

Synthesis of *N*-Substituted Indoles via Aqueous Ring-Closing Metathesis

Valerio Sabatino¹, Dario Staub¹, Thomas R. Ward^{1*}

¹ Department of Chemistry, University of Basel, Mattenstrasse 24a, Biopark Rosental, 4058 Basel, Switzerland

* Corresponding author: Thomas.ward@unibas.ch

Abstract

We report herein the synthesis of *N*-substituted indoles resulting from the ring-closing metathesis of indole precursors bearing *N*-terminal alkenes. The aqueous metathesis of the indole precursors gave good yields of *N*-substituted indoles (up to 72 %) with commercial metathesis catalysts and with artificial metalloenzymes based on the biotin-streptavidin technology. Strikingly, the yield of the *N*-acetylindole increases in presence of a second metathesis substrate.

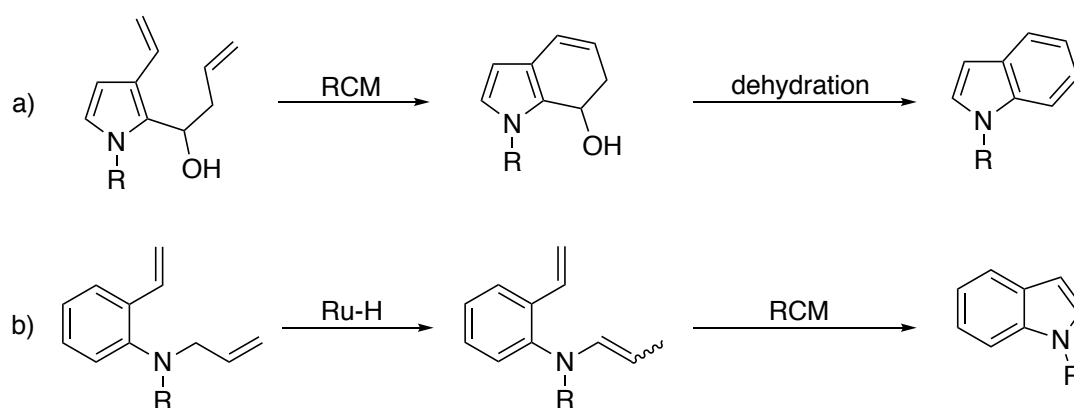
Keywords Aqueous catalysis, Ring-closing metathesis, Homogeneous catalysis, Artificial metalloenzymes

Introduction

Indoles are important synthetic scaffolds [1,2]. The indole core is present in many compounds which possess biological activity, such as naturally-occurring alkaloids and chemotherapeutic drugs [3-6]. Additionally, indole is a metabolite in the biosynthetic pathway of tryptophan, an essential amino acid playing a critical role in the metabolism of eukaryotic and prokaryotic cells [7].

Synthetic strategies relying on ring-closing metathesis (RCM) for the synthesis of indole derivatives include *i*) the formation of a pyrrole ring from a functionalized benzene precursor and *ii*) the formation of a benzene ring from a functionalized pyrrole precursor.

We sought to design suitable substrates for the synthesis of *N*-substituted indoles via aqueous RCM. Two protocols for the RCM of indoles derivatives in organic solvents are displayed in Scheme 1 [8,9]. Yoshida and coworkers generated indoles via a tandem RCM/1,2-elimination [10,11] sequence (Scheme 1a) and Nishida and coworkers reported on a mechanism of selective isomerization of terminal olefins promoted by a ruthenium hydride, followed by RCM to yield indoles (Scheme 1b).



