

Size-Dependent Joule Heating of Gold Nanoparticles Using Capacitively Coupled Radiofrequency Fields

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ABSTRACT

Capacitively coupled shortwave radiofrequency fields (13.56 MHz) resistively heat low concentrations (~1 ppm) of gold nanoparticles with a thermal power dissipation of ~380 kW/g of gold. Smaller diameter gold nanoparticles (< 50 nm) heat at nearly twice the rate of larger diameter gold nanoparticles (≥ 50 nm), which is attributed to the higher resistivity of smaller gold nanostructures. A Joule heating model has been developed to explain this phenomenon and provides critical insights into the rational design and engineering of nanoscale materials for noninvasive thermal therapy of cancer.

KEYWORDS

Resistivity, radiofrequency, gold, nanoparticles, cancer, thermal

Introduction

Nanoparticle-enhanced thermal destruction of tumors has been demonstrated by using near infrared (NIR) light to heat gold nanoshells or by using strong alternating magnetic fields to heat iron oxide nanoparticles [1–3]. Although these

methods show significant promise in treating cancer, they have disadvantages that limit their practical implementation as therapies. Noninvasive NIR heating of nanoshells is only effective at treating subsurface cancer (~ a few mm deep) due to significant scattering and attenuation of NIR light by biological tissues [4]. Magnetic field-based hyperthermia

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is effective at treating deep tissue cancer, but suffers from limited thermalization by iron oxide nanoparticles. The highest reported thermal power dissipation by iron oxide is a relatively low ~500 W/g of nanomaterial (at a magnetic field amplitude of ~11 kA/m). Therefore, the concentrations of iron oxide required for effective therapy are much higher than can be reasonably achieved *in vivo* [3, 5].

Recently, we have shown that gold nanoparticles heat under capacitively coupled radio frequency fields [6, 7]. RF heating of gold nanoparticles within cancer cells overcomes the major limitations associated with other noninvasive nanoparticle heating methods, since RF energy penetrates well into the body to efficiently heat gold nanoparticles within deep tissue tumors [8]. Previous studies evaluated the therapeutic benefits of capacitive RF heating of gold nanoparticles, but the mechanism of thermalization of RF energy by gold nanoparticles has remained unaddressed and poorly understood.

This paper aims to clarify the physical basis of gold nanoparticle heating by examining the capacitive RF heating properties of gold nanoparticles with respect to their volume fraction and diameter. The analysis contained herein provides critical insights into how the physical properties of gold nanoparticles influence their RF thermal delivery, which will aid in the further development of nanoscale materials for the treatment of cancer.

1. Experimental

1.1 Materials

Aqueous spherical gold nanoparticles with diameters between 5 and 250 nm were purchased from Ted Pella, Inc. (Redding, CA, USA). Gold nanoshells, provided by Nanospectra Biosciences, Inc. (Houston, TX, USA), consist of a 120 nm diameter silica core (Precision Colloids, LLC., Cartersville, GA, USA) with a 10–15 nm thick outer shell of gold (~10–15 nm thick).

1.2 Materials characterization

The size and morphology of all nanoparticles were confirmed using transmission electron microscopy

(TEM). TEM was performed using a JEM 1010 transmission electron microscope (JEOL, Inc., Peabody, MA, USA) with an accelerating voltage of 80 kV and digital images were acquired using an AMT Imaging System (Advanced Microscopy Techniques Corp., Danvers, MA, USA). Drops of nanoparticle suspensions were placed onto a poly-L-lysine treated formvar coated copper grid for 1 h. Grids were blotted dry with filter paper and air dried before TEM observation (see Fig. S-1 in the Electronic Supplementray Material (ESM)). All gold concentrations are reported in ppm by mass and were determined via inductively coupled plasma-mass spectrometry (ICP-MS) analysis.

