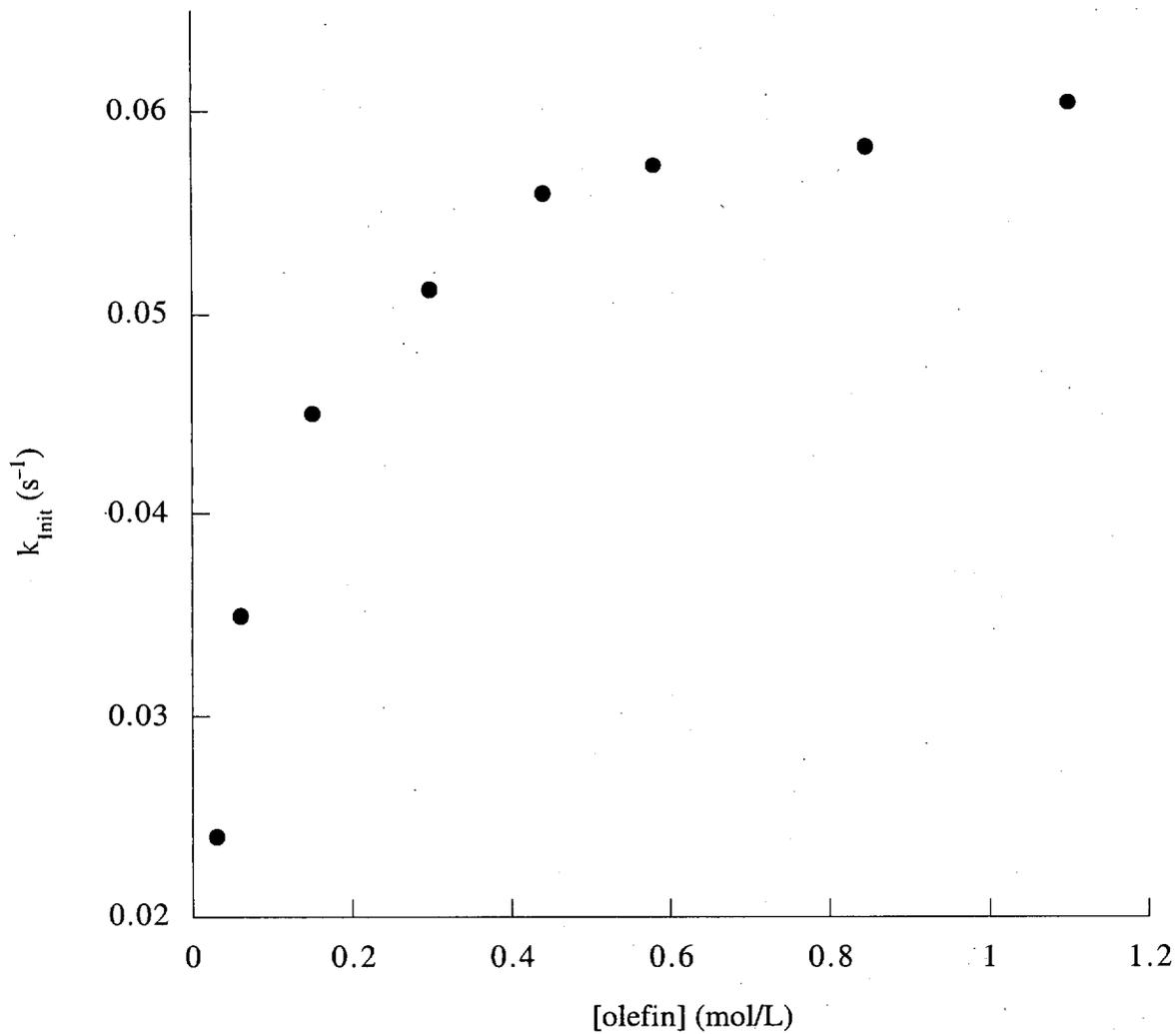


Figure S13.  $k_{\text{Init}}$  versus [olefin] for Catalyst 5.