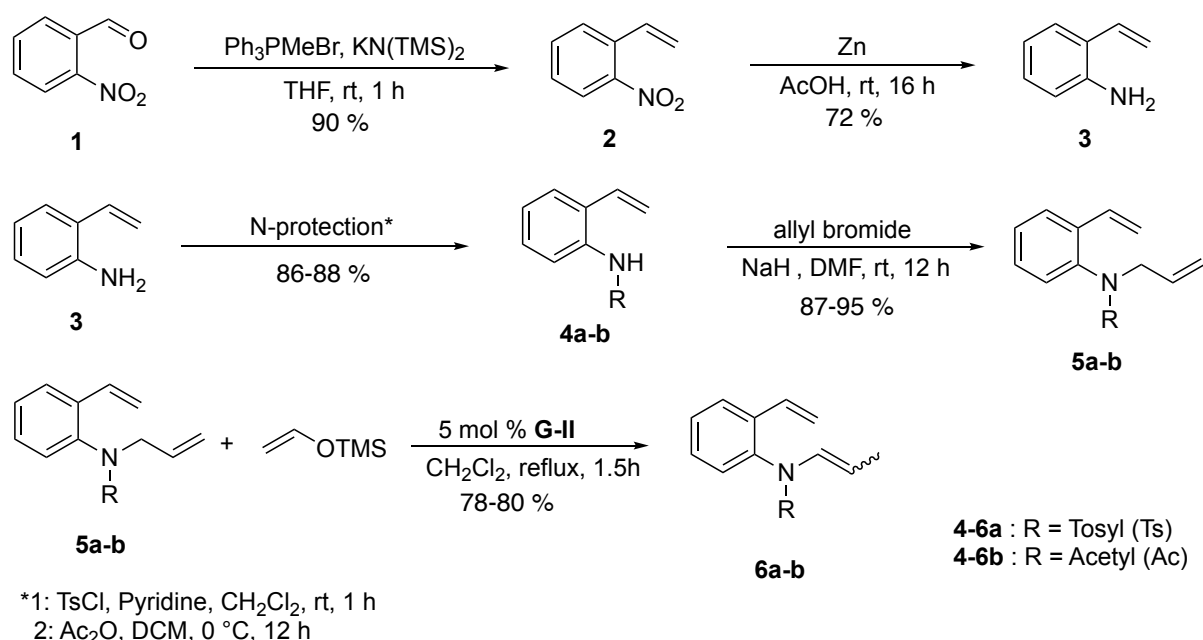
Scheme 1. Reported examples for the synthesis of indole derivatives via ring-closing metathesis [10,11].

Herein, we report on our effort to synthesize *N*-protected indoles starting from *N*-substituted anilides via RCM in aqueous solution. Both homogeneous and an artificial

metathase based on the biotin-streptavidin technology were evaluated, Figure 1 [12-20].

Results and discussion

We initially synthesized the precursors **6a-b** via the synthetic route in Scheme 2. Based on the work of Nishida and coworkers [8], starting from *o*-nitrobenzaldehyde **1**, we reproduced the synthesis of the two substituted indoles **7a-b** by generating the indole precursors **6a-b** containing an internal alkene via a Ru-H promoted isomerization (Scheme 2) [21-24].



Scheme 2. Synthesis of *N*-substituted indole precursors **6a** and **6b**.

Next, we tested the RCM in aqueous buffer under mild conditions with different commercially-available metathesis catalysts (**G-II**, **HG-I**, **HG-II** and **Aquamet**) and with the biotinylated catalyst **Biot-Ru**, Figure 1.

