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# TARGETING POLYMER THERAPEUTICS TO BONE

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# Abstract

An aging population in the developing world has led to an increase in musculoskeletal diseases such as osteoporosis and bone metastases. Left untreated many bone diseases cause debilitating pain and in the case of cancer, death. Many potential drugs are effective in treating diseases but result in side effects preventing their efficacy in the clinic. Bone, however, provides an unique environment of inorganic solids, which can be exploited in order to effectively target drugs to diseased tissue. By integration of bone targeting moieties to drug-carrying water-soluble polymers, the payload to diseased area can be increased while side effects decreased. The realization of clinically relevant bone targeted polymer therapeutics depends on (1) understanding bone targeting moiety interactions, (2) development of controlled drug delivery systems, as well as (3) understanding drug interactions. The latter makes it possible to develop bone targeted synergistic drug delivery systems.

## Keywords

Bone-targeting; Drug delivery; Cathepsin K; HPMA copolymer; PLGA copolymer; Hydroxyapatite; Poly(ethylene glycol) (PEG); Prostaglandin; Statins; PTH1-34; Osteoporosis; Bone metastasis; Rheumatoid arthritis; Bone fracture

# 1. Introduction

Bone is a complex organ responsible for structure and calcium storage as well as being a structure for hematopoiesis to occur. Adult bone is 50-70% mineral, 20-40% organic matrix, 5-10% water and 1-5% lipids. Hydroxyapatite (HAp)  $(Ca_{10}(PO_4)_6(OH)_2)$  is the main mineral component of bone. The crystallinity of HAp increases with time, however, the needle-like crystals never grow much more than 300 Å in length [1–3]. These crystals are continuously being broken down and replaced during bone turnover. Initially, monocytes receive several signals pushing them to differentiate into osteoclasts. Critical to this process, osteoblasts present receptor activator of nuclear factor  $\kappa B$  (RANK) surface receptor triggering the TNF receptor associated factor 6 (TRAF6) cascade in monocytes, ensuring osteoclast development [4,5]. Mature osteoclasts then perform the catabolic process on bone. The osteoclast creates a

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sealing zone (Fig. 1) over resorption sites (lacunae) and proceeds to release cathepsin K (Cat K) to degrade the organic matrix and HCl to dissolve the HAp. These mechanisms in combination with reactive oxygen species (ROS) produced by tartrate-resistant acid phosphatase (TRAP) degrade the bone preparatory for osteoblasts to lay down organic matrix called osteoid. The osteoid, primarily composed of type I collagen, is then calcified and becomes new bone [6,7].

When catabolism or anabolism in bone turnover is altered, the bone may become diseased. Osteoporosis, where catabolism effectively outweighs anabolism, predominately affects men and women over 50. Approximately 44 million people in the US suffer from low bone mass and 10 million people already have osteoporosis. An estimated 61 million will have the disease by 2020 [8]. Complications of this disease include debilitating vertebral and hip fractures cost between in \$13.7 billion and \$20.3 billion in 2005 [9]. Other imbalances in bone turnover include osteosarcoma and bone metastasis. Osteolytic and tumor creating cancers of the bone both result in significant pain for patients and morbidity. Osteoarthritis, osteomyelitis, infections, and fractures all increase the economic toll on health systems worldwide as well as decrease quality of life for millions of people each year.

Nanomedicine plays an important role in recent advancements in bone therapeutics. Diseases that affect the bone may carry a range of receptor-specific targeting opportunities or they may carry none at all. However, by exploiting the mineral portion of the bone, bone diseases can be targeted using molecules such as bisphosphonate and acidic oligopeptides. Nanomedicines can be easily conjugated to several ionic targeting ligands, increasing binding avidity to bone. HAp crystal structures or HAp exposure are disease specific, and as such, modification of the targeting ligand itself or the density of the ligand may provide specific targeting opportunities. In addition to targeting, nanomedicines create an opportunity to image and synergistically combine multiple drugs to treat the diseased portion of the bone (Fig. 2).

Aspects of this field have been reviewed previously [10-15]. In this review we discuss the selection of proper targeting agents, polymers, and drugs for application to specific bonedisease states. The analysis also includes methods of increasing local drug accumulation by modification of the molecular weight and drug-polymer release mechanisms. We will also elaborate on how various bone anabolic agents interact as well as how synergism of chemotherapeutics can be achieved by nanomedicine.

# 2. Targeting

Each bone disease state has similarities in that each causes local inflammation and/or results in the exposure of HAp to blood. These two characteristics can be utilized in order to deliver high drug loads specifically to diseased tissue. Inflamed tissue fills with exudate as various cytokines dilate vasculature. Macromolecular therapeutics then are able to travel into the tissue due to vascular pressure and be retained in the tissue due to lack of diffusivity back into the vasculature. This enhanced permeability and retention (EPR) has been well documented in several publications and is extensively discussed in several reviews [16-19]. Delivery of therapeutics to bone also depends on the extravasation through vessels in or near the bone [20]. Larger molecules, for example albumin and polyvinylpyrrolidone (Mw 35 kDa), can extravasate and have been found in bone after i.v. administration [21]. In fact, it has been long known that the vasculature in bone is fenestrated with pore sizes up to 80 nm [22], which exceeds the hydrodynamic size of most circulating nanomedicines. Exposure of HAp to blood carries its own targeting possibilities. Various tetracyclines, bisphosphonates, acidic oligopeptides, chelating compounds and salivary proteins have all been employed to target bone diseases. These compounds bind to the inorganic HAp part of bone and have

specificity for a certain size of HAp crystal. Therefore, as the crystallinity of newly deposited bone in a tumor is different from the crystallinity of HAp exposed during a bone fracture, so targeting molecules can be selected for specific disease states.

# 2.1 Tetracycline

Tetracycline was introduced as an antibiotic in 1947 [23]. The antibiotic is derived from *Streptomyces rimosus* and inhibits aminoacyl-tRNA from entering into bacterial ribosomes, inhibiting protein elongation [24]. Tetracycline affects general bacterial metabolism and is prescribed for a wide range of gram-positive and gram-negative bacteria. Soon after its incorporation into medicine, however, tetracycline was found to bind strongly to bone. Its usage was discontinued in pediatric medicine as the high affinity to HAp caused children's teeth to stain yellow [25]. Furthermore, other early studies found that tetracycline may inhibit skeletal growth in children [26,27]. Its use as an antibiotic in adults and a targeting ligand in bone disease though, has not persisted.

Early studies indicate that tetracycline must be in the correct orientation in order to bind HAp. Oxygens bound to C2, C10, and C12 are among the primary HAp binding atoms [28]. As such, modifications around the 5, 6, and 7 carbons can be made with minimal effects on binding and biological activity (Fig. 3). Other studies have shown that bone binding and antimicrobial activity may be retained even with a simplified tetracycline molecule. Neale et al. tried to reduce potential side effects caused by tetracycline's biological activity by minimalizing tetracycline structure so that it retains no biological activity yet is still able to bind to HAp. They found that 3-amino-2,6-dihydroxybenzamide retains 50% of the ability to bind to HAp when compared to native tetracycline (Fig. 3). To achieve bone anabolism, the new targeting ligand was bound to estradiol via a succinate linker. Following conjugation, the compound had a binding affinity of 105% over tetracycline alone. As estradiol alone did not bind to HAp the increased affinity may be attributed to the addition of a succinate linker [29].