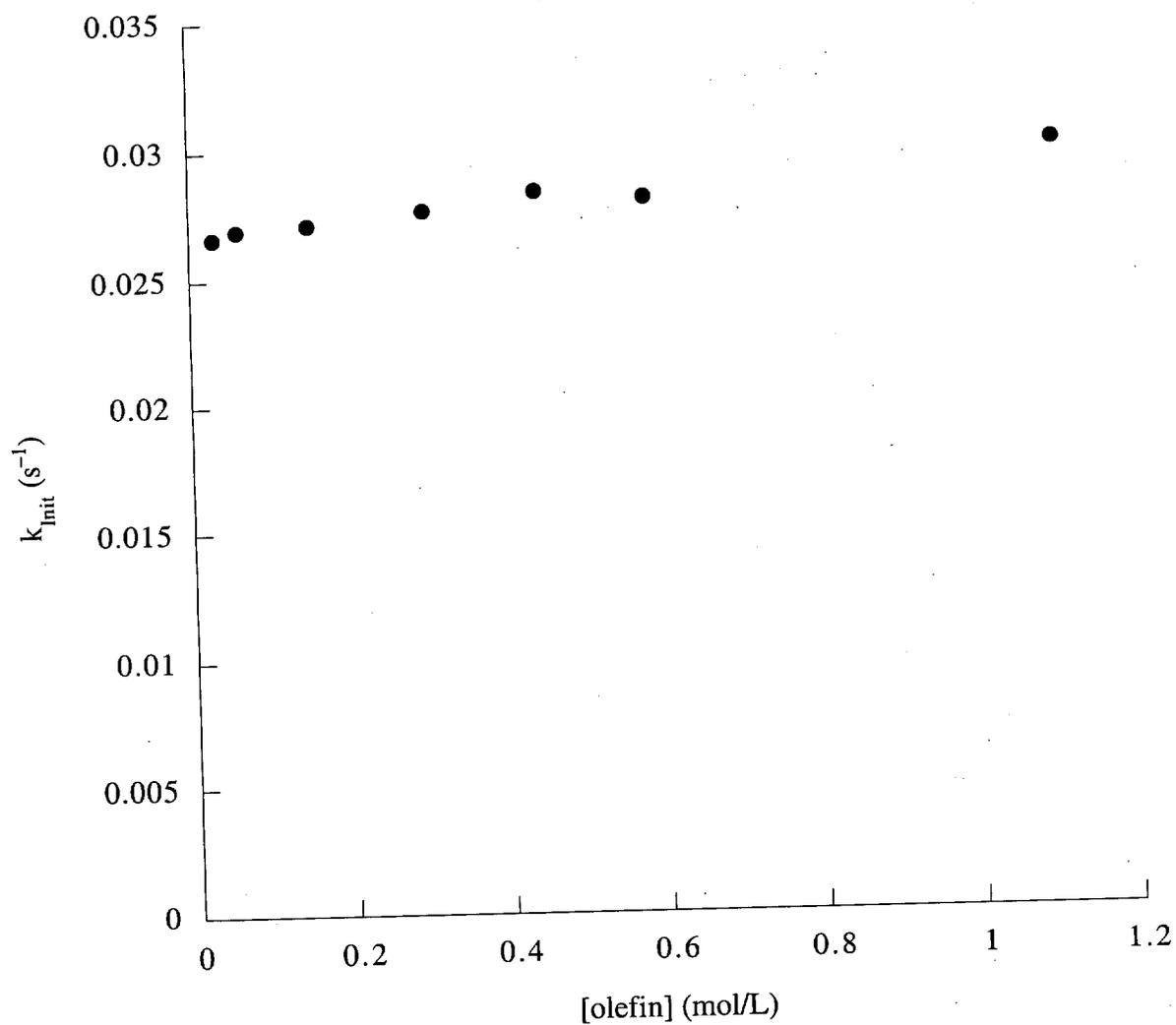
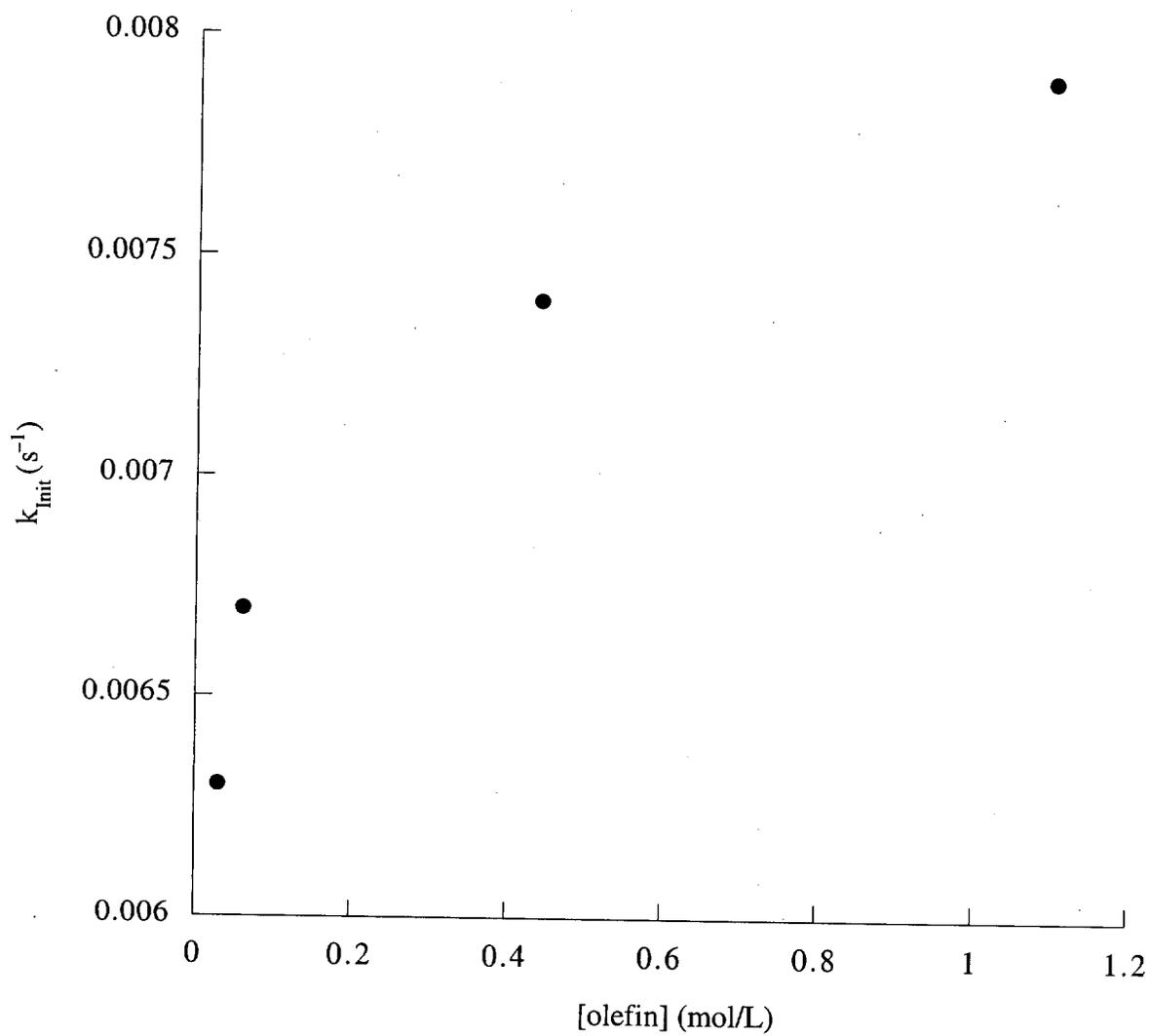