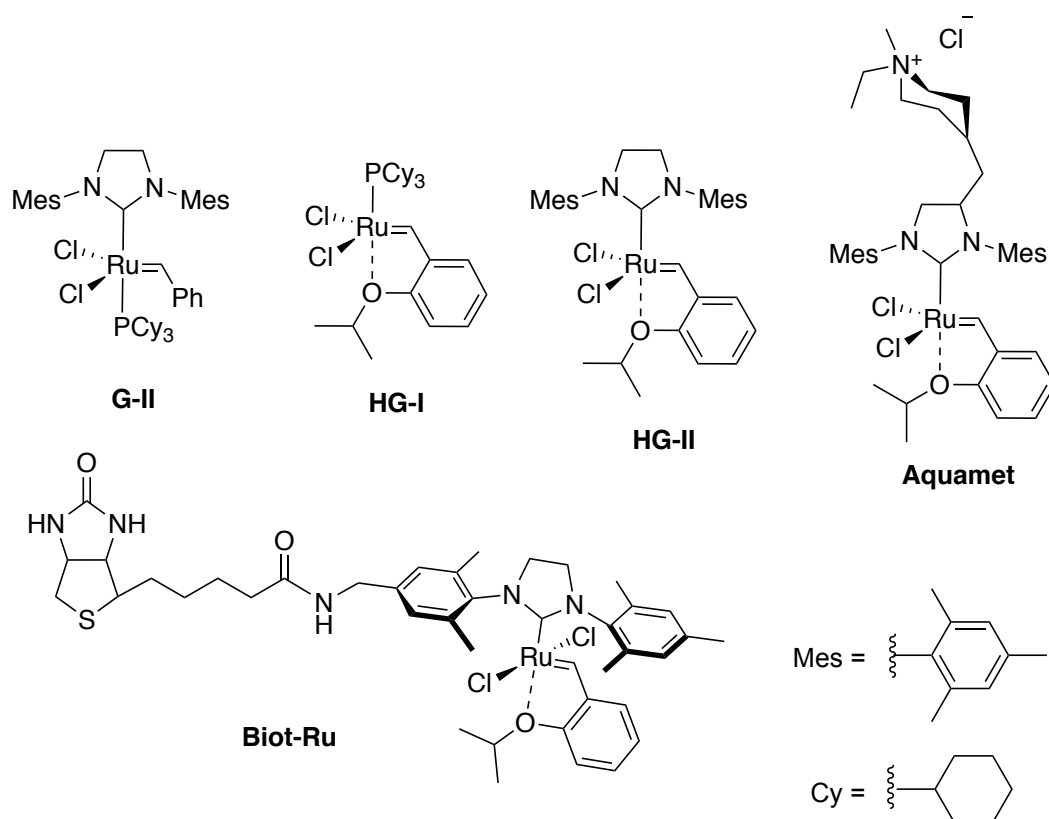
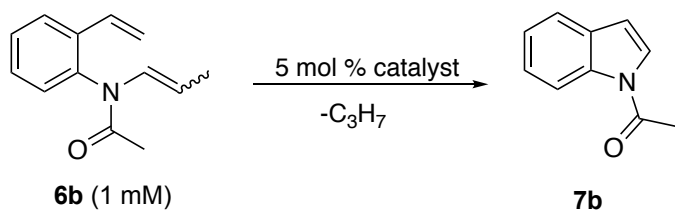


Figure 1. Ruthenium-based metathesis catalysts evaluated for the aqueous RCM of indole precursors.