#### 2.2 Bisphosphonates

Bisphosphonates (BPs) also bind strongly to HAp and retain much of the binding affinity after conjugation to other molecules or carriers. The first biological activity associated with BPs was noticed in 1968 [30]. Later, their effects on bone metabolism were discovered and they were prescribed to treat osteoporosis. BPs are derived from pyrophosphate and contain a carbon substitution for pyrophosphate's central oxygen. From the central carbon most BPs have been functionalized with a hydroxyl group and an R group by which they are classified (R groups containing nitrogen and those that do not) (Fig. 3). Non-nitrogen containing BPs incorporate into AMP creating a modified ATP, which cannot be hydrolyzed. Buildup of this ATP analog inside osteoclasts leads to apoptosis and reduced bone turnover. Nitrogen containing BPs (nBP) are much more potent. They inhibit farnesyl pyrophosphate synthase activity and by so doing disrupt the mevalonic acid pathway. Without the mevalonic acid pathway, protein prenylation is inhibited and osteoclasts lose their functionality or apoptose [31]. This inhibition may not completely arrest bone turnover as mevalonic acid pathway inhibitors increase bone morphogenic protein production as well [32-34]. Clinically, the activity of BPs has a profound effect on osteoporosis. Clinical data demonstrates that BPs vield a 47% reduction in vertebral fractures as well as wrist fractures, 50% in hip fractures, and a 19% overall drop in all non-vertebral fractures [35].

BP affinity to raw HAp is also clinically relevant in bone scintigraphy. Tc99 labeled methylene diphosphate (MDP) or hydroxy methylene diphosphate (HMDP) are used clinically for imaging bone abnormalities ranging from osteomyelitis to stress fractures. In many bone diseases bone is deposited (as in a tumor) or degraded (as in osteoporosis and

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osteoarthritis). Consequently, HAp crystals are exposed and can be targeted using bisphosphonates such as MDP (Fig. 4).

The use of BPs as polymer/nanoparticle targeting moieties has several advantages. nBPs contain a primary amine and can therefore readily be conjugated to carboxylic acids. Secondly, as with tetracycline, if they are conjugated to nanomedicines via a degradable linker such as a pH sensitive hydrazone bond, they are pharmacologically active and may produce synergistic effects when coupled with appropriate drugs. Finally, the patent protection of many BPs will expire soon, which makes them an economic option for a targeting moiety.

BPs are one of the most studied bone targeting molecules and much research has been done determining their binding properties to HAp [37-39]. Hrubý et al. determined that the number of targeting moieties on a polymer did not dramatically affect the rate of adsorption. Rather, the amount of adsorbed polymer was affected by the number of targeting moieties as more ligands yielded higher binding avidity. The authors also concluded that there is a need for multiple targeting moieties to ensure that at least one is exposed [40]. Logically then, multiple targeting ligands become more critical as the nanomedicine size increases [41].

BP adsorption to HAp is a thermodynamic process primarily driven by entropy. Mukherjee et al. measured the thermodynamics of adsorption for several clinically used BPs as well as modified BPs in order to determine what aspects of the molecules are important for HAp binding. For instance, strongly basic side chain substituents, such as the nitrogen in alendronate, contribute dramatically to the binding in the free form. Furthermore, when measuring the difference between one Pi bond forming with HAp versus two Pi bonds forming (one phosphate versus both phosphates binding) they found that the hydroxyl group originating from the central carbon atom is necessary for both phosphates to bind [42]. By reducing a nanomedicine's thermodynamic strength of adsorption, either by manipulating the number of ligands or by modification of the ligand itself, one can target only the thermodynamically favored diseased area over the kinetically favored skeleton.

Numerous BP conjugates with natural or synthetic macromolecules demonstrate bone targeting *in vivo* [43-45]. Targeted osteoprotegerin (OPG) is one example. OPG is naturally produced by osteoblast's Wnt- $\beta$ -catenin signaling pathway and is a potent RANKL inhibitor preventing osteoclastogenesis. With the intent to treat osteoarthritis, OPG was modified with a thiol-BP and administered intravenously (i.v.) to rats. Overall bone deposition of BP targeted OPG was twice that of free OPG. In an osteoarthritis rat model targeted OPG accumulated in the tibia 4x and femur 6x that of free OPG, demonstrating BP targeting abilities [46].

BPs exhibit anti-angiogenic properties by causing apoptosis, inhibiting migration, and reducing angiogenic sprouts of endothelial tissue making it a good antineoplastic agent but reducing capabilities to heal bones [47]. Moreover, the half-life of BPs such as alendronate in the bone is more than 10 years [48]. The inhibition of osteoclasts for extended periods of time can have serious detrimental effects on bone turnover as osteoclasts secrete insulin-like growth factors and BMPs to promote the maturation of osteons to osteoblasts. Jawbones have a high blood supply and high turnover resulting in BP accumulation. The combination of low bone turnover and lower blood flow causes osteonecrosis of the jaw (ONJ). ONJ is primarily associated, but not limited, to potent BPs used for chemotherapy, which completely inhibit osteoclast function [48,49]. Other toxicities include nephrotoxicity, hypocalcemia, and ocular dysfunction [50]. Therefore, BPs should not be used without considering some of the side effects.

# 2.3 Acidic oligopeptides (AO)

Bone sialoprotein is one of several naturally occurring proteins that exhibit strong affinity for HAp. Sialoprotein has several strings of acidic amino acids. Modeled after sialoprotein, glutamic acid and aspartic acid oligopeptides 4-10 amino acids (AA) long provide a more biocompatible option when adequate blood flow and bone turnover are needed [51]. Biocompatibility can be further increased as D peptides, which are not easily recognized by the body's immune system, can replace D peptides. Similarly to BPs, the HAp binding capabilities of these oligopeptides are retained after conjugation to a nanomedicine carrier via the peptide's alpha amino group [52].

AO's chain length can be modified to fit an application. Sekido et al. compared binding rates of  $Asp_{2-10}$  and  $Glu_{2-10}$  to HAp (Table 1). Increase in AA chain length resulted in enhanced binding rate ( $B_{max}$ ) up to hexapeptides; then the rate of binding plateaued and further increase in chain length was without effect. The dissociation constant ( $K_d$ ) decreased with increasing numbers of AA throughout the experiment and no differences were found between  $_D$  and  $_L$  AA. This indicates that an AO over 6 AA long is ideal for optimal binding efficiencies [53]. Nanomedicines, however, are not limited to a single targeting moiety; rather multiple ligands may also decrease  $K_d$ . Ouyang et al. demonstrated the effect of multiple sets of AOs on binding rates. Dendrimers were synthesized with 1, 2 and 3 chains of Asp<sub>4-6</sub> and a HAp binding of one alone but that the addition of a third chain resulted in poorer binding than two chains alone [52]. In the case of dendrimers, this means that steric hindrance may prevent binding of multiple AO chains. The extra chains then decrease the binding by increasing the size of the dendrimer itself or by reducing the ionic attraction of other dendrimers to the HAp by increasing the negative charge on the surface.

#### 2.4 Estradiol analogs

Estrogen replacement therapy is being used for the treatment of osteoporosis related to low estrogen levels. Crooks and coworkers developed estradiol analogs that would localize in bone tissue but are lacking estrogenic properties. They designed bone-targeting compounds by attaching calcium chelators to an estradiol moiety through succinoyl [54] or carboxyethyl [55] linkers. To further improve the targeting potential they prepared a series of phosphate esters of the carboxyethyl linker-containing compound that possessed similar bone-targeting properties as tetracycline [56].