Figure S14.  $k_{\text{Init}}$  versus [olefin] for Catalyst 7.



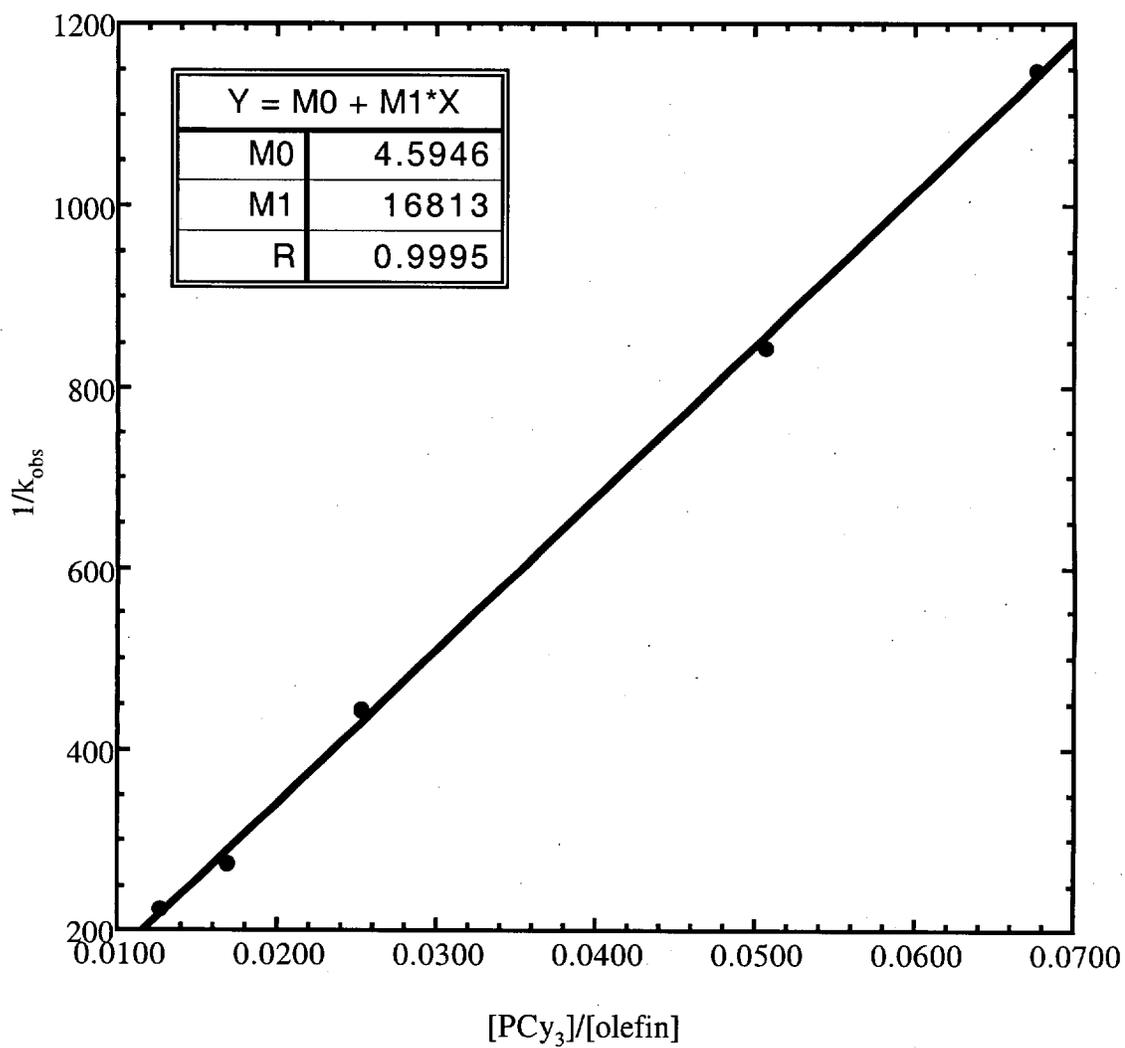
**Figure S15.  $1/k_{obs}$  versus  $[PCy_3]/[Olefin]$  for Catalyst 1**