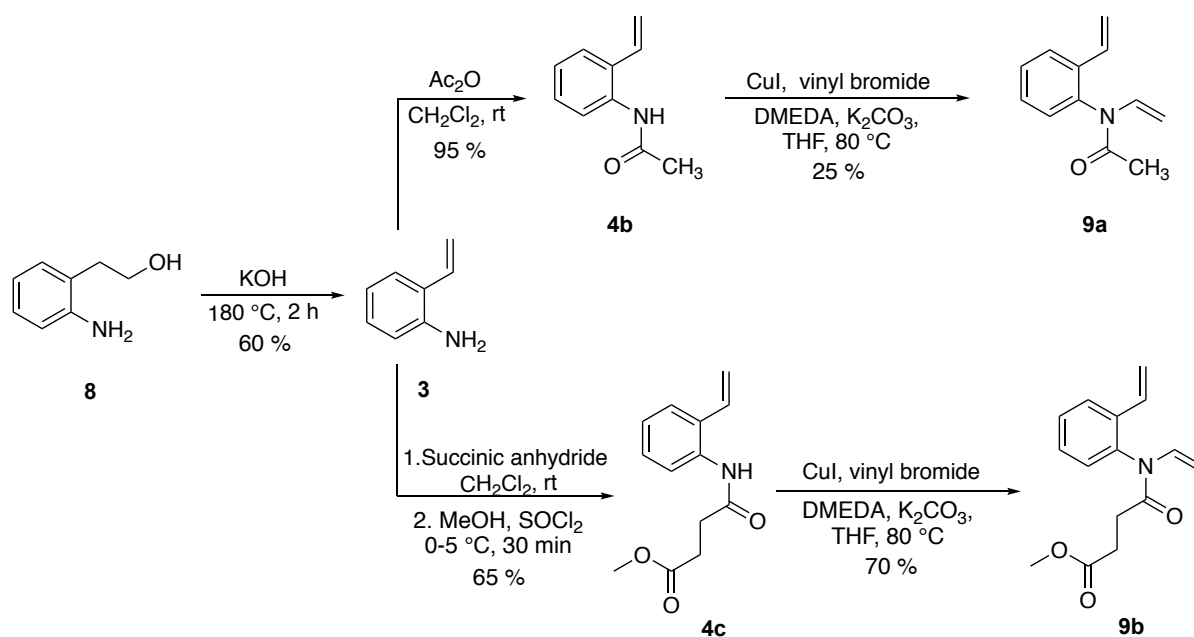
Substrate **6a** is insoluble in water even in the presence of up to 20 % of organic solvent. It forms either a milky suspension or a precipitate. The solubility in water improves with the *N*-acetylindole precursor **6b**, but activity screening in the presence of different buffers barely achieved a single turnover as summarized in Table 1.

Table 1. RCM of indole precursor **6b** in aqueous buffer.

Entry	Solvent ^a	Catalyst (mol %)	Product (μM)	TON
1	PBS + HSA	Aquamet (5)	—	—
2	AcOH/AcONa + HSA	Aquamet (5)	—	—
3	PBS	Aquamet (5)	—	—
4	AcOH/AcONa	Aquamet (5)	60.5	1.2
5	PBS	Aquamet (10)	0.0	—
6	PBS	Aquamet (10)	0.0	—
7	AcOH/AcONa	Aquamet (10)	104.8	1.0
8	AcOH/AcONa	Aquamet (10)	71.8	—
9	PBS	HG-II (10)	1.4	—
10	AcOH/AcONa	HG-II (10)	16.7	—

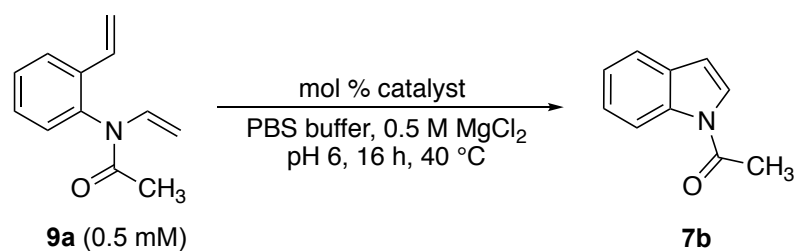
^a PBS: phosphate buffer saline, pH 7.4. HSA: human serum albumin, 0.5 mM. Substrate concentration: 1.0 mM. Reactions were carried out at 40 °C for 24 hours.

Speculating that RCM with an internal olefin is more challenging in aqueous solution [25], we designed the *N*-vinylanilide derivatives **9a** and **9b**, Scheme 3. These substrates are conveniently synthesized in three steps starting from the neat distillation of the commercially available 2-(2-aminophenyl)ethan-1-ol **8** [26], to afford 2-vinylaniline **3**. The amino group is functionalized with the acetyl or succinyl appendages to yield respectively **4b** and **4c**. The last step is a Cu-catalyzed *N*-vinylation of the secondary amine to afford substrates **9a** and **9b**, Scheme 3.