1.3 Capacitively coupled RF heating of nanoparticle solutions

Figure 1 shows the 13.56 MHz capacitive RF heating system (Therm Med, LLC, Inc., Erie, PA, USA). A cylindrical quartz cuvette (1 cm diameter, 1.5 cm height) containing 1.5 mL of aqueous colloid was positioned on a thin (~0.5 cm) horizontal Teflon® platform (not shown) located mid-plane between two insulated copper electrodes at a spacing of 7.5

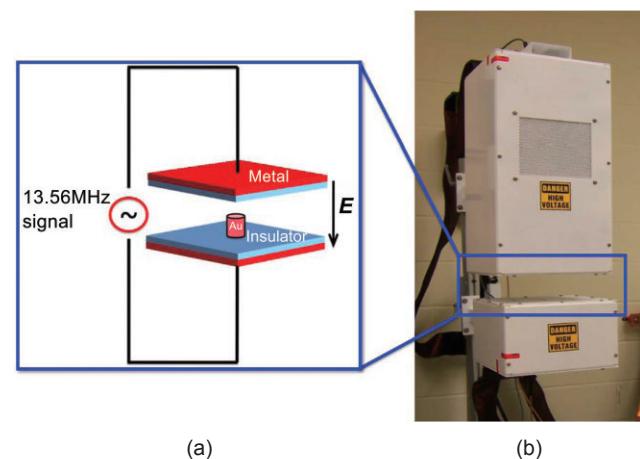


Figure 1 (a) Diagram of the capacitive RF heating system. A 13.56 MHz signal is applied across two metal electrodes (red) that are coated with an insulating Teflon® layer (light blue) which produces a high-voltage RF field ($|E| = 15 \text{ kV/m}$ at 600 W of RF generator power) over a variable air gap. Gold nanoparticle suspensions contained within a cylindrical cuvette are placed on a Teflon platform (not shown) between the two electrodes. (b) Prototype capacitive RF heating device. The metal chassis (upper white box) contains high-voltage matching circuits to efficiently produce RF fields between the electrodes. The location of the electrodes within the metal chassis is highlighted by the blue rectangle



cm. All suspensions were exposed to 600 W of RF generator power resulting in an RF field of 15 kV/m. Reflected RF power was monitored and minimized by impedance matching during RF exposure [7–9].

1.4 Thermal imaging

The temperature of the cuvette was continuously monitored using an infrared camera and thermal emissions were recorded (Amber Engineering Inc., Goleta, California, USA). All colloidal suspensions were exposed to RF fields until they reached a final temperature of 70 °C or for a maximum duration of 2 min, whichever was achieved first.

2. Results and discussion

Figure 2 shows a time-lapsed sequence of captured infrared images of an aqueous suspension of 10 nm gold nanoparticles (36 ppm) capacitively coupled to an RF field. Upon RF absorption, the temperature of the gold nanoparticle suspension rapidly increased from 25 to 70 °C in less than 20 s and the solution boiled within 30 s of RF exposure. The false color thermal map shows uniform heating of the suspension with a sharp temperature gradient at the base of the cuvette. The thin non-conducting platform supporting the cuvette remained at room

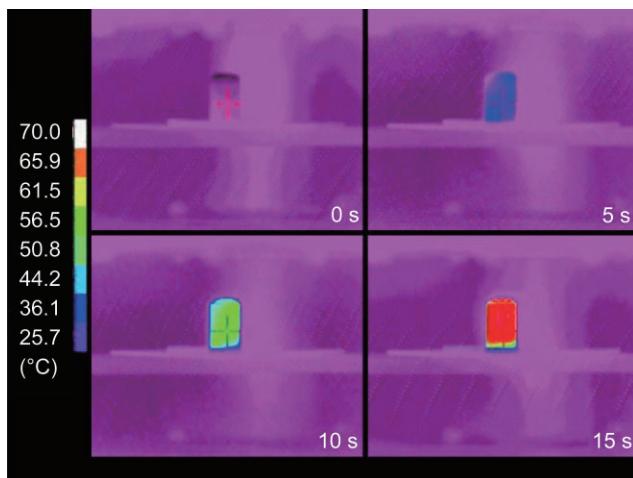


Figure 2 Time-lapsed infrared images, taken 5 s apart, of a cylindrical quartz cuvette filled with 1.5 mL of 10 nm gold nanoparticles in water (~36 ppm) exposed to a 15 kV/m RF field. RF energy is absorbed by gold, which boils water in less than 30 s. The temperature of each sample was recorded at the center of the crosshair

temperature throughout RF exposure.

Figure 3 plots the change in temperature versus time for selected concentrations of 10 nm gold nanoparticles capacitively coupled to RF fields. Deionized water alone and physiological saline (not shown) exhibited a 5 °C rise from room temperature over a 2-min period, which equates to 0.2 W of thermal power dissipation. But the addition of just ~1 ppm of gold by mass, resulted in a five-fold increase in thermal power dissipation, giving a temperature rise (ΔT) of 35 °C in 2 min.