Figure S16.  $1/k_{obs}$  versus  $[PCy_3]/[Olefin]$  for Catalyst 2

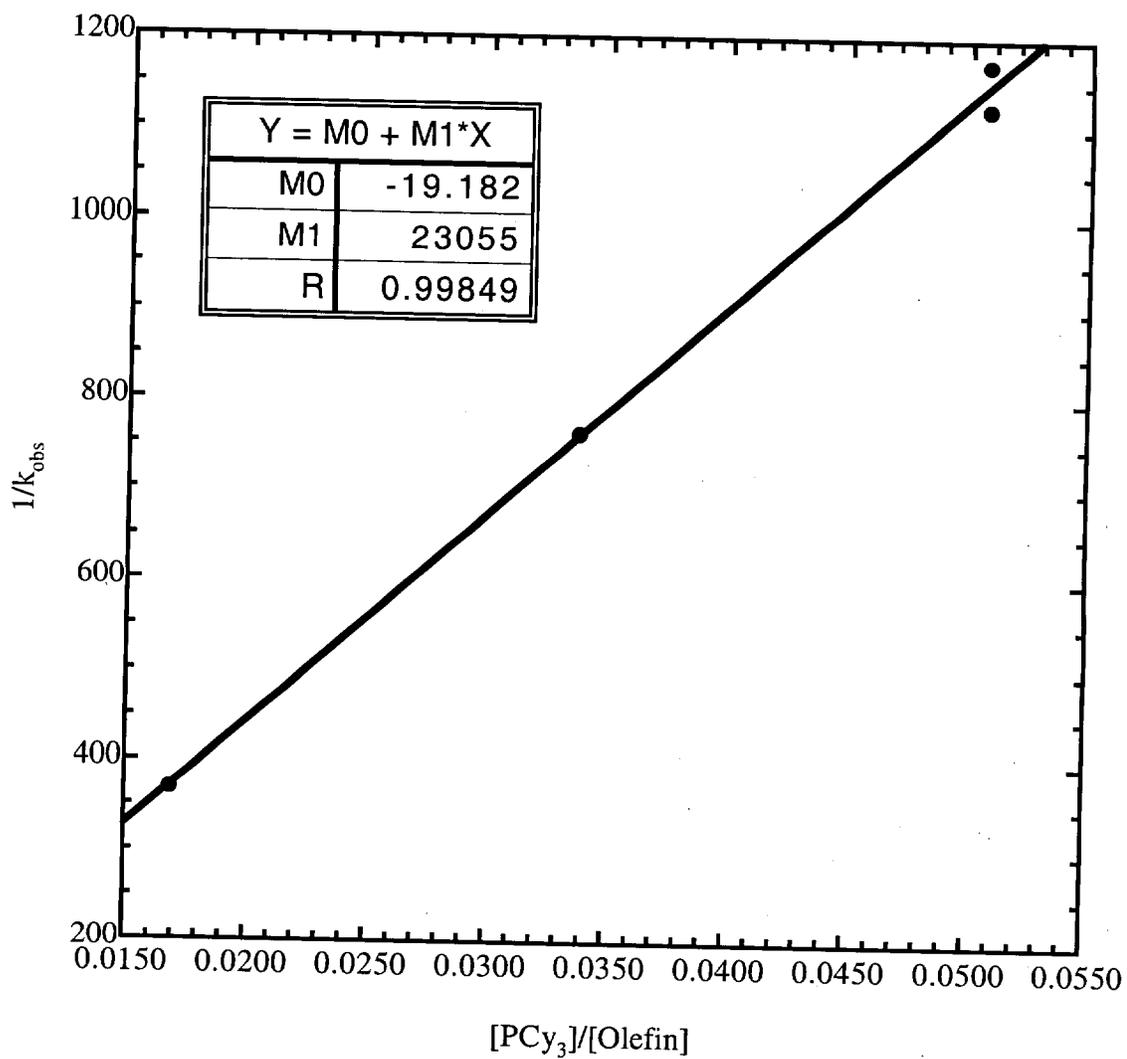


Figure S17.  $1/k_{obs}$  versus  $[PCy_3]/[Olefin]$  for Catalyst 3

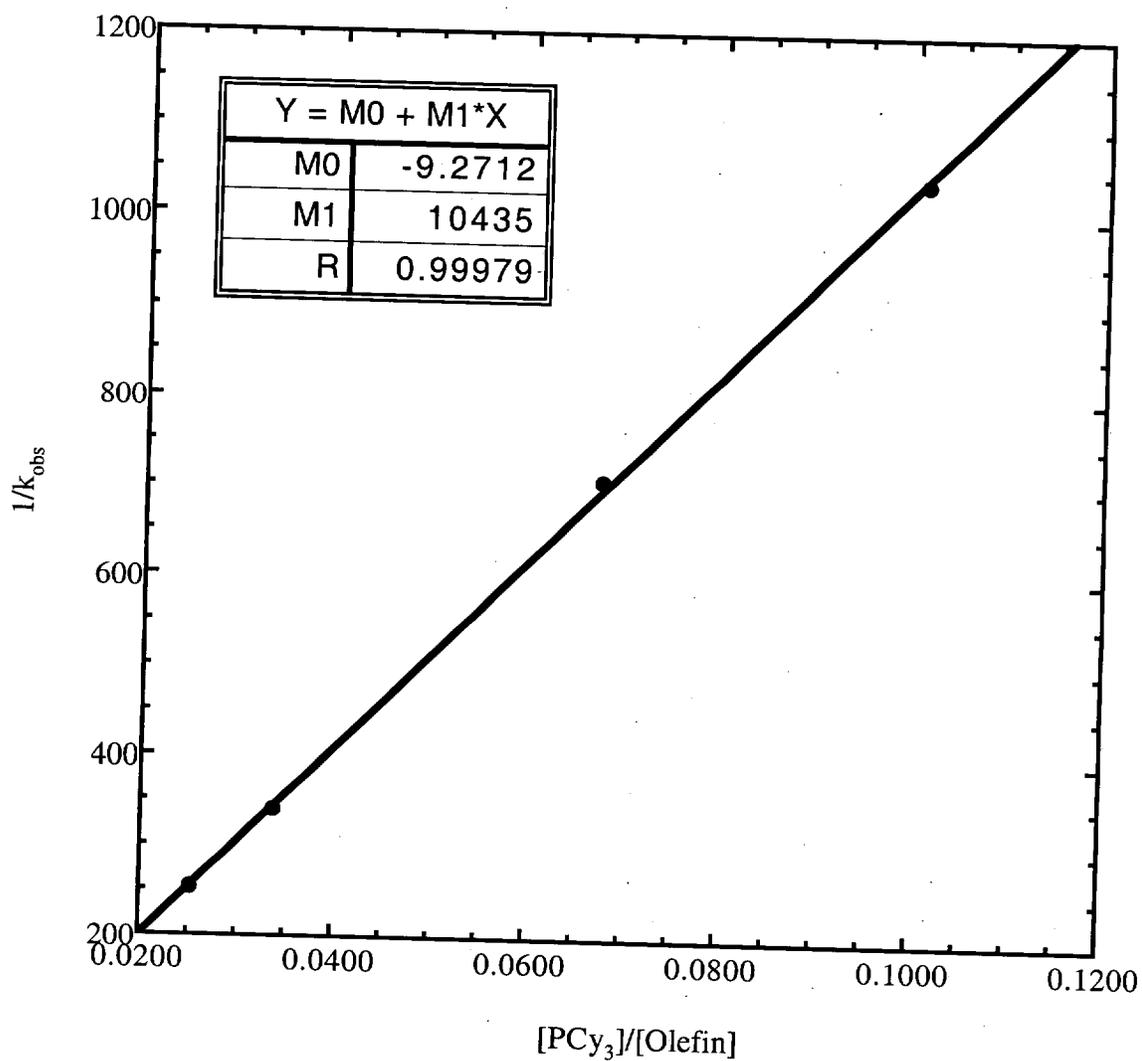


Figure S18.  $1/k_{obs}$  versus  $[PCy_3]/[Olefin]$  for Catalyst 6