Scheme 3. Synthesis of the *N*-acetylindole precursor **9a** and *N*-succinylindole methyl ester precursor **9b**.

Table 3 summarizes the results of the aqueous RCM with the substrate **9a**. The RCM activity assay reveals modest to good yields of indole **7b** with four metathesis catalysts, among which **G-II** (10 mol % catalyst loading) gave the highest yield of *N*-acetylindole (72 %, Table 2, entry 6). The water-soluble catalyst **Aquamet** (5 mol % catalyst loading) afforded 66 % of indole **7b** (Table 2, entry 7).

Table 2. Summary of the screening results for the aqueous RCM of substrate **9a**.^a

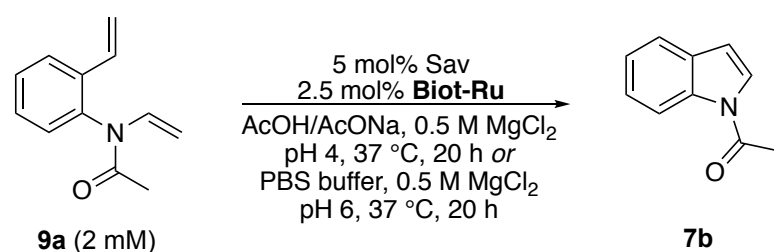
Entry	Catalyst (mol %)	Yield (%) ^b	TON
1	Biot-Ru (5)	35 ± 10	7 ± 2
2	Biot-Ru (10)	41 ± 15	4 ± 1.5
3	HG-I (5)	56 ± 5	11 ± 1
4	HG-I (10)	63 ± 0	6 ± 0
5	G-II (5)	67 ± 15	13 ± 3
6	G-II (10)	72 ± 1	7 ± 0
7	Aquamet (5)	66 ± 6	13 ± 1
8	Aquamet (10)	66 ± 5	6 ± 1

^a Substrate concentration: 0.5 mM; ^b results from two independent catalytic runs.

Although the *N*-acetylindole precursor **9a** afforded modest conversion with the biotinylated catalyst **Biot-Ru** (Table 2, entries 1, 2), we screened the RCM activity with artificial metalloenzymes (ArMs) based on the biotin-streptavidin technology. The presence of a biotin anchor on an Hoveyda Grubbs-derived catalyst ensures that, in the presence of equimolar amounts streptavidin (Sav) isoforms, the metathesis catalyst is quantitatively embedded within the Sav host. Site-directed mutagenesis at close-lying residues (e.g. S112 and K121, ref. 13) allows to genetically improve the RCM activity [27-30]. A screening of > 50 Sav mutants at pH 4, 5 and 6 was carried out (see Supporting info, Figure S7-9). Selected results of the RCM activity of substrate **9a** are collected in Table 4. This screening reveals the following trends: i)

The free catalyst **Biot-Ru** performs best at pH = 6 (Table 3, entry 6-8). Up to 47 % yield is achieved with 2.5 mol % **Biot-Ru** (e.g. 19 TON). ii) In contrast, the ArMs perform best at pH 4, affording up to 16 % yield of *N*-acetylindole (Table 3, entry 2-4, 6 TON). iii) Mutations at position K121 have a positive effect on the RCM activity. We hypothesize that removal of the basic lysine residue contributes to lower the local pH around the Ru-center [31]. Accordingly, the ArM **Biot-Ru**·Sav^{K121L} outperforms **Biot-Ru**·Sav^{WT} at pH 6, yielding 18 % (7 TON) of the ring-closed product **7b**.

Table 3. Summary of RCM activity of ArMs **Biot-Ru**·Sav^{K121L} using diolefin **9a**.