The thermal power dissipated per gram of material is the ratio of the thermal energy delivered into solution to the total mass of material in suspension. Therefore, the thermal power dissipation of 1 ppm gold nanoparticles under our RF field conditions was a remarkable ~380 kW/g. Additionally, for the maximum concentration (36 ppm) of 10 nm gold nanoparticles, the heating rate was 3.4 °C/s, which corresponds to a 70-fold increase in the thermal power dissipation by gold (14 W) as compared to water alone.

In order to investigate the dependence of RF heating on nanoparticle diameter, we capacitively applied RF fields to various sizes of gold nanoparticles (5–250 nm). Figure 4 displays the heating rates of several different diameters of gold

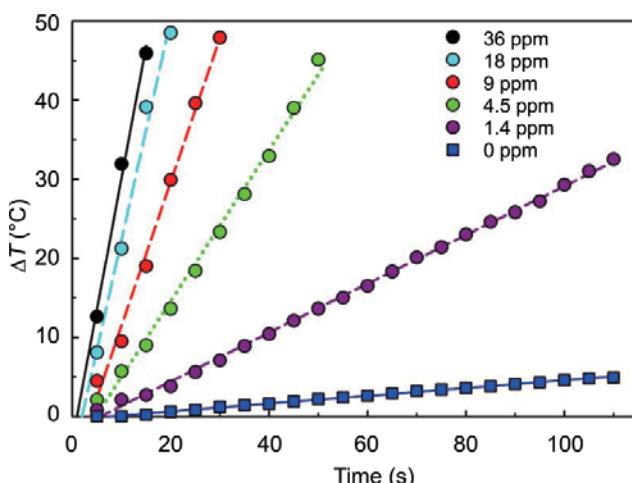


Figure 3 Change in temperature with time of 10 nm diameter gold nanoparticle suspensions (circles) and water alone (squares) exposed to RF fields. Lines shown are linear fits by least squares regression of temperatures recorded from the infrared images shown in Fig. 2. RF energy absorbed by 1.4 ppm of gold nanoparticles exhibited a final ΔT ~35 °C in 2 min, whereas water alone exhibited a ΔT ~5 °C

nanoparticles versus their gold volume fraction. This plot reveals three important properties of gold nanoparticles under capacitively coupled RF fields. First, the rate of RF heating increases nonlinearly and approaches a horizontal asymptote as the nanoparticle concentration increases. This finding is similar to what we observed for the concentration-dependent RF heating of carbon nanotubes [8]. Second, at equivalent volume fractions, small diameter gold nanoparticles (<50 nm) exhibit heating rates that are approximately twice as high as those of gold nanoparticles with diameters of 50 nm and above. Third, solid gold nanoparticles with diameters between 50–250 nm heat at nearly equivalent rates with no distinguishable diameter dependence.

In order to independently examine nanoparticle diameter and gold content effects under RF fields, we tested gold nanoshells. Since nanoshells are composed of an inner dielectric core (SiO_2) and a thin outer shell of gold, the total volume fraction of gold in solution can be varied while holding the overall diameter of the nanoparticle constant. Surprisingly, at equivalent gold volume fractions, 150 nm diameter gold nanoshells (ca. 10–15 nm gold shell thickness) exhibit RF heating rates comparable to solid gold nanoparticles ≤ 20 nm in diameter. Note, that in a

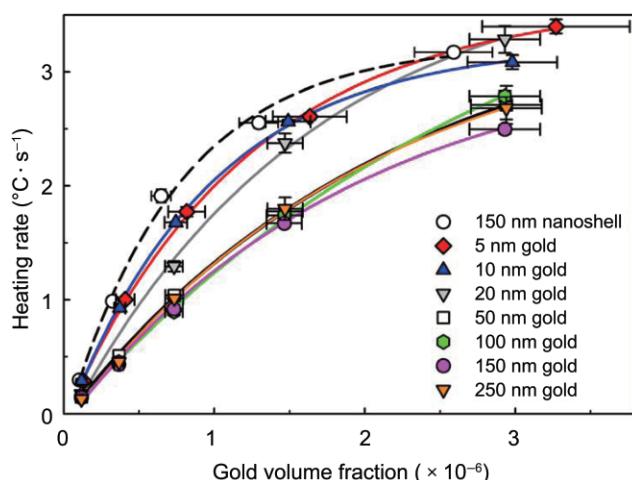


Figure 4 Size-dependent RF heating rates of gold nanoparticles versus gold volume fraction. Heating rates were obtained from the slopes of the linear fits shown in Fig. 3. There is a clear size-dependent clustering of RF heating rates with 150 nm diameter gold nanoshells (with a shell thickness of 10–15 nm) exhibiting heating rates comparable to smaller solid gold nanoparticles, due to the increased resistivity of smaller gold nanostructures. Note, some data points are obscured by overlapping symbols

control experiment 120 nm diameter uncoated silica nanoparticles exhibited an RF heating rate equivalent to that of water.