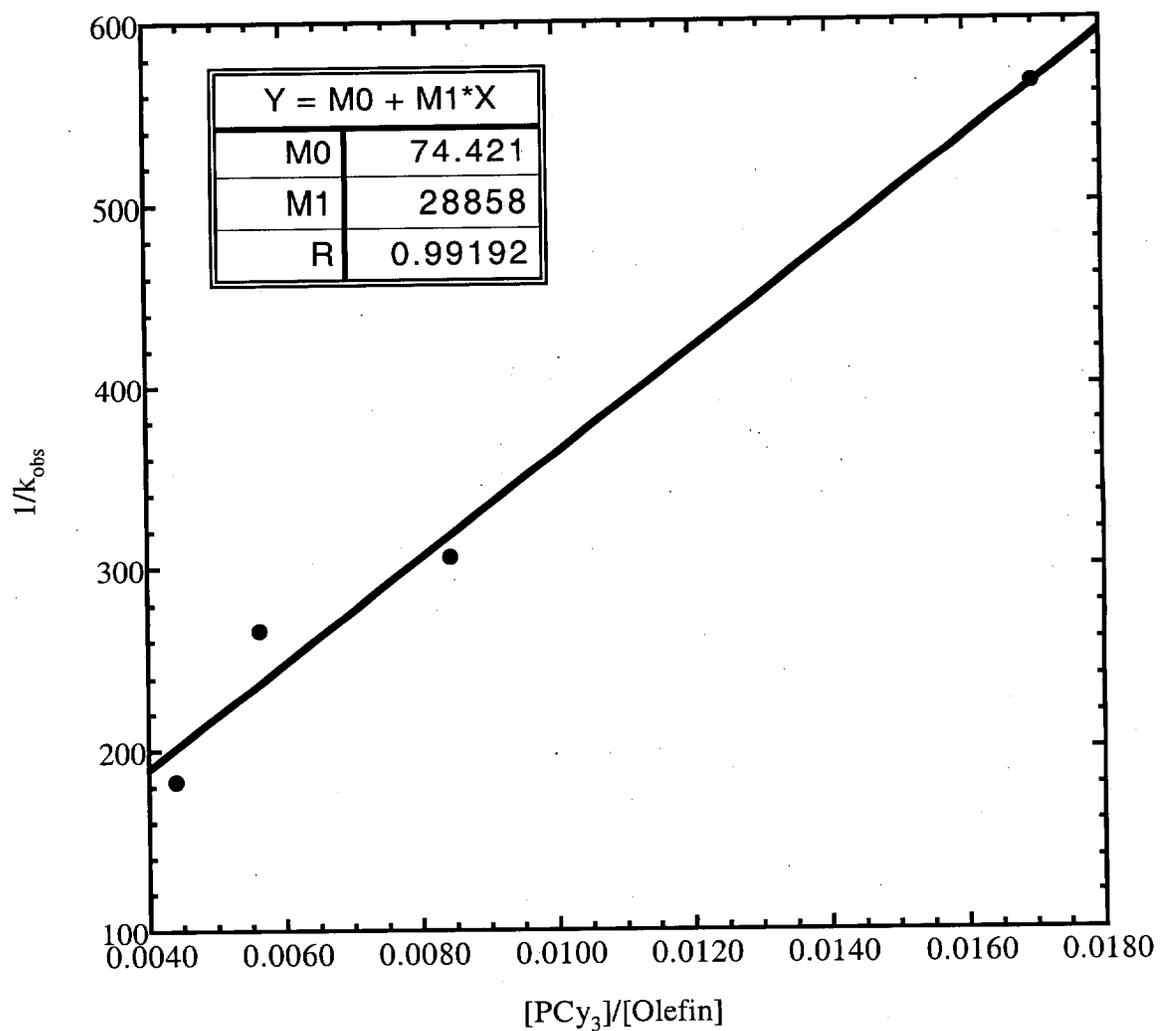


Figure S19.  $1/k_{obs}$  versus  $[PCy_3]/[Olefin]$  for Catalyst 8

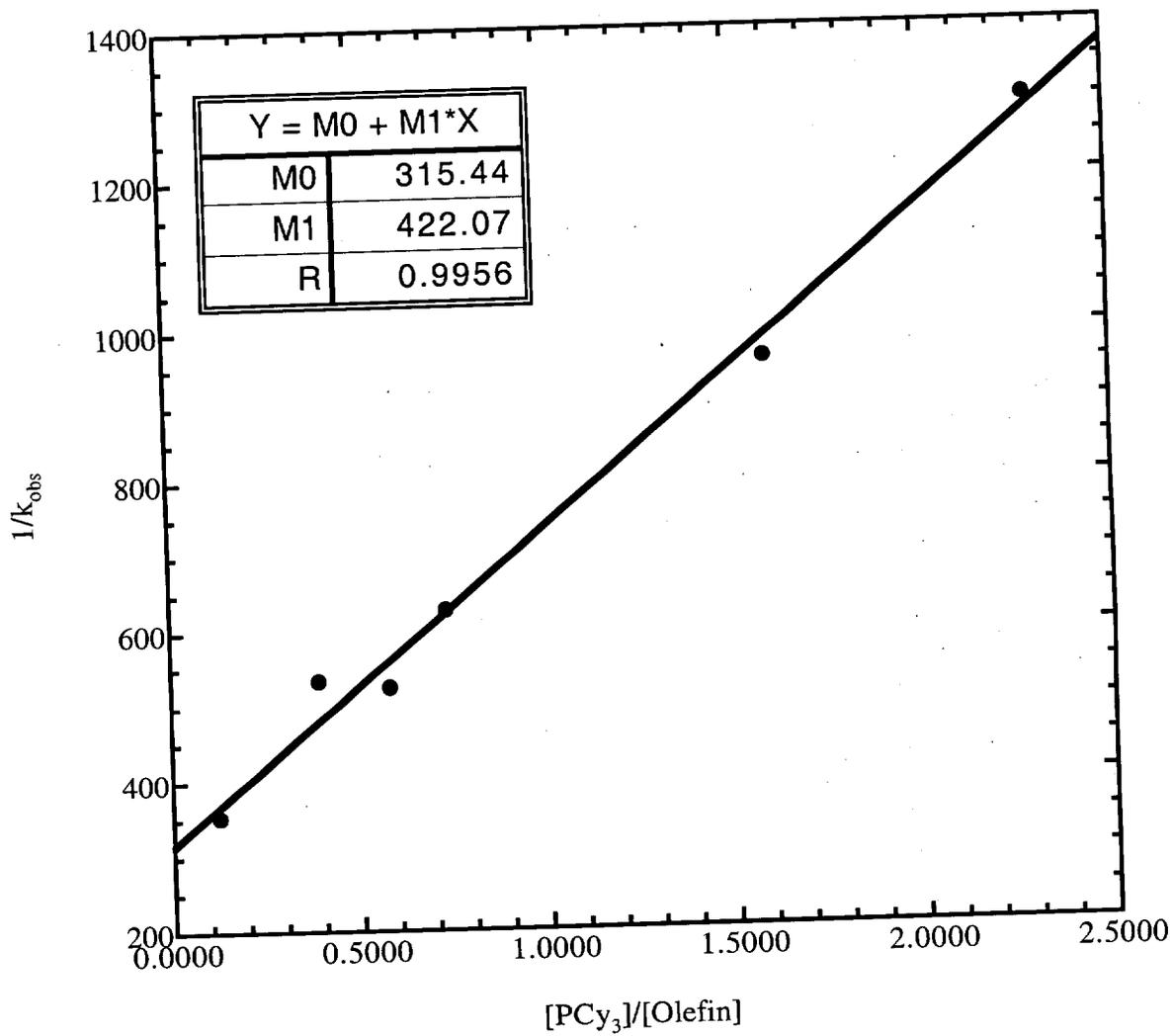