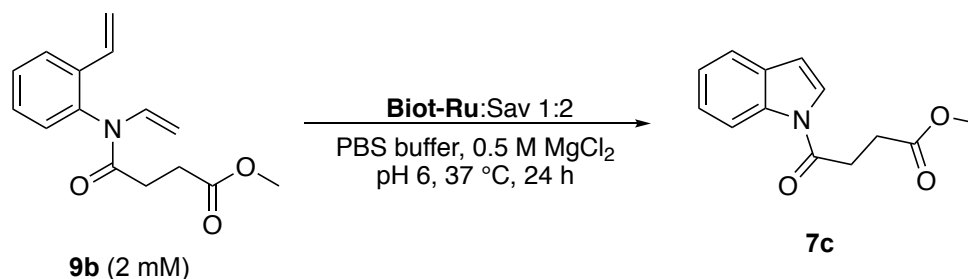
Entry ^a	Sav	pH	Product (μM) ^b	Yield (%)	TON
1	—	4.0	155 ± 7	8 ± 0	3 ± 0
2	WT	4.0	220 ± 2	11 ± 0	4 ± 0
3	K121L	4.0	323 ± 24	16 ± 1	6 ± 1
4	S112N	4.0	282 ± 7	14 ± 0	6 ± 0
5	—	6.0	940 ± 4	47 ± 0	19 ± 0
6	WT	6.0	167 ± 1	8 ± 0	3 ± 0
7	K121L	6.0	359 ± 26	18 ± 3	7 ± 1
8	S112N	6.0	118 ± 10	6 ± 0	2 ± 0

^a [9a] = 2.0 mM in PBS buffer and acetone (5 % V/V); ^b results from two independent catalytic runs.

The catalytic activity of ArMs with the substrate **9b** is collected in Table 4. Again, removal of the lysine in position 121 had a positive effect on RCM, giving a threefold increase in activity: from 7 % yield of indole **7c** with **Biot-Ru**·Sav^{WT} (2 TON) to 18 % with **Biot-Ru**·Sav^{K121A} and 20 % with both **Biot-Ru**·Sav^{K121F} (6 TON) and **Biot-**

Ru·Sav^{K121L} (Table 4, entries 2-5). The free cofactor performed best yielding 26 % of *N*-succinylindole methylester (8 TON, Table 4, entry 1).

Table 4. Summary of RCM activity of ArMs **Biot-Ru·Sav^{K121L}** using diolefin **9b**.

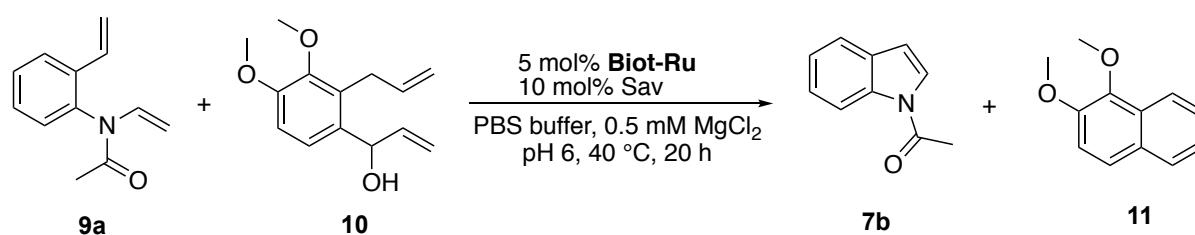


Entry	Sav	mol % Biot-Ru	Yield (%)	TON
1	—	3.33	26.0	8
2	WT	3.33	7.0	2
3	K121A	3.33	18.0	5
4	K121F	3.33	20.0	6
5	K121L	3.33	20.0	6

