Chemical Elements in the Environment

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Chemical Elements in the Environment

Factsheets for the Geochemist and Environmental Scientist



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PREFACE

CHEMICAL ELEMENTS IN THE ENVIRONMENT

The idea of compiling geochemical and environmental data from existing literature probably has occurred to many of our readers. We can hardly claim any originality here! In our case, the constant frustration of searching, sometimes vainly, sometimes repeatedly, for data with which we could compare the results from a major multimedia, multielement ecogeochemistry project on the Kola peninsula, was the impetus for commencing this task. Thus, we set out to collect real data from regional, multielement studies available in existing literature. It was quite an eye-opener to realise how few such reliable studies exist, how limited in scope most are (either spatially or in terms of analytical breadth), and how haphazardly organised some previous attempts at data collation appear.

We have thus tried to collect reliable information on the 92 elements reported to occur in nature. The resulting factsheets, which form the core of this book, consist of four pages of data for each element. For each of these elements, we have started the factsheet by giving essential physico-chemical information, followed by the abundances in typical rocks and crustal materials, and an indication of which important minerals are likely to contain the element in question. Subsequently, we have listed the concentrations of each element in various geochemical or environmental sample media, ranging from rocks, to soils, plants, waters, and even to human body fluids. As explained below, we have selected carefully the sources from which we extracted the data. Finally, we close each factsheet with a discussion of the biological significance, uses, environmental pathways, mobility, action levels, production and prices of each element. We have done our utmost to reproduce these values faithfully, and we hope that our readers will inform us of any mistakes or omissions, so that possible later editions might be improved.

We would also like to mention that while preparing the manuscript for this book we met several scientists who were reluctant to contribute their data. Their unwillingness, rational as it may first appear, was generally fuelled by the suspicion that they would not be credited for their work, whereas we would. It is important to stress here that the purpose of this book is to prepare a collection of easily comparable data sets. We urge readers wanting to use these data in their own work to go back to, and cite, the original sources, or at least the contributor's name (e.g., "Manning as cited in Reimann and Caritat"). In any case, it is always strongly advisable to acquire and consult the original reference to find out more about sampling strategy, exact area covered, number of samples collected and analysed for each element, analytical details, and quality assurance and quality control procedures. The tendency to "hide" original data presents a real danger to the further development of science. In recent years, more and more papers have been published without data tables or even without data statistics. In other cases, only data for the three or four "most important" elements are given, even though very many more have obviously been analysed for. A good and easily

accessible collection of existing data is fundamental to advance the understanding of the behaviour of chemical elements in the environment, to identify gaps in our knowledge and, last but not least, to generate new research ideas.

Chemical Elements in the Environment is intended as a workbook for the professional geochemist and environmental scientist, to which, we hope, they will often refer when interpreting their own data sets. While working on the manuscript of this book, we also bore in mind students in natural sciences at the university or college level who would be curious about the distribution of chemical elements on our planet - e.g. where are the main reservoirs of beryllium and rubidium? What do we know about tin in the environment? By how many orders of magnitude can the concentration of uranium in groundwater vary? The answers to these and many other questions can be found within these pages!

In addition to the libraries of our fellow scientists, we hope that this book will find a place on the shelf, or preferably in the hands, of legislators in environmental protection or management around the world. We believe that they, perhaps more than anyone else, need a book like this one to keep in touch with real data and hard facts about the environment.

We hope all readers will find this work useful. However, we cannot accept responsibility for any consequences resulting from the use of information in this book.

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Clemens Reimann and Patrice de Caritat Trondheim, February 1998

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Ac Ag Al Ar As At Au	Actinium	30 34 38 42 46			
B Ba Be Bi Br	Boron . Barium . Beryllium . Bismuth . Bromine .	56 60 64			
C Ca Cd Ce Cl Co Cr Cs Cu	Carbon Calcium Cadmium Cerium Chlorine Cobalt Chromium Cesium	76 84 88 92 96 100			
Dy	Dysprosium				
Er Eu	Erbium				
F Fe Fr	Fluorine	24			
Ga Gd Ge	Gallium Gadolinium Germanium	34			
H He	Hydrogen				

Hf Hg Ho	Hafnium Mercury Holmium	154
l In Ir	lodine	166
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La Li Lu	Lanthanum Lithium Lutetium	186
Mg Mn Mo	Magnesium	198
N Na Nb Nd Ne Ni Np	Nitrogen 2 Sodium 2 Niobium 2 Neodymium 2 Neon 2 Nickel 2 Neptunium 2	210 214 218 222 226
O Os	Oxygen	
P Pa Pb Pd Po Pr Pt Pu	Phosphorus 2 Protactinium 2 Lead 2 Palladium 2 Polonium 2 Praseodymium 2 Platinum 2 Plutonium 2	244 248 252 256 260 264
Ra Rb Re Rh Rn Ru	Radium 2 Rubidium 2 Rhenium 2 Rhodium 2 Radon 2 Ruthenium 2	274 278 282 286
S Sb Sc Se Si Sm Sn Sr	Sulphur 2 Antimony 2 Scandium 3 Selenium 3 Silicon 3 Samarium 3 Tin 3 Strontium 3	298 302 306 310 314 318
Ta Tb Te Th Ti	Tantalum 3 Terbium 3 Tellurium 3 Thorium 3 Titanium 3	326 330 334 338

TI Tm	Thallium	346 350		
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