The increased RF induced heating rates of gold nanoparticles <50 nm in diameter and gold nanoshells 10–15 nm thick can be explained by the higher resistivity of small metal nanostructures compared to bulk metals [10]. Recent studies have shown that the resistivity of silver nanowires 15 nm in diameter is approximately twice that of bulk silver. Particles and shells of gold with dimensions on the order of 10 nm are expected to exhibit a similar increase in resistivity due to increased electron-surface scattering since the size of the metal is significantly smaller than the mean free path of electrons in gold, which is on the order of 50 nm [10–12].

If the heat released is due to resistive (Ω) dissipation by gold, then the measured heat should scale with Joule's law, $P = I^2R$, where P is the power dissipated as heat, I is the current, and R is the resistance. For a given volume fraction of gold nanoparticles, an increase in the resistivity should lead to a commensurate increase in the amount of heat generated by ohmic dissipation. As such, the observed doubling of the heating rate for small gold nanoparticles and nanoshells is consistent with the hypothesis that gold nanoparticles are heating resistively under capacitively coupled RF fields.

Radiofrequencies are far below the electronic resonance frequencies for gold nanoparticles, which are typically in the infrared-visible spectrum; because of this, capacitive coupling to the RF field dominates the movement of charge through nanoparticles. Therefore, the amount of Joule heat generated by each gold nanoparticle under RF oscillations can be directly estimated by approximating gold nanoparticles as conductors of constant cross section:

$$I^2R = \sigma(\omega) \cdot [a \cdot d \cdot E^2] \quad (1)$$

where $\sigma(\omega)$ is the Drude model conductivity of gold ($\sim 4.3 \times 10^7 \Omega^{-1} \cdot \text{m}^{-1}$ at an angular frequency of 85.16 MHz), E is the magnitude of the electric field, a is the cross-sectional area of the gold nanoparticle, and d is its diameter [13–15]. Assuming efficient heat transfer from n nanoparticles to the surrounding



water, the volumetric RF induced heating rate (dT/dt) of gold nanoparticles is shown in Eq. (2):

$$\frac{dT}{dt} = \left[\frac{\sigma(\omega) \cdot a \cdot d \cdot |\mathbf{E}|^2}{v \cdot c_w} \right] \cdot n \quad (2)$$

where v is the volume of the suspension and c_w is the heat capacity of water ($4.18 \text{ J} \cdot \text{mL}^{-1} \cdot ^\circ\text{C}^{-1}$) [16].

A 1.5 mL suspension of 100 nm diameter gold nanoparticles ($n = 9 \times 10^9$) exposed to an effective RF field strength in water of $|\mathbf{E}| \sim 200 \text{ V/m}$ exhibited an experimental heating rate of $\sim 2.8 \text{ }^\circ\text{C/s}$, whilst Eq. (2) yields a comparable RF induced theoretical heating rate of $1.9 \text{ }^\circ\text{C/s}$. The agreement between the theoretical and experimental RF induced heating rates, as well as the increased heating observed for smaller gold nanostructures, provides further evidence that Joule heating of gold nanoparticles by capacitively coupled RF fields is involved.

3. Conclusions

In summary, gold nanoparticles at low concentrations deliver a remarkable $\sim 380 \text{ kW/g}$ of thermal power dissipation under capacitively coupled RF fields (13.56 MHz). Gold nanoparticles resistively heat under RF fields, with small gold nanoparticles ($\leq 20 \text{ nm}$) and thin gold nanoshells (shell thickness of $\sim 10\text{--}15 \text{ nm}$) exhibiting heating rates approximately twice that of gold nanoparticles of 50 nm or larger, due to the higher resistivity of smaller nanostructures. By quantifying the amount of heat produced by gold nanoparticles as a function of size and concentration, this study establishes the critical design metrics necessary for formulating RF responsive nanoscale materials to enhance the noninvasive thermal destruction of cancer.

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