Figure S20.  $1/k_{obs}$  versus  $[PCy_3]/[olefin]$  for Catalyst 10

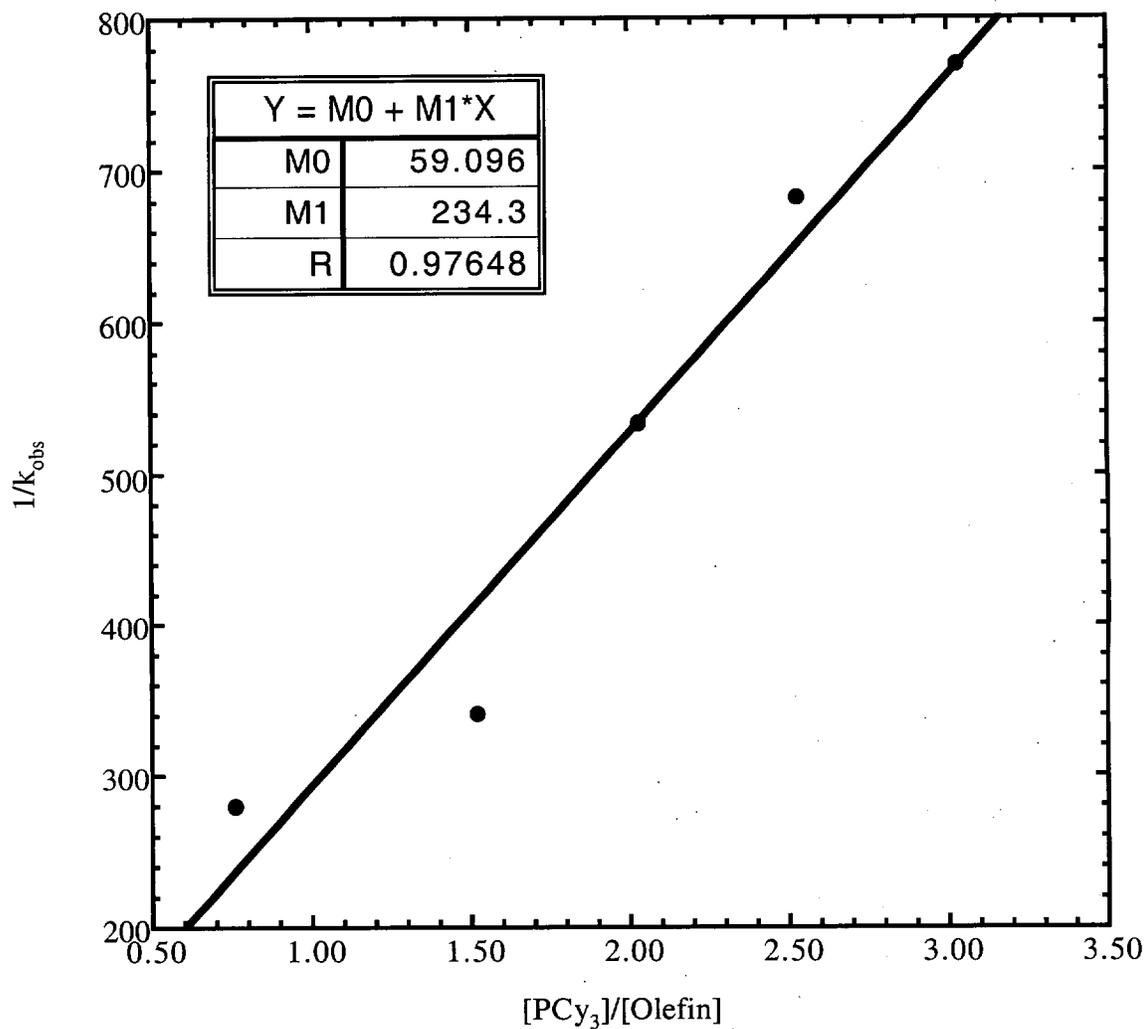


Figure S21.  $1/k_{obs}$  versus  $[PPh_3]/[Olefin]$  for Catalyst 11