Next, we tested the influence of the presence of an additional diolefinic substrate **10** on the RCM activity. For this purpose, the RCM of the *N*-acetylindole precursor **9a** was carried out in the presence or absence of diolefin **10**, while keeping the overall catalyst concentration at 5 mol %. Catalysis was performed either with the free catalyst **Biot-Ru** or with the ArMs **Biot-Ru·Sav^{WT}** and **Biot-Ru·Sav^{K121L}**. Unexpectedly, the yield of *N*-acetylindole **7b** increased when the catalysis was performed in the presence of both substrates **9a** and **10**. The free cofactor gave 71 % yield of *N*-acetylindole **7b** compared to 38 % yield when using substrate **9a** alone (Table 5, entry 1-3). This difference in yield was not noticed in previous RCM experiments with substrate **9a** alone, not even when the catalyst concentration was increased from 5 % to 10 %

(Table 2, entry 1-2). We hypothesize that the presence of a second and more reactive substrate delays the catalyst decomposition, probably due to the suppression of the Ru methylidene species which is known to be highly unstable [32-35]. This effect occurs also with the ArMs, producing over sixfold increase in yield with **Biot-Ru**·Sav^{WT} (Table 5, entry 4-6). The ArM **Biot-Ru**·Sav^{K121L} produced an increase in yield from 11 % to 42 % (Table 5, entry 7-9). The influence of the pH on this competition assay was evaluated with selected Sav mutants at pH 4,5 and 6 (see supporting info, Figure S10). We finally evaluated the aqueous RCM of the substrate **10** alone using the same conditions. Strikingly, in the presence of **9a**, the yield of **11** drops from 20 % to 11 % (Table 5, entry 5-6) with **Biot-Ru**·Sav^{WT} and from 65 % to 39 % with **Biot-Ru**·Sav^{K121L} (Table 5, entry 8-9). In contrast, the free catalyst **Biot-Ru** improves the yield of **11** from 35 % to 57 %, suggesting that the ArMs preferentially lead to higher yields of *N*-acetylindole **7b** from an equimolar mixture of substrates **9a** and **10**.

Table 5. Summary of the results of competition RCM of substrates **9a** and **10** in the presence of ArMs.



Entry	Sav	Substrate (mM)	Yield (%) 7b	Yield (%) 11	TON (7b)	TON (11)
1		9a (0.4)	38 ± 0	—	7.6 ± 0.0	—
2	—	10 (0.4)	—	35 ± 0	—	7.0 ± 0.0
3		9a (0.2) + 10 (0.2)	71 ± 0	57 ± 0	7.1 ± 0.0	5.7 ± 0.0
4	WT	9a (0.4)	2 ± 0	—	< 1	—
5		10 (0.4)	—	20 ± 0	—	4.0 ± 0.0

6		9a (0.2) + 10 (0.2)	13 ± 3	11 ± 3	1.3 ± 0.3	1.1 ± 0.3
7		9a (0.4)	10 ± 1	—	2.0 ± 0.2	—
8	K121L	10 (0.4)	—	65 ± 0	—	13.0 ± 0.0
9		9a (0.2) + 10 (0.2)	42 ± 0	39 ± 0	4.2 ± 0.0	4.0 ± 0.0

Conclusion

In conclusion, we report herein the synthesis of *N*-substituted indoles via aqueous RCM of *N*-vinylanilide derivatives. RCM with commercially available catalysts yielded up to 72 % of *N*-acetylindole in PBS buffer at pH = 6 and 2-5 % (V/V) organic cosolvents. In the presence of ArMs based on the biotin-streptavidin technology, up to 42 % of *N*-acetylindole was achieved. Interestingly, the yield of *N*-acetylindole could be improved by addition of a metathesis substrate in the reaction medium, revealing that the addition of a second metathesis substrate has a beneficial effect on the RCM of the indole precursor.

Acknowledgment T.R.W. thanks the University of Basel, the NCCR Molecular Systems engineering, the SNF (grant 200020_182046) and the ERC (the DrEAM, advanced grant 694424) for generous funding.

Conflict of interest The authors declare that they have no conflict of interest.