#### 2.5 Specific applications

Reflecting their molecular structures, tetracycline, BPs, and AO do have some unique binding capabilities, which may affect the disease states for which they are most appropriate. The rate of binding of AOs to HAp is faster than that of BP [57]. One may hypothesize this to be a result of the larger binding area of AOs. Meanwhile, BPs have a greater binding strength, most likely due to greater specificity of BPs for HAp [57]. Wang et al. found that BP binds to all bone while Asp<sub>8</sub> binds preferentially to higher crystalline HAp and thus to resorption surfaces [58]. In a follow-up study, AFM cantilever tips were modified with either Asp<sub>8</sub> or alendronate (ALN). Interactions with HAp such as binding, adhesion events and rupture forces were measured (Fig. 5). ALN binding at zero distance was 190 pN as opposed to 105 pN with Asp<sub>8</sub>. Furthermore, ALN had higher rupture forces [3]. There were two sets of rupture forces for ALN, which lends itself to the bonding mechanism described by Mukherjee et al. [42]. Using two HA surfaces with differing crystallinity, Wang et al. confirmed that ALN is less affected by crystallinity and that Asp prefers higher crystalline HAp [3]. Bringing targeting via ionic interactions full circle, Miller et al. demonstrated using fluorescence microscopy that tetracycline deposits preferentially onto growing surfaces, or those with low crystallinity (Fig. 6) [59]. Application of these principles means

that targeting osteoclasts should be done with AOs, which prefer absorbing surfaces, while targeting osteoblast should be done with tetracycline, which prefers growing surfaces of bone.

# 3. Polymer carriers

The conjugation of multiple targeting ligands to a polymeric backbone is an obvious advantage that polymeric drug delivery systems have over small molecule systems. In addition to targeting ligands, modification of polymer size has a profound effect on blood circulation time as well as on whole body biodistribution. Once in target tissue, polymeric drug delivery systems have several modes of drug release, which range from passive diffusion from a degrading nanosphere to enzymatic cleavage of drug-polymer linkers. These drug release mechanisms in combination with biodistribution play a critical role in ensuring the drug is delivered to the right area for the right duration of time.

#### 3.1 High molecular weight vs. low molecular weight carriers

The molecular weight and ultimately the hydrodynamic radius of polymeric carriers have a profound effect on the efficacy of drug delivery. Higher molecular weight polymers have longer circulating half-lives and ultimately higher deposition in diseased sites. However, as the molecular weight increases, so does uptake in the reticuloendothelial system (RES). Therefore, careful selection of molecular weights with appropriate targeting ligand concentration (number per macromolecule) is essential to ensure biocompatibility and high deposition of drug in the affected area. Wang et al. compared 24 kDa, 46 kDa, and 96 kDa Asp<sub>8</sub> containing HPMA copolymers in a biodistribution study. Each polymer contained 4.4, 5.6, and 9.6 Asp<sub>8</sub> chains per macromolecule, respectively. Also included in the study was Tyr-p-Asp<sub>8</sub> as a control. As hypothesized, the study demonstrated enhanced blood circulation time for high molecular weight conjugates with the highest bone accumulation being 96 kDa HPMA copolymer by several fold. However, the highest bone to RES accumulation ratio was the Asp<sub>8</sub> targeting ligand by itself, followed by 24 kDa HPMA copolymer conjugates [58]. The same group performed a similar biodistribution study on ALN containing HPMA copolymers of two molecular weights - 20 kDa and 90 kDa. In addition to molecular weight, the concentration of ALN (0%, 1.5%, and 8.5%) was also varied (Fig. 7). All conjugates showed an initially high uptake in the heart, lungs, and kidneys, which reduced over time. Similarly to [58] the low molecular weight, 20 kDa conjugate cleared the body first as expected and each targeted molecule increased bone accumulation over time. The high molecular weight (90 kDa) conjugate yielded a higher RES uptake. Similarly, higher ALN content also yielded higher RES uptake. This indicates that the identification of ideal targeting ligand concentrations needs to take into account the polymer's overall charge (and therefore RES uptake), number of targeting ligands per macromolecule (related to charge), and molecular weight. In the case above, the drug with the highest bone to RES ratio was the 20 kDa, 1.5% ALN polymer [60]. In conditions where metabolism of a drug by the RES is a concern, moderate molecular weights with a low percentage of targeting moieties are beneficial. However, uptake by the RES generally means phagocytosis by mononuclear phagocytes. Phagosomes eventually fuse with lysosomes, creating an acidic environment with several proteolytic enzymes. Therefore, delivery of proteins and other molecules, which will be degraded by lysosomal enzymes, may not create biocompatibility issues following RES uptake. In these situations, a high molecular weight and a high targeting moiety content may be the best option.

#### 3.2 Degradable linkers

As critical as the biodistribution of a drug, the proper release rate and mechanism will determine the efficacy of a drug delivery system. Modification of the linker, which connects

polymer with the drug, can create release profiles that deliver high amounts of drug immediately upon arrival or deliver small amounts over time, extending the effective dose time of the drug. Furthermore, by utilizing bone specific enzymes, disease site-specific prodrugs can be created.

The bone itself has several enzymes, which can be used to release drugs. Among these enzymes are matrix metalloproteinases (MMP) 1, 2, 3, 4, 7, 8, 9, 12, 13, and 14 and Cat K. MMPs are expressed by osteoclasts, osteoblasts and are overexpressed in many bone metastases [61-71]. Most MMPs function by cleaving peptide bonds holding collagen and other glycoproteins together. MMPs assist osteoclasts' function by degrading collagen within the bone at resorption lacunae while bone metastases use MMPs to degrade the basement membrane and spread through out the body. Using MMP-sensitive linkers may provide site-specific delivery of drugs to bone; however, bone delivery using MMPs is limited by the fact that MMPs are inhibited by tetracycline and derivatives such as doxycycline [72]. Cat K on the other hand is expressed at resorption lacunae and is not sensitive to tetracycline derivatives (Fig. 1). Due to expression of Cat K *outside the cell*, Cat K specific linkers are an excellent choice for signaling compounds such as prostaglandins and bone morphogenic protein-2 (BMP-2), whereas hydrazone bonds, disulfide bonds and cathepsin B (Cat B)-sensitive linkers are better options for drugs such as chemotherapeutics, which rely on internalization by the cells prior to release.

Many classic degradable drug-polymer linkers rely on receptor-mediated endocytosis of cells or phagocytosis by osteoclasts. Following endocytosis, the endosomal compartment acidifies and ultimately fuses with lysosomes into an acidic organelle (pH ~5). Acid-sensitive linkers are also effective in the low pH present in the interstitial space of tumors (pH 5-6) and the resorption lacuna of osteoclasts (pH 4-4.5) [73]. Hydrazone bonds are acid-sensitive linkers and are able to rapidly release an unaltered drug at pH 5.5 but remain intact at physiological pH 7.4. Hrubý et al. conjugated doxorubicin to an HPMA copolymer using a hydrazone linker and measured release kinetics of the HAp adsorbed polymer as well as the polymer in solution at pH 5 and 7.4. Release kinetics at pH 5 was logarithmic, with over 95% of the copolymer conjugate in solution releasing the DOX within 10 h. Release kinetics of the HAp bound polymer was slower as after 10 h approximately 55% of the DOX was released and the remaining DOX was nearly all released over the first 50 h. Meanwhile, at pH 7.4 the drug linker was relatively stable, only releasing 20% of drug after 50 h [40].

There are other release systems that rely on the endosomal environment. As the lysosome and endosome fuse, polymers within the endosome are exposed to proteases such as Cat B. Hrubý et al. also studied the release kinetics of a Cat B sensitive tetrapeptide GFLG (glycine-phenylalanine-leucine-glycine) [74]. Release kinetics were nearly linear and much slower than the hydrazone bond discussed earlier. When the polymer was in solution, after 50h less than 45% of the drug was released. Steric hindrance, preventing Cat B from accessing the tetrapeptide linker, reduced the DOX cleavage rate in HAp bound polymer to half the rate in solution, as only 22% was released after 50 h incubation [40].