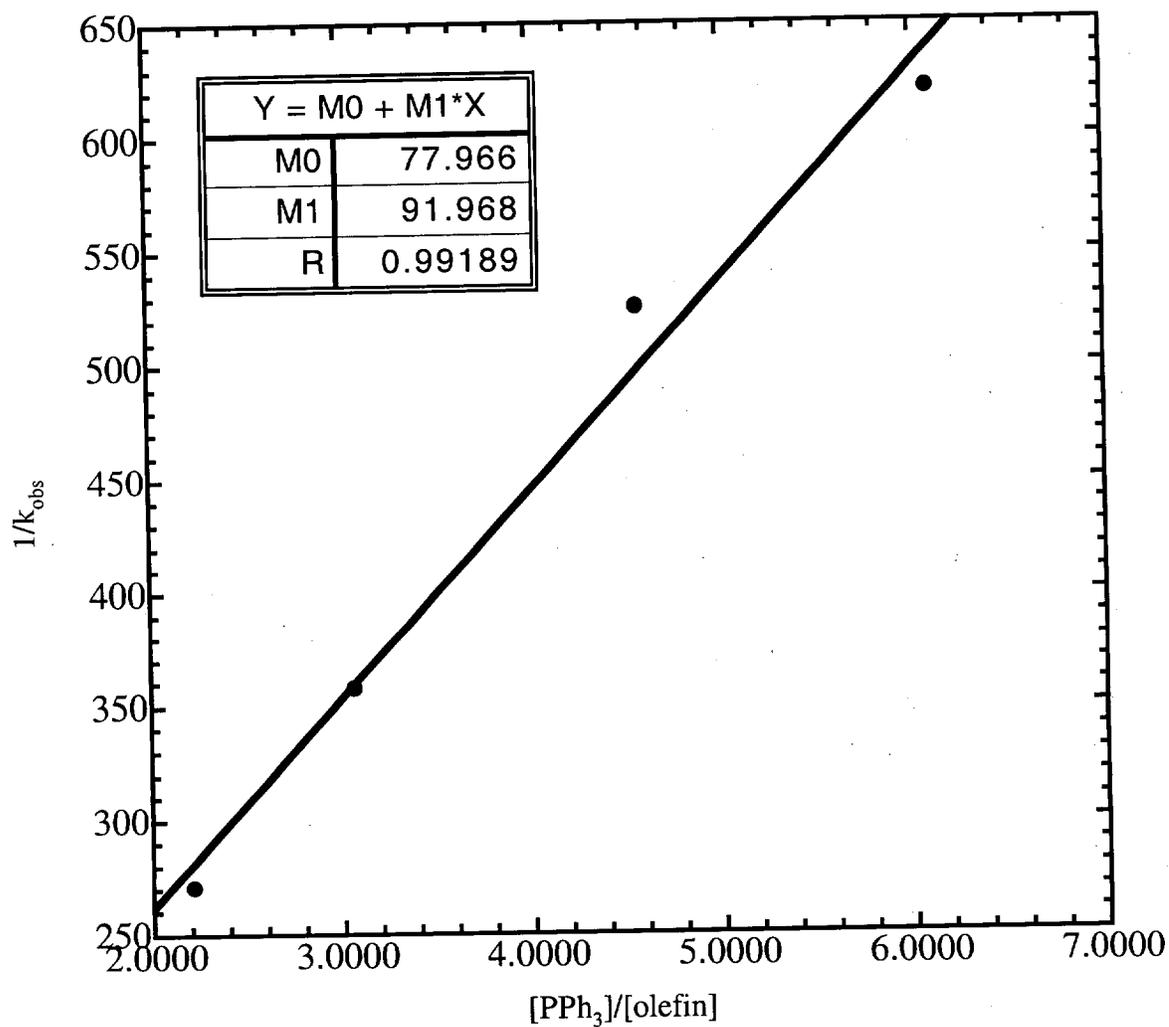


Figure S22.  $1/k_{\text{obs}}$  versus  $[\text{PBn}_3]/[\text{Olefin}]$  for Catalyst 12