References

1. Joule JA, Mills K (2000) Heterocyclic Chemistry. Blackwell Science, Oxford, UK.
2. Sundberg RJ (1996) Indoles. Academic Press, San Diego.
3. Kaushik NK, Kaushik N, Attri P, Kumarn N, Kim CH, Verma AK, Choi EH (2013) Molecules 18: 6620.

4. Taber DF, Tirunahari PK (2011) *Tetrahedron* 67:7195.
5. Lee JH; Lee J (2010) *FEMS Microbiol Rev* 34:426.
6. Nelson DL, Cox MM (2005) *Principles of Biochemistry*, 4th ed. W. H. Freeman, New York.
7. Crawford IP (1975) *Bacteriol Rev* 39:87.
8. Arisawa M, Terada Y, Takahashi K, Nakagawa M, Nishida A (2006) *J Org Chem* 71:4255.
9. Yoshida K, Hayashi K, Yanagisawa A (2011) *Org Lett* 18:4762.
10. van Otterlo WAL, de Koning CB (2009) *Chem Rev* 109:3743.
11. Donohoe TJ, Orr AJ, Bingham M (2006) *Angew Chem Int Ed* 45:2664.
12. Kajetanowicz A, Chatterjee A, Reuter R, Ward TR (2013) *Catal Lett* 144:373.
13. Jeschek M, Reuter R, Heinisch T, Klehr J, Panke S, Ward TR (2015) *Nature* 537:66.
14. Sabatino V, Rebelein JG, Ward TR (2019) *J Am Chem Soc* 141:17048.
15. Zhao J, Kajetanowicz A, Ward TR (2015) *Org Biomol Chem* 13:5652.
16. Burtscher D, Grela K (2009) *Angew Chem Int Ed* 48:442.
17. Tomasek J, Schatz J (2013) *Green Chem* 15:2317.
18. Piola L, Nahra F, Nolan SP (2015) *Beilstein J Org Chem* 11:2038.
19. Skowerski K, Bialecki J, Tracz A, Olszewski TK (2014) *Green Chem* 16:1125.
20. Novak BM, Grubbs RH (1988) *J Am Chem Soc* 110:7542.
21. Arisawa M, Theeraladanon C, Nishida A, Nakagawa M (2001) *Tetrahedron Lett* 45:8029.
22. Alcaide B, Almendros P, Luna A (2009) *Chem Rev* 109:3817.
23. Hansen CL, Clausen JW, Ohm RG, Ascic E, Le Qument ST, Tanner D, Nielsen TE (2013) *J Org Chem* 78:12545.

24. Clark JR, Griffiths JR, Diver ST (2013) *J Am Chem Soc* 135:3327.
25. Guidone S, Songis O, Nahra F, Cazin CSJ (2015) *ACS Catal* 5:2697.
26. Dolman SJ, Schrock RR, Hoveyda AH (2003) *Org Lett* 5:4899.
27. Schwizer F, Okamoto Y, Heinisch T, Gu Y, Pellizzoni MM, Lebrun V, Reuter R, Köhler V, Lewis JC, Ward TR (2018) *Chem Rev* 118:142.
28. Davis HJ, Ward TR (2019) *ACS Central Science* 5:1120.
29. Liang, AD, Serrano-Plana J, Peterson RL, Ward TR *Acc Chem Res* 52:585.
30. Sabatino V, Ward TR (2019) *Beilstein J Org Chem* 15:445.
31. Adjiman CS, Clarke AJ, Cooper G, Taylor PC (2008) *Chem Comm* 24:280.
32. Schwab P, Grubbs RH, Ziller JW (1996) *J Am Chem Soc* 118:100.
33. Adlhart C, Chen P (2004) *J Am Chem Soc* 126:3496.
34. Hong SH, Day MW, Grubbs RH (2004) *J Am Chem Soc* 126:7414.
35. Elaridi J, Jackson WR, Robinson AJ (2005) *Tetrahedron: Asymmetry* 16:2025.