Elongated spacers, which separate the enzymatically cleaved bond from the drug by a selfeliminating group, were designed by several groups [75-79]. The predominant example of an electronic cascade spacer is the 1,6-elimination spacer [77]. It contains, e.g., a bifunctional p-aminobenzyl alcohol group that is linked to an enzymatically cleavable group through an amine moiety. After enzymatic cleavage, the strong electron-donating amine group of the 1,6-elimination spacer is unmasked and immediately initiates an electronic cascade that leads to cleavage of the benzyl-carbamate bond and release of carbamic acid. The unstable carbamic acid rapidly releases carbon dioxide and yields the unmodified drug

[78,79]. Use of such spacers was shown in bone-targeted delivery of  $PGE_1$  [59] and anticancer drugs to bone [80-82].

Once a polymer exits the endosome, it enters the cytosol. The cytosol has a reducing environment that is able to reduce disulfide linkers. Evidence also suggests that this reductive environment may start as early as the endosome during receptor-mediated endocytosis [83]. Kurtoglu et al. used disulfide linkers for dendrimer drug delivery and described the release at pH 5 and pH 7.4. Using either cysteine or glutathione at pH 7.4, release occurred within 20 min, while at pH 5, complete drug release occurred over 6 h or more. Furthermore, the amount of drug released was dramatically affected by glutathione and cysteine concentrations as one cysteine can release only one drug [84].

**3.2.1 Example of cathepsin K sensitive spacer design**—The design of an oligopeptide sequence susceptible to cleavage by Cat K is demonstrated here. The aim was to design a conjugate that would deliver prostaglandin  $E_1$  (PGE<sub>1</sub>) to bone with subsequent PGE<sub>1</sub> release by osteoclast-secreted Cat K in the resorption lacuna (for macromolecular entry into osteoclasts see [85-87]).

The Cat K sensitive spacer was designed based on known interactions of peptide substrates with the active site of the enzyme [88]. The specificities of S1–S4 subsites (subsite nomenclature from Schechter and Berger [89]) of the Cat K active site have been screened with positional scanning libraries [90]. Their most striking result was the unique preference for proline in P2 position; therefore, it was selected. The favored amino acid residues in the P1 position were lysine and arginine. However, to ensure that the spacer is stable in the circulation [91], basic amino acid residues have been avoided and norleucine, the most active neutral amino acid residue, was chosen for position P1. It was shown that Gly-Pro in positions P3 and P2 were favorable for cleavage [90]. Consequently, glycine was seleceted for position P3. It had been observed previously that expanding the length of the oligopeptide spacer in side-chains of HPMA copolymers from three to four amino acid residues resulted in decreased steric hindrance upon the formation of the enzyme-substrate complex for chymotrypsin [92] and Cat B [74]. Thus, glycine was introduced into the position P4.

Chemically, the C end of Gly-Gly-Pro-Nle spacer cannot be directly connected to  $PGE_1$  because the latter lacks a  $NH_2$  group. To overcome this problem, a 1,6-elimination spacer based on 4-aminobenzyl alcohol (AB-OH) was inserted. The latter was used to link the peptide to the C-1 COOH group of  $PGE_1$ . Upon enzymatic cleavage of the peptide spacer, the strong electronic-donating amine group of the 1,6-elimination spacer is unmasked [75,76] and initiates an electronic cascade that leads to the cleavage of the ester bond and immediate release of unmodified  $PGE_1$  (Fig. 8). Finally, a methacryloyl group was attached at the N-terminus of the construct; the resulting macromonomer, MA-Gly-Gly-Pro-Nle-AB-PGE<sub>1</sub>, may be copolymerized with HPMA to produce new  $PGE_1$ -containing macromolecular therapeutics. The release of  $PGE_1$  from HPMA + MA-Gly-Gly-Pro-Nle-AB-PGE<sub>1</sub> copolymer indicated that a high rate of cleavage occurred in osteoclast cultures followed by osteoblast cultures. Polymer incubation with non-skeletal and precursor cells resulted in minimal release of  $PGE_1$ , demonstrating the specificity of this linker [93]. The specificity of the linker was also validated *in vivo* [59,80,81].

#### 3.3 Different polymers studied

Each polymer exhibits unique characteristics based on chemical and structural properties. We have classified polymers by their structural properties whether they form linear hydrophilic polymers or spherical micelle and nanoparticle structures.

#### 3.3.1 Linear polymers

**3.3.1.1 Cationic polymers:** Polyethyleneimine (PEI) and poly(L-lysine) (PLL) are frequently-used cationic polymers in drug and gene delivery systems. Uluda and coworkers incorporated a BP, 2-(3-mercaptopropylsulfanyl)-ethyl-1,1-bisphosphonic acid, to both polymers using heterobifunctional reagents. *In vitro* (HAp) and *in vivo* (unprocessed bone matrix from femurs/tibiae) mineral affinity of BP containing polymers was evaluated and compared with unmodified polymers. Interestingly, they found that *in vitro* binding to HAp was similar or lower after conjugation to BP and that both PEI and PLL were able to bind nearly 100% to HAp in many conditions on their own. This is not surprising, as positively charged molecules should adsorb to HAp. *In vivo* results on a rat model also showed no significant difference in bone affinity of modified and unmodified polymers [94]. In addition, PEI and other cationic polymers are associated with toxicity *in vivo* [95,96].

Recently, modified PEI was used for the development of bone-seeking radiopharmaceuticals. To this end, PEI was functionalized with methylene phosphonate groups thereby incorporating targeting and metal chelating ability in the polymer itself. *In vitro* tests demonstrated that fractionated N,N',N'-trimethylenephosphonate-poly(ethyleneimine) (PEI-MP) was able to carry radioactive tin ions to an HAp surface and then release them upon contact. By doing so, PEI-MP may work well as an imaging agent and as a possible radiotherapy delivery agent for metastatic bone cancer [97,98]. As such, prevention of any biocompatibility issues may be avoided by integrating PEI-MP into a block-copolymer of a more biocompatible polymer.

**3.3.1.2 Poly[N-(2-hydroxypropyl)methacrylamide]:** HPMA is one of the more studied polymer therapeutics to bone. PolyHPMA's biodistribution and bone targeting abilities have been well documented and are discussed throughout this paper [58-60,80,81]. Other advantages of HPMA copolymers include a low toxicity profile and the ability to control the molecular weight and therefore biodistribution by RAFT polymerization [99-101]. This process also permits the incorporation of methacryloylated drug derivatives into the polymer by copolymerization [99,100], eliminating the chance of having unreacted (residual) groups remaining on the polymer backbone after post-polymerization modification. HPMA copolymers have also been used in the design of micelles [102,103] and dendrimers [104].

**3.3.1.3 Other polymer carriers:** Korzhikov et al. synthesized aldehyde-containing copolymers based on 2-deoxy-2-methacrylamido-<sub>D</sub>-glucose (MAG). Two methods were used for the introduction of aldehyde groups into the polysaccharide chains: i) copolymerization of MAG with *N*-vinylpyrrolidone and a comonomer that can be easily converted to an aldehyde, such as diethylacetal acrolein; and ii) the periodate oxidation of the saccharide units. The water-soluble polymers adsorbed on HAp. The ultimate aim is to use these polymers for construction of composite scaffolds for tissue regeneration [105].

Wang and coworkers synthesized a polyrotaxane delivery system targeted by ALN. They conjugated ALN to a-cyclodextrin and threaded it onto PEG (2000 mol. wt.) producing a pseudopolyrotaxane. Then, employing copper(I)-catalyzed Huisgen 1,3-dipolar cycloaddition, the pseudopolyrotaxane was copolymerized with bulky monomers containing imaging and/or therapeutic agents. The osteotropicity of the constructs was confirmed in vivo [106].

#### 3.3.2 Vesicular polymer carriers

**<u>3.3.2.1 Micelles and liposomes:</u>** Micelles and liposomes are able to hold a great deal of drug when compared to linear polymers due to their three-dimensional shape. Although *in vivo* stability may be an issue for many polymeric spherical micelles and liposomes, these

delivery systems are able to release their contents over long periods of time [107,108]. Finally, as alluded to before, this class of drug carriers may be able to protect drugs they carry from proteasomal degradation as well as protect the body from non-targeted effects of the drugs.

Critical to their success, bone targeted liposomes are able to retain their targeting ability. Hengst et al. designed a 1,2-diacyl-*Sn*-glycero-3-phosphocholine (EPC) liposome containing a BP targeting agent anchored by cholesterol. Each of the four conjugates tested had a low polydispersity of 0.1. Mole percents of the targeting ligand, cholesteryl-trisoxyethylenebisphosphonic acid (CHOL-TOE-BP), were 0%, 3%, 14%, and 25% with zeta potentials of  $-18 \pm 2$ ,  $-28 \pm 1$ ,  $-34 \pm 2$ , and  $-40 \pm 2$ , respectively. The determination of the binding of each liposome revealed that 50% HAp binding was achieved using the 3 mol% CHOL-TOE-BP and binding progressed nearly linearly until 100% binding was achieved using the 25 mol% CHOL-TOE-BP liposome. This data suggests the bone targeting potential of this liposome [109].

Uluda and coworkers incorporated BP-derivatized liposomes into collagen/hydroxyapatite scaffolds and decreased the rate of model drug release from the constructs. This provides a sustained drug release approach for bone regeneration [107].

The liposomal drug formulations for the treatment of rheumatoid arthritis have been recently reviewed [110].

**3.3.2.2 Poly(lactic-co-glycolic acid):** (PLGA) nanospheres are a well-studied drug delivery modality that has been applied to bone targeting. Choi et al. designed PEG-PLGA block copolymers intermixed with ALN functionalized PLGA to create surface modified nanoparticles. The monomethoxy PEG formed a hydrophilic layer on the surface of nanoparticles with the potential to decrease RES accumulation, and ALN was used as the targeting moiety. The study of the relationship between structure and adsorption on HAp revealed decreasing adsorption correlated with a decreasing concentration on ALN at the surface; the increase in PEG mol. wt. (550, 750, and 2000) resulted in decreased adsorption apparently due to shielding of the ALN moieties by longer PEG chains [41]. As PLGA degradation yields naturally occurring products it also has a low toxicity profile. Cenni et al. was able to synthesize ALN functionalized nanospheres with a polydispersity index of 0.348  $\pm$  0.020 nm and test their toxicity. Each *in vitro* test (hemolysis, leukocyte proliferation, platelet number, coagulation, and complement consumption) showed acceptable blood compatibility and an absence of cytotoxicity [111]. In follow-up studies, the authors loaded ALN functionalized PLGA nanoparticles with doxorubicin (DOX) and evaluated their properties in an orthotopic mouse model of breast cancer bone metastases. Both free DOX and DOX-loaded nanoparticles reduced the occurrence of metastases in mice. Loaded and drug-free nanoparticles decreased the number of osteoclasts at the tumor site; the authors hypothesized that this may be the result of ALN activity [112].

Other advantages of PLGA include well-studied drug release kinetics. By modification of lactic glycolic acid ratios, PLGA nanospheres can release their contents over a wide time scale [113-115]. More *in vivo* tests such as bone targeting and whole body biodistribution are needed, however, in order to truly determine the efficacy of these targeted polymeric delivery systems.

<u>**3.3.2.3 Other vesicular structures:**</u> Ozcan et al. used nanoprecipitation to produce  $poly(\gamma - benzyl-L-glutamate)$  (PLGB) nanoparticles (< 80 nm; low polydispersity), which were functionalized with PEG-ALN for bone targeting. ALN presence did increase bone targeting

dramatically *in vivo*. By labeling the PBLG-PEG-ALN nanospheres with FITC, Ozcan et al. was able to demonstrate preferential accumulation of targeted drug in bone [116].

Wang et al. evaluated bovine serum albumin (BSA) nanoparticles surface-coated with PEI-PEG-thiolBP and used them for the delivery of bone morphogenetic protein-2 (BMP-2). They found that the nanoparticles were not effective in bone targeting after i.v. administration. However, they were useful in localized delivery of BMP-2 in bone repair and regeneration [117].

#### 3.4 Dendrimers, dendrons, dendronized polymers

Dendrimers are symmetrical vesicular carriers, whereas dendrons are tree-like fragments of dendrimers that can be attached to water-soluble polymers [118]. Thus, the "dendron-like" systems possess complex structures characterized by high density of functional groups and/ or targeting moieties. Ideally (without structural defects), dendrimers are monodisperse, spherical compounds. Their size depends on the generation, i.e., the number of successive synthetic steps used. Numerous attempts have been made to exploit dendrimers in drug delivery including targeting to bone. Ouyang et al. synthesized naproxene containing polyamide and poly-ether-amide dendrimers and decorated them with 2 or 3  $Asp_{(4-6)}$ peptides and demonstrated affinity for HAp [52]. In the following study the same group synthesized the "next generation" of naproxene containing dendritic compounds with four or eight  $Asp_{(4-6)}$  fragments. They did not observe a difference in HAp binding of constructs containing low or high Asp<sub>(4-6)</sub> content and the total HAp binding only ranged from 63% to 75% [119]. Clementi et al. synthesized PEG-based dendrimers from heterobifunctional PEG; they were decorated with  $H_2N$ -PEG- $\beta$ -Glu-( $\beta$ -Glu)<sub>2</sub>-(COOH)<sub>4</sub>, exposing four carboxylic groups for the attachment of ALN and/or paclitaxel. This construct was able to achieve nearly 100% HAp binding and nearly 80% HAp binding following conjugation to paclitaxel (PTX). These conjugates were designed for the treatment of bone neoplasms and used a pH-sensitive linker for rapid release of the drug. In vitro studies demonstrated that drug release was highly effective resulting in conjugated PTX having a similar IC<sub>50</sub> as PTX alone, apparently due to fast drug release at physiological pH [120].

## 3.5 Vesicular vs. linear polymer carriers

The decision to use a spherical or linear polymer depends on the application. Both types can be designed so that they release drugs rapidly. Longer periods of time (such as a day to a week) may be achieved using linear polymers conjugated to drugs via enzyme degradable linkers. For periods over a week, spherical polymers may be designed to release drug as the polymer degrades. The question then arises as to which yields better targeting and how it affects the treatment of disease. This may be answered by looking at zeta potentials. Excess surface charge is associated with high RES uptake [121]. This means that for a specific polymer at a specific hydrodynamic radius, the addition of surface charge, in this case as a targeting moiety, could increase RES uptake. As such, linear polymers may have an advantage. Linear polymers are able to bend and conform to a surface thereby reducing the number of targeting moieties needed to achieve bone specificity. Spherical polymers on the other hand have a three-dimensional shape, which reduces the amount of surface that can bind to bone thereby increasing the need for several targeting moieties and increasing the zeta potential. Though this theory needs to be tested, one may consider that if RES uptake does not contribute to a decrease in biocompatibility (such as with drugs that can be degraded by RES lysosomes) then a spherical polymer as well as a linear polymer would work. However, if RES uptake results in toxicity, as with nondegradable chemotherapeutics, then one may consider the use of a linear polymer over a spherical one.

# 4. Therapeutics

Life expectancy in the developed world continues to increase at a constant rate [122]. As such, age-related health complications also increase and the need for therapeutics to treat bone degenerative diseases and cancers expands. Critical to solving problems associated with bone disease is the selection of proper therapeutics. Bone disease causes a wide range of problems: too little bone growth, cancer and overgrowth, inflammation, and infection. In this paper, focus will be on bone anabolic agents (for osteoporosis) as well as cytotoxic agents (for cancer). Other topics such as inflammation [123] and siRNA delivery to bone [124] are discussed in other reviews within this volume.

#### 4.1 Osteoporosis and other low bone mineral density (BMD) related diseases

Osteoporosis is among the most common of the degenerative diseases. It is estimated that 1 out of 5 women over the age of 50 have osteoporosis and that approximately half of the women over 50 will fracture their hip, wrist or vertebra [125]. BPs, although helpful for some people, have several side effects as described previously (section 2.3). Furthermore, a need exists for alternatives to Cat K and RANKL inhibitors [126-130]. Ultimately, these drugs yield inadequate bone turnover and therefore poor bone quality. The 1-34 fragment of recombinant parathyroid hormone (PTH), FDA approved in late 2002, is an effective bone anabolic agent. When compared to oral 10 mg/day ALN, the administration of s.c. injections 40 µg/day PTH yielded greater bone density [131]. However, there is a concern that lifelong injections may cause osteosarcomas to develop, and consequently, the drug may only be prescribed for 2 years during one's lifetime [132]. Other bone anabolic agents have also been investigated for treatment of osteoporosis but ultimately have significant drawbacks preventing their use. Prostaglandins are in clinical use for several medical conditions. In bone, they exhibit strong anabolic effects; however, the short half-life and serious cardiovascular side effects of free prostaglandins have prevented their use in osteoporosis. BMPs have been clinically applied to fuse vertebrae. BMPs have low systemic side effects, which may be in part due to their short blood half-life, which prohibits effective systemic administration. Targeting bone anabolic agents and/or extending their circulation half-life may open the door for new PTH alternatives.

## 4.2 Anabolic agents

Several potential bone anabolic agents exist. We will discuss their interactions in the hopes of elucidating possible synergistic mechanisms, which may be employed to create superdrugs (Fig. 9).

**4.2.1 Prostaglandins**—Prostaglandins (PGEs), especially the E series (such as PGE<sub>1</sub> and PGE<sub>2</sub>), have been studied extensively for their bone anabolic properties. Their importance in bone turnover is demonstrated by the lack of bone repair when patients are administered NSAIDs [133-137]. Systemically administered prostaglandins, however, cause diarrhea, lethargy, and flushing and thus are too toxic for non-local or non-targeted dosing [138]. Locally administered to bone, PGE affects osteoblasts by binding to four receptor subtypes (EP<sub>1-4</sub>) of PGEs. EP<sub>2</sub> and EP<sub>4</sub> are the primary receptors associated with bone turnover; EP2 is normally associated with bone anabolism and EP<sub>4</sub> generally affects catabolism by upregulating RANKL through the MAPK pathway [139-143]. Both EP<sub>2</sub> and EP<sub>4</sub> have traditionally been associated with a dramatic increase in cAMP levels; however, these two receptors also contribute to the p38 MAPK and ERK MAPK signaling pathways, respectively [144]. There is also evidence that prostaglandins upregulate BMP-2 production [32]; however, another study contends that it is not through EP2 or EP4 [144].

Several studies have attempted to treat osteoporosis by either modification of PGE<sub>2</sub>, making it specific for EP<sub>2</sub> or EP<sub>4</sub> receptors, or by conjugation to targeting moieties [59,138,145,146]. Miller et al. has attached PGE<sub>1</sub> via a Cat K sensitive spacer, Gly-Gly-Pro-Nle, to an HPMA copolymer (Mw 37.2 kDa), P-Asp<sub>8</sub>-FITC-PGE<sub>1</sub>. Ovariectomized female Sprague-Dawley rats were administered a single i.v. bolus of either 10 mg P-FITC, P-Asp<sub>8</sub>-FITC or P-Asp<sub>8</sub>-FITC-PGE<sub>1</sub> (0.61 mg of PGE<sub>1</sub> equivalent), followed by a tetracycline label administration at the end of the study on day 25. Measuring the distance between the FITC-labeled polymer and the tetracycline label on the bone surface, they were able to determine the surface-referent bone formation rates ( $\mu$ m<sup>2</sup>  $\mu$ m<sup>-1</sup> day<sup>-1</sup>). Bone deposition rates attributed to P-FITC, P-Asp<sub>8</sub>-FITC or P-Asp<sub>8</sub>-FITC-PGE<sub>1</sub> were 0.0362 ± 0.0062, 0.0541 ± 0.0108, and 0.1142 ± 0.0077, respectively, demonstrating the potential anabolic effects of targeted PGE<sub>1</sub> against osteoporosis [59].

**4.2.2 Statins/bisphosphonates**—Statins and BPs affect bone turnover by interacting with the mevalonic acid pathway, and down-regulating protein prenylation. It is believed that down-regulating protein prenylation, down-regulates RhoA and by so doing upregulates eNOS. The presence of eNOS increases synthesis of BMP-2 and therefore increases bone anabolism [147,148]. Furthermore, statins and BPs may also directly upregulate the PGE pathway, as COX-2 inhibitors dramatically reduce statin induced bone anabolism by up to 77% [149]. Statins also induce VEGF, which contributes to osteoblast differentiation [150,151]. Statins, having been administered systemically for lowering cholesterol, also affect on bone growth. Clinical evaluation of systemic delivery has shown no decrease in fractures, minimal increase in BMD and no increase in bone turnover markers in postmenopausal women [152,153]; however, local administration in animals has shown significant bone turnover [33,154-158]. Systemic delivery of bone-targeted statins may increase local bone concentrations enough for higher BMD and bone turnover.

**4.2.3 TGF** $\beta$  **superfamily**—BMP-2 is a member of the TGF $\beta$  superfamily. It is produced by osteoclasts and the signal differentiation of osteoblasts through the SMAD5/8 pathway. SMAD5/8 combines with SMAD1 and translocates to the nucleus where Cbfa1/Runx2 is upregulated. CBFA1/Runx2 is responsible for promotion of several osteoblast differentiation genes as well as the upregulation of COX2 [159]. Although overlap can be seen in the two pathways, BMP-2 operates seperately from PGE. Research suggests that BMP-2 has a synergistic effect with PGE [160]. BMP-2 is clinically applied locally for spinal, oro-maxillary, and trauma surgeries [161-169]. Also part of the TGF $\beta$  superfamily, TGF- $\beta$ 1 and IGF-I are produced by osteoclasts [15] and affect transcription via several SMAD proteins. TGF- $\beta$ 1 and IGF-I activate NF $\kappa$ B leading to increased levels of COX-2 and PGE2 production [170,171]. TGF- $\beta$ 1 is generally considered to inhibit osteoclasts, however, as with many growth hormones, their action is dose dependent [172-174].

**4.2.4 Fragment 1-34 of parathyroid hormone (PTH)**—PTH1-34 (PTH contains 84 AA) is prescribed for treatment of osteoporosis. Systemically, PTH1-34 affects calcium resorption in the kidneys and intestine [175]. Within bone, PTH 1-34 affects osteoblasts and leads to increased cAMP/PKC production and differentiation even when COX-2 is inhibited [176]. Increased cAMP levels also open up calcium channels leading to increased calcium levels in the cell. Consequently, the calcineurin pathway is triggered, which activates NFAT and yields increases in COX-2 expression [177]. Elevated levels of PTH1-34 reduce apoptosis by increasing PKC levels there by upregulating FOXO3a DNA repair mechanisms [178]. Overexpression of cAMP/PKC leads to deleterious proteasomal (smurf1) degradation of CBFA1/Runx2, and therefore, regulation of BMP-2 effects [179-181]. Furthermore, proteosomal degradation will negate the anti-apoptotic effects of FOXO3a [182]. Therefore,

intermittent administration of PTH1-34 leads to anabolism, while chronic administration leads to catabolism [183].

# 4.3 Delivery of Proteins

Numerous attempts have been made to modify protein-based therapeutics with bone-seeking moieties to enhance their localization at skeletal sites [11,12,184]. Bansal et al. synthesized clustered BP-based targeting moieties composed of four bisphosphonic acid moieties linked through a benzene ring and conjugated them to bovine serum albumin (BSA) as a model protein. *In vitro* binding in both saline and 60% serum solutions resulted in less than 60% bound protein. However, *in vivo* results indicated a 4.7 fold increased accumulation in rat tibiae over controls (BSA and BSA targeted with a thiol-BP moiety). Thus the clustered BP possessed a considerably increased bone-seeking capacity *in vivo* when compared to HS-BP [185]. Interestingly, the accumulation of thiolBP modified osteoprotegerin in osteoarthritic rats undergoing active bone remodeling increased more than 4x over that of control unmodified OPG 24 h after i.v. administration [186].

Proteins such as BMP-2 have a circulation half-life of only about 1 min [187]. For successful systemic administration they may have to be modified by semitelechelic (ST) water-soluble polymers. It is well known that covalent attachment of PEG [188] or ST polyHPMA [189] may extend the intravascular half-life of proteins and/or vesicular drug carriers.

#### 4.4 Bone metastases

Paget's 1889 "seed and soil" theory of bone metastases remains the generally accepted theory regarding metastasis locating to bone. This theory simply reasons that as metastatic cells (the seeds) enter into the bloodstream they will be carried everywhere, but they will only attach and grow where the conditions are right (the soil) [190]. Examples of this include bone marrow expression of CXCL12 [191,192], whose complementary CXCR4 is found on several prostate and breast cancers [193-195]. Once in the bone, bone anabolism and catabolism causes patients severe bone pain and increased mortality [196]. Treatment of these metastases often involves small molecule therapeutics, which target various osteoclast mechanisms. Several small molecule osteoclast targeted therapeutics have been studied: vacuolar H+-ATPase inhibitors, Reveromycin A, Methyl Gerfelin, C-Src Inhibitors, a<sub>V</sub>b<sub>3</sub> integrin inhibitors, Cat K inhibitors) [197] and BPs [50]. Although these do reduce bone pain and some degree of metastatic growth, their effects may be increased as other synergistic drugs are added to reduce growth and tumor progression.

As bone metastases progress, HAp is exposed, providing a target for BP attachment [198]. The Satchi-Fainaro group conducted several experiments combining the anti-angiogenic properties of ALN with other chemotherapies. Paclitaxel (PTX) has anti-angiogenic properties in low doses. Miller et al. used ALN targeted PTX containing HPMA copolymers to reduce migration and proliferation of human umbilical vein endothelial cells (HUVEC). Cat B is a protease expressed by both proliferating endothelial and prostate cancer cells. Miller et al. incorporated a Cat B sensitive linker (GFLG) for both PTX and ALN. *In vitro* data suggest that the HPMA copolymer inhibited migration and proliferation of both human prostate adenocarcinoma (PC3) and HUVEC cells [199].

In a follow up study Miller et al. demonstrated the *in vivo* tumor inhibition and safety profiles on Balb/c mice. The PTX, ALN, HPMA copolymer showed no significant WBC toxicity while free PTX significantly reduced WBC counts. Intratibial injections of mCherry-labeled 4T1 cells were used to mimic breast cancer metastasis to the bone. Following i.v. administration of either PTX, ALN HPMA copolymer or free PTX+ ALN in

combination, the polymer conjugate inhibited tumor growth by 60%, while free PTX + ALN only inhibited 37% as compared to controls [200].

Segal et al. conducted a similar set of experiments using TNP-470 bound to an HPMA copolymer conjugate targeted with ALN. TNP-470 demonstrated high efficacy in clinical trials but too many side effects prevented clinical applications [201]. Segal et al. sought to reduce TNP-470 side effects by conjugation and targeting. Both ALN and TNP-470 were conjugated to HPMA using Cat K sensitive linkers (Gly-Gly-Pro-Nle). *In vivo* studies demonstrated (Fig. 10) that not only do ALN and TNP-470 have synergism, but revealed that the HPMA copolymer decreased osteosarcoma growth by 96% compared to the control, as opposed to 45% with free ALN in combo with TNP-470 [80]. Similar to the PTX, ALN, HPMA copolymer, TNP-470, ALN HPMA copolymer's toxicity is low, as opposed to ALN + TNP-470, which caused *in vivo* weight loss, neurological dysfunction, and low WBC counts [81].

#### 4.5 Radiation therapy

Targeted delivery of radioisotopes provides opportunities to open up the field of theranostics. Several groups have effectively reduced tumor size and alleviated bone pain by radiolabeling BPs [202,203]. PEI-MP bound tin is one of the few studies which uses polymer attributes in radiation therapy to bone. Jansen et al. demonstrated effective HAp binding of Sn<sup>2+</sup> and Sn<sup>4+</sup> chelated PEI-MP. In order to demonstrate the EPR effect in combination with HAp adsorption, however, an *in vivo* study is still necessary [97,98]. Although more tests are needed in order to determine the efficacy of delivering radiolabeled polymers to bone metastases, one must still be curious about the possibilities. Although radioactive polymers may need to be kept small as to reduce circulation time, one may still see greater tumor accumulation due to the EPR effect. Furthermore, the dose can be increased with polymer therapeutics as several chelating moieties can be attached to one molecule. Finally, a single polymer may be able to destroy cancer stem cells by radiosensitizing a tumor using a compound such as perifosin and then killing the stem cells using radiation [204].

# 5 Conclusions

Targeted polymer therapeutics continue to find niches in the medical world. Much of this research has focused on the EPR effect and delivery of chemotherapeutics and siRNA to solid tumors. Although this is excellent research, the characteristics of bone supply a fertile ground for future clinically relevant targeted-polymer drug research. Many applications favor macromolecular drug delivery over small molecule drugs. For example, one must consider that achieving the highest binding constant possible may not suit one's needs. Having an extremely low dissociation constant means that the polymer will most likely bind kinetically to the first bone that it comes in contact with. Modification of mole percent targeting ligands, or the ligand itself may reduce the  $K_d$ , thereby favoring thermodynamic adsorption over kinetic adsorption. By doing so, one may reduce promiscuity of the targeting ligand by limiting adsorption to the diseased area, such as a tumor. This method of targeting lends itself to targeted polymer therapeutics rather than their small molecule counterparts. Furthermore, many bone diseases are chronic conditions. Osteoporosis and osteoarthritis sufferers would benefit from extended controlled drug release mechanisms, which suit macromolecular drug delivery systems. As an underutilized application for targeted polymer drug delivery, bone targeted polymer therapeutics have yet to reach their full potential.

#### 5.2 Gaps in current research

Many novel bone-drug applications have yet to be explored, but first it is important to better develop disease-specific targeting and discover specific biochemical pathways involved in disease states. More accurate targeting to specific diseases involves finding out what bone conditions are present in each disease and then tailoring targeting mechanisms to those conditions. In this paper we discussed several mechanisms by which modification of targeting mechanisms will lead to drug accumulation on varied HAp crystalline states. Currently, most papers stop prior to in vivo data, and pre-in vivo binding studies use a standard HAp. Appropriate pre-in vivo testing should include binding assays using HAp crystals sized for specific disease conditions. Information also lacking with regards to many bone diseases is a deep understanding of the pathology and biochemistry which governs each disease. For example, this paper discussed some of the basic principles of bone anabolism affecting diseases such as osteoporosis. Some crosstalk between signaling pathways such as BMP-2 and PGEs were discussed, but great gaps in our knowledge remain. Studies such as Lee et al., where basic research on bone anabolic pathways are elucidated can be performed side-by-side with applied research on discovering novel delivery approaches and are critical for progression of the field [115]. The realization of synergistic drugs delivered to bone depends on this basic research.

## 5.3 Future of the field

With attention to proper targeting techniques, the applications in bone-targeted therapeutics have several functions it can fulfill. Relatively few novel developments of bone targeted antineoplastic agents delivered by polymers have been explored. Bone metastases are manifestations of severe cancers representing a largely unsolved problem in oncology. As such, small improvements may go a long way in the clinic. As previously discussed, locally applied radiation therapy delivered with a sensitization drug, or two chemotherapeutics acting on different pathways are a good place to begin in solving this problem.

Combination therapies can also be applied to osteoporosis. It is clear that combining drugs such as BMP-2 and PGE2 is worth exploring but dosing regimens may also yield some dogma-changing discoveries. Development of bone anabolic agents for the treatment of osteoporosis has always been a tricky balancing act. As seen with PTH1-34, overstimulation of an anabolic biological pathway over time can result in osteosarcoma. Reducing the chance of osteosarcoma may be accomplished by the development of several targeted bone anabolic agents that affect different pathways and applying them over different lengths of time, never exacerbating one particular pathway.

Lessons learned from targeting osteoporosis may then be applied to developing anabolic agents to aid healing following orthopedic surgeries and reduce long-term bone resorption around prosthetic stems. Furthermore, treatment of deep bone infections may be reduced by a pre- and post- surgery targeted antibiotic rather than antibiotic-loaded bone cements, which some argue have toxicity issues.

Bone diseases continue to increase in prevalence with aging populations. Successful advancements in bone targeted polymer therapeutics will yield clinically relevant drugs that will reduce pain and increase quality of life for millions of people.

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# Figure 1.

General osteoclast structure and function demonstrating unique environments, which can be utilized for site-specific release of drugs. Osteoclasts sequester portions of bone by sealing off areas called lacunae. The adjacent membrane to the bone ruffles and releases cathepsin K, and HCl, reducing the pH to 4-4.5 and dissolving the bone. Calcium from the bone is then transported to the secretory domain and released into the interstitial space. Although not specific to osteoclasts endosomes/lysosomes reduce pH to 5-6 and contain cathepsin B, two environmental specific attributes which can be used for the design of polymer-drug linkers.

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Figure 2.

General structure of bone-targeted polymeric nanomedicines.



# Figure 3.

Structures of several bone-targeting molecules including tetracycline as well as minimalized tetracycline (3-amino-2,6-dihydroxybenzamide) and its conjugate with estradiol [29]. Also shown, are the structures of several bisphosphonates in comparison with pyrophosphate. Acidic oligopeptides such as aspartic acid (shown) or glutamic acid 4-10 amino acids long are also excellent bone-targeting molecules.

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# Figure 4.

 $\dot{\text{MDP-Tc}}^{99}$  scan demonstrates the targeting specificity of bisphosphonates. Shown here is a primary osteogenic sarcoma from a 72-year-old man with Paget's disease. Dark arrows indicate a metastasis to inguinal lymph node. Reprinted with permission from reference [36].

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#### Figure 5.

Atomic force microscopy histograms demonstrating rupture forces of (**A**) alendronate and (**B**)  $_{\text{D}}$ -Asp<sub>8</sub> modified cantilever tips from a tooth enamel surface. (**C**) Binding ability of FITC labeled HPMA copolymer-ALN conjugate (P-ALN-FITC) and HPMA copolymer- $_{\text{D}}$ -Asp<sub>8</sub> conjugate (P- $_{\text{D}}$ -Asp<sub>8</sub>-FITC) to hydroxyapatites with different crystallinity; black bars are high crystallinity and white bars exhibit low crystallinity. Adapted from reference [3].



#### Figure 6.

Initial uptake and localization of FITC labeled HPMA copolymer-Asp<sub>8</sub> conjugate in bone compared with the uptake of tetracycline. (**A**) The conjugate preferentially incorporates in scalloped-appearing eroded surfaces in cancellous bone (white arrows); tetracycline (yellow label) is incorporated onto active bone mineralization surfaces. (**B**,**C**) Stained (**B**) and unstained (**C**) section of the same region of the proximal tibial growth plate and primary spongiosa. Tetracycline (yellow label) incorporated into the mineralizing zone of the growth plate (**C**, orange arrowhead) as expected, whereas HPMA copolymer-Asp<sub>8</sub> conjugate (green label) localized in the resorption areas of the primary spongiosa (**C**, white arrows). Magnifications: A=150x; B,C=125x. Reprinted with permission from reference [59].

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# Figure 7.

Biodistribution in BALB/c mice of <sup>125</sup>I-labeled HPMA copolymer-Asp<sub>8</sub> conjugates (P-Asp<sub>8</sub>) 24 h and <sup>125</sup>I-labeled HPMA copolymer-ALN conjugates (P-ALN) 48 h after i.v. administration. The impact of molecular weight and ALN content was evaluated. Adapted from references [58,60].



#### Figure 8.

Scheme of release of unmodified prostaglandin  $E_1$  (PGE<sub>1</sub>) from HPMA copolymer-Asp<sub>8</sub>-PGE<sub>1</sub> conjugate. Rate controlling cathepsin K cleavage is followed by fast 1,6 elimination.



#### Figure 9.

General pathways affecting bone anabolism and cross talk between osteoclasts and osteoblasts. Most notably, PGE levels affect BMP-2 levels and vice versa, however, each contributes to bone anabolism by independent signaling pathways. Statins upregulate both BMP-2 and PGE through independent pathways. PGE and PTH1-34 upregulate cAMP however, stimulation of EP2 or EP4 by PGE will also trigger MAPK cascades. Not shown, PTH1-34 affects calcium levels in the body by regulating resorption in the kidneys and intestine. Wnt plays a critical role in bone turnover by production of osteoprotegerin and therefore inhibition of RANKL-RANK interactions. Also of note, COX-2 represents basic components PGE production rather than upregulation of COX-2 enzyme.

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## Figure 10.

Treatment of mCherry-labeled MG-63-Ras osteosarcoma tumor-bearing mice with free ALN and TNP-470, HPMA copolymer-ALN-TNP-470 conjugate, or vehicle alone. (A) Structure of the HPMA copolymer-ALN-TNP-470 conjugate. (B) Intravital non-invasive fluorescence imaging of the tumor. (C) Tumor volume of mCherry-labeled MG-63-Ras tumor-bearing mice treated with free ALN and TNP-470 (open triangles), HPMA copolymer-ALN-TNP-470 conjugate (closed triangles), or vehicle alone (closed squares) [80].

# Table 1

Maximal binding rate  $(B_{max})$  and dissociation constant  $(K_d)$  of acidic oligopeptides with increasing number of amino acids.

Compound	$K_{d}\left(\mu M ight)$	B <sub>max</sub> (nmol/h/100µg HAp)	B <sub>max</sub> /K <sub>d</sub>
Fmoc-(L-Asp) <sub>n</sub>			
n=2	>100	n.d.	
4	12.1 ± 1.3 *	$0.34\pm0.04$	0.028
6	$6.03\pm0.84$	$1.57\pm0.16$	0.026
8	$5.24\pm0.55$	$1.61\pm0.18$	0.307
10	$2.52\pm0.23$	$1.66\pm0.17$	0.659
Fmoc-(L-Glu) <sub>n</sub>			
n=2	>100	n.d.	
4	$13.2\pm1.1$	$0.42\pm0.05$	0.032
6	$6.38\pm0.71$	$159\pm0.17$	0.249
8	$5.21\pm0.61$	161 ± 0.13	0.309
10	$2.39\pm0.25$	$1.66 \pm 0.16$	0.695

\* Standard errors are shown. Adapted from reference [53].